| 1  | Scale-up and economic analysis of biodiesel production from municipal primary sewage            |
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| 2  | sludge  |
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| 11 |   |
| 12 | Abstract  |
| 13 | Municipal wastewater sludge is a promising lipid feedstock for biodiesel production, but the    |
| 14 | need to eliminate the high water content before lipid extraction is the main limitation for     |
| 15 | scaling up. This study evaluates the economic feasibility of biodiesel production directly from |
| 16 | liquid primary sludge based on experimental data at laboratory scale. Computational tools       |
| 17 | were used for the modelling of the process scale-up and the different configurations of lipid   |
| 18 | extraction to optimise this step, as it is the most expensive. The operational variables with a |
| 19 | major influence in the cost were the extraction time and the amount of solvent. The optimised   |
| 20 | extraction process had a break-even price of biodiesel of 1232 \$/t, being economically         |
| 21 | competitive with the current cost of fossil diesel. The proposed biodiesel production process   |
| 22 | from waste sludge eliminates the expensive step of sludge drying, lowering the biodiesel        |
| 23 | price.  |
| 24 |   |

#### 25 Keywords

Sewage sludge, Lipids, Biodiesel, Process modelling, Economic evaluation

#### 28 **1. Introduction**

29 Biodiesel is one of the most promising renewable fuels that is biodegradable, less toxic, 30 may generate similar amount of energy to fossil diesel and can be directly used with current 31 engine and refuelling technology/infrastructure without major modification (Siddiquee and 32 Rohani, 2011; Kwon et al., 2012; Atabani et al., 2012). Biodiesel, *i.e.*, fatty acids methyl ester 33 (FAME), is mainly produced from edible vegetable oils, however the high cost of vegetable 34 oils which constitutes between 70-85% of the overall biodiesel production cost, strongly 35 influences the final price of this biofuel, limiting its expansion (Mondala et. al., 2009; 36 Siddiquee and Rohani, 2011; Kwon et al., 2012; Atabani et al., 2012). Furthermore, the 37 cultivation of edible oilseeds for biofuels raises the concerns of food shortage, which 38 competes with fuel production (Atabani et al., 2012; Khan et al., 2014). 39 The possibility of using municipal sewage sludge as non-edible lipid feedstock is gaining 40 more attention due to the large amounts of sludge generated in the developed countries, and 41 high amount of lipids contained within these wastes, up to 30 wt% (Dufreche et al., 2007; 42 Olkiewicz et al., 2014, 2015a; Tyagi and Lo, 2013; Yi et al., 2016). The amount of lipids 43 strongly depends on the sludge type. The lipid yield in secondary sludge was found in the 44 range of 2-12 wt% (Huynh et al., 2010; Siddiquee and Rohani, 2011; Tyagi and Lo, 2013; 45 Olkiewicz et al., 2012, 2015a), while in primary sludge usually ranges between 15-30 wt% 46 (Willson et al., 2010; Siddiquee and Rohani, 2011; Pastore et al., 2013; Olkiewicz et al., 47 2015a, 2015b; Yi et al., 2016).

48 On the other hand, the sludge formed during treatment of wastewater needs specific
49 management before disposal and represents a major cost in wastewater treatment plant

50 (WWTP) operation (Dufreche et al., 2007; Pastore et al., 2013). Therefore, the sewage sludge 51 can be envisaged as a low-cost, readily available in abundance and non-edible feedstock, 52 which can make biodiesel production profitable. Recent studies have indicated that the lipid 53 contained in sewage sludge could be a potential feedstock for biodiesel production (Dufreche 54 et al., 2007; Mondala et al., 2009; Huynh et al., 2010; Willson et al., 2010; Siddiquee and 55 Rohani, 2011; Kwon et al., 2012; Pastore et al., 2013; Olkiewicz et al., 2012, 2014, 2015a). Nevertheless, the cost of energy necessary to eliminate the high water content (95-98 wt %), 56 57 before lipid extraction, is the main limitation to scale-up, as dewatering and drying constitutes 58 more than 50% of the total biodiesel production cost (Dufreche et al., 2007; Mondala et al., 59 2009). On the other hand, previous research demonstrated the feasibility of lipid extraction 60 from liquid sludge (~96% of water) by direct liquid–liquid extraction using hexane as a 61 solvent (Olkiewicz et al., 2014). Since the production of biodiesel from liquid sewage sludge 62 is feasible, the expensive sludge drying step can be eliminated, and therefore the overall 63 biodiesel production cost can be reduced. However, in order to confirm the stated hypothesis, 64 the economic feasibility of the wet process (direct use of liquid sludge) and its comparison 65 with dry process (use of dry sludge) has to be done. 66 Economic analysis of the production of biodiesel from dry sewage sludge has already been 67 reported. Dufreche et al. (2007) estimated the cost of biodiesel production from dry sludge by 68 direct *in situ* transesterification, without the extraction step, to be 933 \$/t. However, in this 69 research short-cut economic methods were used without giving details about the cost of 70 methods used. A more detailed breakdown of estimated costs also for in situ 71 transesterification was calculated by Mondala et al. (2009), based on data published by others, 72 e.g., the cost of sludge drying was taken from Dufreche et al. (2007). They obtained a break-73 even price of biodiesel of 970 \$/t. Pokoo-Aikins et al. (2010) presented a full economic 74 feasibility study, based on process design and simulation, to choose the best option to produce

75 biodiesel from sludge, using two-step process: preliminary lipid extraction, evaluating four 76 solvents (hexane, toluene, methanol and ethanol), and subsequent conversion of the lipids into 77 biodiesel. The results indicated that hexane and toluene were cheaper solvents, giving 868 and 78 838 \$/t of biodiesel, respectively. These excellent results were obtained considering that dry 79 sludge was free of cost, charging the sludge drying to WWTPs. Certainly, if sludge drying 80 were also taken into consideration, the final price of biodiesel would increase significantly. 81 In short, on the one hand, in the aforementioned economic studies, some assumptions were 82 underestimated and in some cases not all process steps were considered for the estimation of 83 the final biodiesel cost. Therefore, to fairly estimate the biodiesel production cost from 84 sewage sludge, all assumptions must be taken with a constructive criticism and include 85 realistic values. On the other hand, the biodiesel production from wastewater sludge has a 86 promising future but it is still in research stage. Therefore, further large scale studies are 87 required to realize the benefits of this new biotechnology.

88 The purpose of this research is to critically review the biodiesel production from sewage 89 sludge using the know-how acquired at bench scale experimental work. Laboratory scale data 90 obtained in our previous study (Olkiewicz et al., 2014), where the feasibility of lipids 91 extraction directly from liquid sludge was demonstrated, is analysed by computational tools in 92 order to carry out the scale-up of this novel biodiesel production process. In particular, the 93 lipid extraction from liquid primary sludge is optimised by using computational tools to 94 model the process performance and the economic evaluation of the process alternatives. 95 Process options are envisaged to estimate a realistic scenario considering the technology 96 currently available. Finally, the optimised biodiesel production process from liquid sludge is 97 compared to the *in situ* and two-step processes using dry sludge (also simulated in this study) 98 in order to decide on the most economically favourable process.

99

#### 100 2. Materials and Methods

101 A production plant with a capacity of around 4000 t/year of biodiesel produced from 102 primary sewage sludge is studied. The capacity of the facility will depend on the sewage 103 sludge availability. In this sense, a nearby urban waste water treatment plant (WWTP) to feed 104 60 m<sup>3</sup>/h of primary sewage sludge was considered. This set-up (Fig. 1) can eliminate the cost 105 of transporting the sludge feedstock into the biodiesel production facility, which therefore was 106 not taken into account in the economic study as well as the cost of raw sludge, which is a 107 waste generated during treatment of wastewater. The proposed process aims to improve the 108 biodiesel production from sewage sludge, *i.e.*, lipid extraction process, by the elimination of 109 the energy intensive step of sludge dewatering and drying and also the elimination of the 110 heating process during extraction. Particularly, the process developed is compared with those 111 described in other works, whose main differences are: on the one hand, the use of sludge 112 previously dehydrated, with the consequent increased costs of the raw material, that in some 113 assessments seem to be understated or dismissed (Pokoo-Aikins et al., 2010; Zhang et al., 114 2013; Dufreche et al. 2007; Mondala et al., 2009); and on the other hand, the use of heating 115 during extraction, which also increases the cost of the process (Pokoo-Aikins et al., 2010; 116 Zhang et al., 2013; Dufreche et al. 2007); and finally, the conversion of all lipids into 117 biodiesel (Zhang et al., 2013), since based on experimental studies approximately 70-85% of 118 lipids can be converted into biodiesel (saponifiable lipids) (Pastore et al., 2013; Olkiewicz et 119 al., 2014). The economic evaluation of the process and its potential alternatives is performed 120 based on the previous results experimentally tested (Olkiewicz et al., 2014).

121

#### 122 **2.1. Approaches and assumptions**

123 The presented study aims to put some light in the potential of the wet route that so far has124 been underrated with respect to dry routes. The hypothesis considers experimental results in

125 laboratory and also the pilot-scale experience with other lipid sources in water solutions of 126 similar features (microalgae, vegetal oils, etc.). In this sense, although the economic analysis 127 to probe the feasibility of the process is based on conceptual design (specifically in 128 laboratory-scale experiments), it is in line and even more complete than previously mentioned 129 studies about better-known dry options that can be found elsewhere in the literature. The 130 simulation software supports the customization of unit processes allowing the implementation 131 of specific complexities that the simulator cannot solve in a realistic way related with the 132 sewage sludge properties. Two main objectives are pursued, on the one hand, obtaining 133 preliminary profitability indicators of sewage sludge-based biodiesel processing through wet 134 pathway; and on the other hand, comparing the wet route with dry alternatives (in situ and 135 two-step processes using dry sludge) under the same basis, that are the same process 136 simulation and economic modelling procedures.

#### 137 **2.1.1. Primary sludge**

138 The calculations of the economic feasibility study were performed with the data of primary 139 sludge collected from the municipal WWTP in Reus (Tarragona, Spain) with a capacity to 140 process near 25,000 m<sup>3</sup> of wastewater per day, which serves 200,000 inhabitants. The WWTP 141 of Reus produces an average of 135 m<sup>3</sup>/day of primary sludge. For the calculations, the flow 142 rate was approximated to  $60 \text{ m}^3/\text{h}$  to assimilate the production of a big town, as for example 143 the WWTP near Barcelona, which serves approximately to 2 million inhabitants. 144 Fig. 1(a) shows a schematic diagram of the WWTP in Reus, illustrating the sludge 145 generation in the WWTP facility. The primary sludge was sampled after partial gravity thickening. The composition of primary sludge used for the analysis in AspenHysys V8® 146 147 simulator was determined as described by Olkiewicz et al. (2015b), and is shown in Table S1.

#### 148 **2.1.2 Lipid extraction from liquid primary sludge**

149 The calculations of the economic feasibility study were performed using directly the 150 experimental values obtained in laboratory, with a bench scale experimental device, by our 151 research group (Olkiewicz et al., 2014). In the experimental procedure, the sequential liquid-152 liquid extraction of lipid was conducted in a batch mixer-settler reactor at ambient 153 temperature, using hexane as a solvent, where nine consecutive extractions stages were 154 carried out. In the simulation of the process scale-up, the lipid extraction takes place in a 155 liquid-liquid extraction series of mixers, where hexane is used as solvent, but in a continuous 156 counter-current system, as depicted in Fig. 1(b). Thus, the main difference between the flow 157 diagram presented in Fig. 1(b) and previously presented by Olkiewicz et al. (2014) is the 158 extraction system.

A maximum of five mixers were modelled in the AspenHysys V8<sup>®</sup> simulator, as more than 159 90% of lipids were extracted after 5 consecutive extraction stages in the experimental study 160 161 (Olkiewicz et al., 2014). According to the experimental study, in order to extract all lipids 162 from liquid sludge, sludge acidification until pH 2, prior to extraction is required (Olkiewicz 163 et al., 2014). In order to minimize the addition of acid, sludge acidification to pH 4 was also 164 studied. The sludge was acidified to pH 2 and pH 4 by addition of concentrated HCl. Due to 165 the primary sludge pH varies between 5.8 and 6.5, according to experimental results, it was 166 assumed that the required concentration of HCl in liquid sludge is approximately 1.5% or 0.8 167 % v/v to attain the pH 2 and pH 4, respectively (data experimentally tested).

After sludge acidification, the lipid composition consists of 20% non-saponifiable and 80% saponifiable lipids (convertible to biodiesel). As shown in Table S1, the saponifiable lipid consists mainly of free fatty acids (FFAs) and traces of triglycerides (TG), suggesting that the main reaction during the conversion of sludge lipids into biodiesel is esterification of FFAs.

| 172 | Based on the laboratory experiments, six different process configurations (Table 1) were                                   |
|-----|--|
| 173 | selected to perform the comparative economic study with the aim to optimise the number of                                  |
| 174 | extraction stages and extraction conditions, to find the most economically favourable process.                             |
| 175 | 2.1.3. Biodiesel production  |
| 176 | Due to the high content of FFAs in the sludge lipids, the acid catalysis   |
| 177 | esterification/transesterification was selected to convert saponifiable lipids into biodiesel.                             |
| 178 | Acid catalyst, <i>i.e.</i> , H <sub>2</sub> SO <sub>4</sub> , is used in the simultaneous esterification of FFAs and       |
| 179 | transesterification of glycerides avoiding soap formation, which takes place in the case of                                |
| 180 | conventional alkali catalyst (e.g., NaOH). The assumptions applied in the  |
| 181 | esterification/transesterification reaction were taken from Zhang et al. (2013), modifying the                             |
| 182 | temperature of the reaction.   |
| 183 | According to Zhang et al. (2013), the sludge lipids conversion is considered as 99% under                                  |
| 184 | the following conditions: 6:1 methanol to lipids molar ratio, $1\%$ (v/v) of H <sub>2</sub> SO <sub>4</sub> as catalyst in |
| 185 | methanol, temperature reaction of 50°C and 4 hours of residence time. However, the   |
| 186 | laboratory experimental test applied to the primary sludge lipids, using these conditions but                              |
| 187 | taking into account the correction for saponifiable lipids (i.e., 6:1 methanol to saponifiable                             |
| 188 | lipids molar ratio), gave only 88% of reaction efficiency. An increase in the temperature to 60                            |
| 189 | °C, showed an increase in the reaction efficiency to approximately 99%. According to the                                   |
| 190 | experimental results, the reaction was simulated at 60 °C.   |
| 191 | As only the saponifiable part of sludge's lipids is convertible into biodiesel, the separation                             |
| 192 | of non-saponifiable lipids was performed after biodiesel production, in the purification step,                             |
| 193 | as commented on in subsection 2.2.2 of this paper.   |
| 194 |  |

## **2.2. Process simulation**

196 The economic characterization approach is based on the steady-state simulation of the plant in AspenHysys V8<sup>®</sup>, in a continuous process. Although most of the chemical 197 components involved in the simulation are defined in the AspenHysys<sup>®</sup> component library. 198 199 some compounds have been defined as hypothetical solids, as potassium sulphate, the 200 inorganic matter or ash content in the sludge and the organic matter different from lipids, for 201 which average molecular weights and mass densities are defined. Moreover, certain 202 compounds are used in representation of similar substances. More precisely, a mixture of M-203 palmitate and M-oleate was selected to represent the biodiesel product, the palmitic acid 204 represents the FFA content, the triolein plays the role of triglycerides and a saturated fatty 205 acid ester was selected to represent the non-saponifiable lipids (*i.e.*, cetyl palmitate). The 206 palmitic acid was selected due to the predominance of this acid in the sludge lipids and thus in 207 the biodiesel produced (Olkiewicz et al., 2014, 2015a). The primary sewage sludge (and its 208 lipids composition) was simulated according to the sludge characterization presented in Table 209 S1. The Peng-Robinson Soave (PRSV) equation of state was the fluid package selected to 210 predict the physicochemical properties of the chemical components involved, including 211 solutions such as 95% concentrated sulfuric acid. The Peng-Robinson (Peng and Robinson, 212 1976) versions have been the most successful in vapor-liquid equilibria calculations of 213 conventional as well as non-conventional mixtures of fluids (Ghosh, 1999). The selection of 214 the PRSV method takes into account that extends the original PR method for moderately non-215 ideal systems and because it performs rigorous separation for aqueous systems containing 216 methanol, glycols plus hydrocarbons in second liquid phases. These features have been 217 probed adequate for the stages of reaction, separation and purification. Since the simulation 218 software was not able to reproduce accurately the performance of the lipids extraction phase, 219 user customized units were used to model this particular stage due to the presence of sewage 220 sludge solution, as it is explained in subsection 2.2.1.

The detailed flow diagram obtained directly with AspenHysys<sup>®</sup> simulation package, and the characterization of the main material streams involved can be seen in Fig. 2. The detailed list of main equipments used in the AspenHysys<sup>®</sup> simulation model is presented in Table 2. The simulation is based on the current available technologies. The process simulation is structured in two main sections: the lipid extraction from the liquid (96% water content) primary sludge and the acid-catalyzed esterification/transesterification process. In the following sections a detailed description of both process sections is explained.

#### 228 **2.2.1. Lipid extraction from liquid primary sludge**

229 The primary sludge is stabilized with acid before the extraction (V-100), as commented on 230 in subsection 2.1.2. After that, the lipid extraction takes place in a liquid-liquid extraction 231 series of mixers (CSTR-100, 101), where hexane is used as solvent in a counter-current 232 system. The equilibrium and operating data were taken from the experiments carried out by 233 Olkiewicz et al. (2014). In order to find the number of stages that optimize the economic 234 results, different alternatives are modelled regarding the working pH for the sludge, residence 235 time during the extraction stage and the ratio between sludge and hexane, depending on the 236 ratio of hexane used (Table 1). The separation units are totally controlled by the model 237 programmed in a custom-fitting spreadsheet, building tailored units instead of using 238 predefined operations from the simulator toolbox. The model imports the variables of mass 239 flows and composition of the inputs (sludge and solvent) and exports to the separation units 240 the values of splits and separation efficiencies. These output variables are calculated through 241 mathematical expressions of the material balances and equilibrium equations of the extraction 242 based on the adjustment of the experimental results in the corresponding conditions (solvent 243 ratio, mixing time and pH). The design of the extraction is based on a short-cut method 244 comparable to the McCabe-Thiele stepwise calculation for distillation columns (Perry et al., 245 1999). A constant flow rate of feed solvent and extraction solvent is assumed, and the solute

concentrations are given as the weight ratios of solute to feed solvent and extraction solvent in the raffinate and the extract phases, respectively. The compositions (raffinate and extract) of each extraction stage (operation line) are calculated with the equilibrium curve equation obtained by the fit of experimental data. The curve fitting equations and their coefficients of determination for each extraction configuration are presented in Table 1.

The raffinate composed by the rest of biomass and more than 96% water can be recycled to the WWTP, to be used as a substrate to produce biogas by anaerobic digestion (the process widely implemented in municipal WWTPs, presented in Fig. 1(a)), avoiding the generation of a new waste sludge. It has been already demonstrated that the residual lipid extracted sludge can be easily anaerobically digested, producing biogas which maintains a similar composition, *i.e.*, methane content, to that coming from raw excess sludge (Olkiewicz et al., 2014).

On the other hand, the extract is led to an equilibrium-flash separator V-101 where over
99% of hexane is recovered and recycled to the extraction.

#### 259 **2.2.2. Biodiesel production**

260 Acid catalyzed reaction system is proposed for the production of FAME using methanol as 261 reactant, where two reactions take place (CRV-100): the acid esterification of the FFA to give 262 FAME and water, and the acid transesterification of the triglycerides to obtain FAME and 263 glycerol. Based on experimental results, near 99% is achieved for the sludge's lipids 264 esterification/transesterification under the conditions described in subsection 2.1.3. 265 The products stream is forwarded to a decanter (V-104) where the contact with washing 266 water forces the separation of two-phases. The light phase is conducted to a flash separation 267 (V-107) to reduce further the amount of hexane and traces of methanol that accompanied the

268 obtained biodiesel. The heavy phase includes water, methanol, a low quantity of glycerol as

by-product of the transesterification reaction, and the acid used as catalyst. To recover the

270 methanol for its recycling as excess reagent in the esterification/transesterification reaction,

271 the heavy phase is first neutralized (CRV-102) by the addition of potassium hydroxide 272 obtaining a salt, *i.e.*, potassium sulphate, that is removed (X-101) and that may be considered 273 as a valuable by-product. Then, the neutralized stream is forwarded to a distillation column 274 (T-102) to recover the methanol (79% of the total used in the process) and to obtain a stream 275 of water with traces of glycerol (0.4 wt %) that might be recycled to the WWTP. As the 276 obtained biodiesel contains also non-saponifiable lipids, a crystallization fractionation is 277 applied to split biodiesel into a liquid (low-melting point) and a solid fraction with high 278 melting point, *i.e.*, sterols and/or waxes, achieving a product of more than 98% of FAME. 279 Particularly, the traditional fractionation consists of two stages, the crystallization under 280 strictly controlled cooling rate combined with gentle agitation, and the separation by filtration 281 (Knothe et al., 2005). This process was simulated by units V-106 and X-102 (see Fig. 2 and 282 Table 2), to coarsely estimate the costs derived from the energy and equipment requirements. 283 On the other hand, during the design of the process, energy integration strategies were 284 applied in order to reduce the energy consumption in certain stages of the process that were 285 specially energy consuming. For example, during the hexane recovery and the product 286 purification the streams that leave the separation units at high temperature are used to 287 exchange heat with the input streams so the heating and cooling requirements are reduced. 288

289 **2.3. Dry routes alternatives** 

The wet extraction route, optimised in this study, is compared with the dry route extraction where the sewage sludge is previously dried. The sewage sludge drying process was modelled in a similar way to that used for microalgae biomass harvesting, which consists in a twostaged dewatering process: centrifugation from a concentration of 5 to 20% of solids, and drying in a spray dryer till 95% of solids (Lassing et al., 2008). Two dry routes are considered in the economic comparison. On the one hand, the conventional (two-step) dry route, based on

296 operating data given by Zhang et al. (2013), uses hexane in a ratio of 10 L/kg dry sludge in a 297 mixture with methanol and acetone 3:1:1, achieving 96% of extraction efficiency at 50°C and 298 a residence time of 1 hour. After the solvent recovery, the esterification/transesterification 299 reaction is carried out using the same approach as in the wet route process. On the other hand, 300 a dry route alternative is also assessed where the extraction and the acid catalysed reaction 301 take place simultaneously, called *in situ* transesterification. This alternative was simulated 302 using the operation data detailed by Mondala et al. (2009) including a mixture of 12:1:3.3 303 methanol:sludge:hexane mass ratio and 5% (v/v) of H<sub>2</sub>SO<sub>4</sub> as catalyst in methanol. This 304 alternative reduces the number of equipment involved and eventually the reduction of the 305 costs associated is foreseen. The solvent recovery and product purification phases for both dry 306 extraction alternatives are similar to that described for the wet route, with the only difference 307 of impossibility of by-product formation during the *in situ* process. In this way, simulating the 308 three alternatives (wet rout, two-step dry rout and *in situ* dry route) under common operational 309 basis, *i.e.*, dry sludge composition, drying expenses, economic parameters, diesel 310 specifications, etc., the results obtained are valuable especially in terms of comparison. 311 The detailed process flow diagrams obtained with AspenHysys<sup>®</sup> simulation package, and 312 the characterization of the main material streams involved are presented in Fig. S1 and Fig. 313 S2, for the conventional (two-step) and *in situ* dry routes, respectively.

314

#### 315 **2.4. Automated economic evaluation**

The characterization of the process presented in this work is performed by the automated environmental evaluation tool (AEET) programmed in Matlab<sup>®</sup> R2010b (Torres et al., 2013a) taking advantage of the Component Object Module (COM) automation server capabilities to connect the simulator and Matlab<sup>®</sup>. In this tool the calculations are grouped into different modules, more precisely, it includes an inventory module that retrieves data from the process

321 simulation related with the inputs and outputs of materials and energy in the steady-state, 322 while the economic module computes the capital investment, operating costs and profitability 323 indicators. This tool was adapted to the presented case study; the particular complexity lay on 324 the necessity of analysing a high number of alternatives as result of altering operating 325 variables (solvent ratios, pH, residence time, etc.) and, in a greater extend, changes in 326 topological variables, such as the number of extraction stages and plant configurations (one 327 wet and two dry routes). The code of the module corresponding to the initialization of these 328 parameters in the simulation, called the specifications module, includes the identification of 329 the simulation file with the topological configuration to be assessed and the definition the set 330 of variables and constraints that the simulation should meet. Besides, the modules for the 331 calculation of the capital investment and manufacturing costs were complemented by a 332 specific cost estimation procedure for the equipment in the extraction stages. The automation 333 of the procedure makes the AEET a powerful tool for the evaluation of any process by the 334 acquisition of realistic data from the simulation case. Besides, it allows performing additional 335 analysis, emphasizing among others, generation and discrimination of alternatives (Torres et 336 al., 2013b), retrofit and sensitivity analysis and coupling with external optimization 337 algorithms (Torres et al., 2013a).

The profitability analysis module includes the calculation of the net present value (NPV) and the discounted payback period. However, taking into account that the purpose of this study is to determine whether the primary sludge can be a feasible feedstock for biodiesel production, the break-even price (BEP) is computed because it allows an easy comparison with the biodiesel main competitor, *i.e.*, fossil diesel. The calculations of the capital and production costs are based on Spain/European Union conditions (6% rate of interest for the capital investment and a plant life span of 20 years were assumed).

345 The total capital investment includes the fixed capital cost and the working capital cost, 346 where the second is usually a fraction of the first (15% is used in this work). The fixed capital 347 cost consists of the total bare module capital cost, the contingencies and fees and the auxiliary 348 facilities cost. In this work, the equipment module costing technique is used to estimate the 349 total bare module capital cost of the plant. This technique relates all costs back to the purchase 350 cost of equipment evaluated for some base conditions that imply that the units are fabricated 351 from the carbon steel and operated at near-ambient pressure. The deviation from these base 352 conditions are handled using multiplying factors that depend on the equipment type, the 353 system pressure and the materials of construction. Since cost estimation is a very specialized 354 subject the complexities of a detailed profitability study are out from the scope of this study. 355 For this reason, elements as pressure drops and potential safety issues are not itemized in the 356 procedure. The analysis is based on the conceptual design of the process alternatives to 357 compute an estimate of investment required and the rough cost of production in order to test 358 the feasibility of the proposed process (wet route), optimize the design and decide between 359 process alternatives.

360 The total manufacturing cost includes three different items: direct manufacturing cost (i.e., 361 raw materials, labour fees, utilities, maintenance and repairs, operating supplies, laboratory 362 charges and patents and royalties), fixed manufacturing cost (*i.e.*, overheads, packaging, 363 storage, local taxes, insurances and depreciation), and general expenses (*i.e.*, administration, 364 distribution and selling, and research and development). Detailed information about the 365 equipment module costing technique used to estimate the total bare module of the plant can be 366 found in the literature (Turton et al., 2003; Torres et al., 2013c). For more detailed 367 information, the prices of all raw materials, utilities used and by-product are listed in Table 368 S2.

369

**370 3. Results and discussion** 

The results of cost-effectiveness of the production of biodiesel, directly linked to the extraction of lipids are discussed in this section. Initially, the results of the production of biodiesel obtained from the different extraction configurations are evaluated (Section 3.1). Then, the economic results of the six extraction alternatives studied are discussed (Section 3.2). After that, the details of the best configuration are presented (Section 3.3). Finally, a comparison with production of biodiesel from dried sewage sludge, *i.e.*, conventional dry route and *in situ* dry route, and BEPs comparison are debated (Section 3.4 and 3.5).

378

#### 379 **3.1. Effect of the configurations of lipid extraction on biodiesel production**

The extraction of lipids from sewage sludge is the most important point, as the reaction of esterification/transesterification is not optimised and its procedure is fixed (see subsection 2.1.3 and 2.2.2). The optimisation of the extraction was performed for six different configurations, presented in Table 1. These configurations were selected as the best alternatives regarding the experimental results obtained in laboratory work (Olkiewicz et al., 2014).

386 Fig. 3a presents the annual values of biodiesel production obtained for the six 387 configurations depending on the number of mixers used for the lipid extraction. As it can be 388 seen all configurations present the same behaviour, the production of biodiesel grows with the 389 number of mixers used during the extraction stages. The optimisation was limited to 5 mixers 390 as the costs of the extraction process is directly related to the number of mixers and, the 391 increase in the quantity of produced biodiesel by the last units is much smaller in comparison 392 with the increase of the costs (law of diminishing returns). As stated in subsection 2.1.2., 393 more than 90% of lipids were extracted after 5 consecutive extraction steps in the laboratory 394 work.

395 As shown in Fig. 3a, the highest productions of biodiesel were attained with 5 mixers for 396 the configurations CS1, CS2 and CS3, approximately 4600 tonnes of biodiesel per year. 397 Slightly lower amount, approximately 4400 of biodiesel per year, was achieved by 398 configuration CS6. Comparing the extraction configuration CS3 with CS6, where the only 399 difference is the sludge pH, it can be concluded that the increase of pH from 2 to 4 does not 400 have significant impact on the biodiesel production. Additionally, the use of less acid in the 401 process will reduce the biodiesel production cost (discussed in further detail in section 3.2). 402 The lowest results were obtained in the case of CS4 and CS5, approximately 3600 and 3300 403 tonnes of biodiesel per year, respectively. These two configurations have a mixing time of 404 only 20 min, suggesting that the extraction time seems to be the most important parameter. 405 The lower the extraction time, the lower the yield of lipid extracted and, therefore the lower 406 the amount of biodiesel produced. However, in the case of CS1, also 20 min of mixing, the 407 time is compensated by the higher amount of solvent, ratio sludge/hexane of 1/2. This 408 configuration is able to extract high quantities of lipids, but by using a high quantity of 409 solvent, which is not economically favourable for the process (discussed in further detail in 410 section 3.2). The detailed values of biodiesel production for each configuration in each 411 extraction stage are presented in Table S3 and discussed in the following section.

412

#### 413 **3.2. Optimisation of lipid extraction configuration**

In order to find the most profitable extraction conditions in biodiesel production from
liquid primary sludge, the value of break-even price (BEP), for all configurations tested at
each number of mixers was evaluated. The BEP is defined as:

417 
$$BEP(\$/t) = \frac{\text{Manufacturing costs}(\$/y) - \text{Byproductssales}(\$/y)}{\text{Production of biodiesel}(t/y)}$$
(1)

418 Where the manufacturing costs is the total manufacturing cost as defined in section 2.4; by-

419 products sales are the revenue from the sale of by-products generated during the process (in

this case only one by-product is generated, *i.e.*, potassium sulfate; production of biodiesel in
tonnes of biodiesel produced per year.

422 Fig. 3b presents the evolution of the break-even price for the six configurations studied as a 423 function of the number of mixers used. The six configurations show exactly the same 424 behaviour, the BEP decreases from one to two mixers, giving a minimum value for two 425 mixers, and then increases constantly until the five mixers. The detailed values of break-even 426 price as well as the investment and manufacturing cost for each configuration in each 427 extraction stage are presented in Table S3. As shown in Table S3, independently of the 428 configuration used, the total manufacturing costs increase faster in each stage than the 429 production of biodiesel. The low increment of biodiesel production from 2 to 5 mixers is not 430 compensated by the high increase in the manufacturing costs, resulting in continuous increase 431 of BEP from 2 to 5 mixers (Fig. 3b). Furthermore, using more extraction stages (mixers) 432 entails also the increase of total investment cost, which consequently prolongs the payback 433 period, *i.e.*, time to achieve benefits by the plant. As shown in Fig. 3b, the results of BEP 434 indicates that for all configurations tested, two extraction stages (mixers) are sufficient to 435 extract enough lipids to make biodiesel production profitable. Although additional mixers 436 increase the amount of lipids and, thus the amount of biodiesel in the process, the installation 437 of more than two mixing equipment is not cost-effective.

As presented in Fig. 3b and in more detail in Table S3, the lowest BEP (1232 \$/t) and therefore the best profitability was obtained by the configuration CS6 using two mixers. The detailed economic results for each configuration using two mixers are shown in Table S4. The best configuration, *i.e.*, CS6, works using two times less solvent than CS2 and CS4 and four times less solvent than CS1. Consequently, the cost of total manufacturing for the CS6 was lower, mainly due to the lower amount of solvent that has to be handled (and recovered, and recycled). Despite the fact that the biodiesel production was increased by 19% in the case of

445 CS1 and CS2 compared to CS6, the manufacturing cost increased by about 28% and 26% 446 respectively, leading to higher BEP. In addition, the total investment cost related mainly to 447 storage and separation was also lower in the case of CS6 than the CS1, CS2 and CS4. 448 Although the cost of mixing equipment, in the case of CS1 and CS4 was lower than for the 449 best configuration (CS6), due to the lower residence time which permits to use smaller 450 equipment, the cost of other investments was much higher resulting in higher overall 451 investment cost. Continuing with the comparison, the configuration CS5, where the only 452 difference is lower residence time with respect to CS6, allows a reduction in total 453 manufacturing cost of 12% and also in investment cost of 10%. However, the reduction of 454 extraction time resulted in a significant decrease of biodiesel production by around 25%, 455 giving therefore higher BEP than the best configuration (CS6). Looking at the configuration 456 CS3, which has only lower pH than the configuration CS6, the investment cost was almost 457 equal but the manufacturing cost was slightly increased by about 7% due to the higher amount 458 of acid required. The increased of manufacturing cost for this configuration was not 459 compensated by the insignificant rise of biodiesel production ( $\approx 4\%$ ), thus having higher BEP 460 than CS6. It can be concluded that the optimized configuration, *i.e.*, CS6, does not only allow 461 to minimize the biodiesel production cost, but also to reduce the solvent and acid use, making 462 the process less detrimental for equipment and environment.

463

#### 464 **3.3. Details of the best configuration**

The optimized extraction process (*i.e.*, configuration CS6 with two mixers) allows the plant to produce 3546 tonnes of biodiesel per year, with the minimum biodiesel cost of 1232 \$/t. The biodiesel production plant is divided into four processes: extraction (lipid extraction from liquid sludge), recovery (recovery of lipids from solvent), reaction (biodiesel production), and purification (biodiesel separation, purification and catalyst neutralization). In

470 order to establish the cost requirement of each process in the whole biodiesel production 471 plant, the detailed economic analysis was distributed and the results are presented in Table 3. 472 Regarding the total investment cost, the extraction step is responsible for 54% of the total 473 investment followed by purification (29%), recovery (10%) and reaction (7%). With respect 474 to manufacturing cost, the extraction step is also the most expensive, representing 48% of the 475 total, while purification, recovery and reaction account for 24%, 7% and 21%, respectively. 476 This economic analysis of biodiesel production demonstrates that the proposed direct 477 extraction from liquid sludge is the most cost intensive step for the investment as well as for 478 the manufacturing of the process. Therefore, any improvement of the lipid extraction step 479 would have a high impact on the final profitability of the process and finally on the biodiesel 480 price. On the other hand, the purification step is the second expensive one due to the high 481 amount of water used for the separation and purification of the final product. It is well known 482 that in the conventional synthesis of biodiesel, using acid or basic catalyst, the separation and 483 purification of final product from catalyst is difficult and required high amount of energy 484 (Siddiquee et al., 2011; Andreani and Rocha, 2011). Thus, the improvement of the separation 485 step by using better catalyst for the reaction, easy to separate and reuse, could also reduce the 486 final cost of biodiesel production from liquid sludge. However, this implies the employment 487 of new biodiesel manufacturing technologies based on *e.g.*, heterogeneous catalysis 488 (Siddiquee et al., 2011) or ionic liquid catalyst (Andreani and Rocha, 2011), capable to 489 overcome the problems related to conventional catalyst but that, to our knowledge, are not 490 ready to be commercially used (or are not economically attractive).

491

#### 492 **3.4.** Comparison with dry routes processes

In order to decide on the most economically favourable process, the economic data of thewet route *i.e.*, the optimized biodiesel production process with extraction configuration CS6,

was compared to conventional (two-step) dry route (extraction from dry sludge and
consequent conversion of lipids into biodiesel) and to *in situ* dry route (direct production of
biodiesel from dry sludge). The detailed process flow diagrams were modelled with
AspenHysys<sup>®</sup> simulation package, and the characterization of the main material streams
involved are presented in Fig. 2, Fig. S1 and Fig. S2 for wet, dry conventional and dry *in situ*routes, respectively.

501 As shown in Table 4 the wet route process is much more cost-effective than both dry 502 routes. With respect to investment cost, both dry routes have a total investment cost around 503 twice higher than the wet route. The wet route process requires large equipment size for lipid 504 extraction step due to large volume of liquid sludge associated, and consequently with a 505 higher mixing cost. However, the cost of dewatering and drying equipment in the dry routes is 506 six times higher as compared to mixing equipment used in wet route, making the total 507 investment cost of wet route more profitable. Regarding the manufacturing cost, dry processes 508 are also more costly, with the total manufacturing cost twice higher in the case of dry 509 conventional and approximately four times higher in the case of dry in situ (in both cases 510 compared with the wet route). The much higher manufacturing cost is mainly related to the 511 cost required for sludge drying which represents about 20% (334 \$/t of biodiesel) and 10% 512 (314 \$/t of biodiesel) of the total biodiesel cost for dry conventional and dry *in situ*, 513 respectively. As can be seen in Table 4, although the biodiesel production is higher in both 514 dry routes, owing to more efficient extraction of lipid from dry sludge than from wet one, the 515 much larger increase of manufacturing cost results in higher BEP as compared to wet process. 516 Therefore, the proposed process of biodiesel production using directly liquid sludge is more 517 cost-effective than the conventional dry or *in situ* dry routes, giving the break-even price of 518 1232 \$/t, 1656 \$/t and 3145 \$/t, respectively.

519 Comparing the dry routes, the conventional two-step dry route is more profitable than the 520 *in situ* dry route. This result is in agreement with other authors who demonstrated that two-521 step process is more energetically favourable than the one-step *in situ* (Zhang et al, 2013; 522 Pastore et al., 2013).

523 It is interesting to compare the BEP of biodiesel produced following the *in situ* approach in 524 this work (3145 \$/t) with the BEP obtained by Mondala et al. (2009) (970 \$/t), as they differ 525 widely. On the one hand, the cost of sludge drying used in the present study according to 526 Lassing et al. (2008) was higher than the cost presented by Mondala et al. (2009), who based 527 their calculation on the cost of sludge drying given by Dufreche et al. (2007). Both authors 528 did not described details about the method and equipment used for drying, presenting such a 529 favourable drying cost (2.1 \$/t of wet sludge) as compared to other authors, 4.3 and 52.2 \$/t of 530 wet sludge reported by Lassing et al. (2008) and Kwon et al. (2012), respectively. Therefore, 531 the sewage sludge drying process was modelled in a similar way to the totally defined 532 procedure by Lassing et al. (2008), increasing the BEP as compared to the BEP obtained by 533 Mondala et al. (2009) for the same process. On the other hand, the purification of the final 534 product from non-saponifiable lipids, which was performed in the present study, but was not 535 mentioned in other works (Dufreche et al., 2007; Mondala et al., 2009; Zhang et al., 2013) 536 also contributes to the cost raising. Furthermore, the costs of raw materials presented by 537 Mondala et al. (2009) are also lower than the presented here and should be included in order 538 to compare the approaches. Finally, regarding the *in situ* studies (Dufreche et al., 2007; 539 Mondala et al., 2009), only economic evaluation was performed, however the whole process 540 was not modelled and simulated. The simulation of process scale-up leads to get a full techno-541 economic evaluation of the plant and finally more realistic cost of biodiesel is obtained.

542

#### 543 **3.5. Break-even prices comparison**

The optimised BEP of biodiesel produced from liquid primary sludge was estimated to be 1232 \$/t. This value is comparatively lower than the current price of fossil diesel in Europe, estimated to be 1376 \$/t on January 2016 (IEA, 2016). Therefore, it is evident that the biodiesel obtained from municipal primary sludge has a high potential to economically compete with fossil diesel.

549 On the other hand, it is interesting to compare the BEP of biodiesel from sewage sludge to 550 that from microalgae, the alternative biodiesel feedstock that has been investigated 551 extensively for a very long time but whose industrialisation is not yet economically viable 552 (Rios et al., 2013). The BEP of biodiesel produced from sewage sludge is lower than the 553 obtained from microalgae in different studies; 1344 \$/t calculated with the best scenario 554 available (Lassing et al., 2008), 2953 \$/t for a biodiesel with microalgae produced in open 555 ponds (Davis et al., 2011), and 5700 \$/t obtained in an exhaustive study (Ríos et al., 2013). As 556 municipal sewage sludge is a waste, provided for free from WWTP, the cost of biomass 557 production is eliminated; which is not the case in microalgae biodiesel manufacturing, where 558 biomass production accounts for 65% of the overall cost (Ríos et al., 2013).

559

#### 560 **4. Conclusions**

561 The detailed techno-economic study indicates that the proposed biodiesel production 562 process from liquid primary sludge is economically feasible and more cost-effective than 563 alternatives from dry sludge. The required biodiesel selling price for the optimised lipid 564 extraction step was estimated to be 1232 \$/t, which is lower than the current cost of fossil 565 diesel and the cost of biodiesel from microalgae. Thus, the municipal sludge has a large 566 potential as cost-competitive, plentiful and non-edible feedstock for biodiesel production. 567 Additionally, further improvement of the proposed process (especially lipid extraction from 568 liquid sludge and biodiesel purification) could lower the biodiesel price even more.

569

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- 577

#### 578 **References**

- 579 1. Andreani, L., Rocha, J.D., 2012. Use of ionic liquids in biodiesel production: a review.
- 580 Braz. J. Chem. Eng. 29, 1–13.
- 581 2. Atabani, A.E., Silitonga, A.S., Badruddin, I.A., Mahlia, T.M.I., Masjuki, H.H., Mekhilef,
- 582 S., 2012. A comprehensive review on biodiesel as an alternative energy resource and its
- 583 characteristics. Renew. Sustain. Energy Rev. 16, 2070–93.
- 584 3. Davis, R., Aden, A., Pienkos, P.T., 2011. Techno-economic analysis of autotrophic
- 585 microalgae for fuel production. Appl. Energy 88, 3524–31.
- 586 4. Dufreche, S., Hernandez, R., French, T., Sparks, D., Zappi, M., Alley, E., 2007. Extraction
- 587 of Lipid from Municipal Wastewater Plant Microorganisms for Production of Biodiesel. J.
- 588 Am. Oil Chem. Soc. 84, 181–7.
- 589 5. Ghosh, P., 1999. Prediction of vapor-liquid equilibria using Peng-Robinson and Soave-
- 590 Redlich-Kwong equations of state. Chem. Ing. Technol. 22, 379-399.
- 591 6. Huynh, L.H., Kasim, N.S., Ju, Y.H., 2010. Extraction and analysis of neutral lipid from
- 592 activated sludge with and without sub-critical water pre-treatment. Bioresour. Technol. 101,
- 593 8891–6.

- 594 7. IEA International Energy Agency. Monthly oil prices statistic, January 2016. Available
- 595 from: http://www.iea.org/media/statistics/surveys/prices/mps.pdf
- 596 8. Khan, T.M.Y., Atabani, A.E., Badruddin, I.A., Badarudin, A., Khayoon, M.S.,
- 597 Triwahyono, S., 2014. Recent scenario and technologies to utilize non-edible oils for
- 598 biodiesel production. Renew. Sustain. Energy Rev. 37, 840–51.
- 599 9. Knothe, G., Van Gerpen, J., Krahl, J., 2005. The biodiesel handbook. AOCS Press,
- 600 Champaign, Illinois.
- 601 10. Kwon, E.E., Kim, S., Jeon, Y.J., Yi, H., 2012. Biodiesel Production from Sewage Sludge:
- 602 New Paradigm for Mining Energy from Municipal Hazardous Material. Environ. Sci.
- 603 Technol. 46, 10222–10228.
- 11. Lassing, M., Hulteberg, C., Mårtensson, P., Karlsson, H.T., Børre, E.O., Børresen, T.,
- 605 Svensson, M., Eklund, H., 2008. Final Report on Biodiesel Production from Microalgae: A
- 606 Feasibility Study. Presented to StatoilHydro ASA, Oslo.
- 607 12. Mondala, A., Liang, K., Toghiani, H., Hernandez, R., French, T., 2009. Biodiesel
- 608 production by *in situ* transesterification of municipal primary and secondary sludges.
- 609 Bioresour. Technol. 100, 1203–10.
- 610 13. Olkiewicz, M., Fortuny, A., Stüber, F., Fabregat, A., Font, J., Bengoa, C., 2015a. Effects
- 611 of pre-treatments on the lipid extraction and biodiesel production from municipal WWTP
- 612 sludge. Fuel 141, 250–57.
- 613 14. Olkiewicz, M., Plechkova, N.V., Fabregat, A., Stüber, F., Fortuny, A., Font, J., Bengoa,
- 614 C., 2015b. Efficient extraction of lipids from primary sewage sludge using ionic liquids for
- 615 biodiesel production. Sep. Purif. Technol. 153, 118–125.
- 616 15. Olkiewicz, M., Caporgno, M.P., Fortuny, A., Stüber, F., Fabregat, A., Font, J., Bengoa,
- 617 C., 2014. Direct liquid–liquid extraction of lipid from municipal sewage sludge for biodiesel
- 618 production. Fuel Process. Technol. 128, 331–8.

- 619 16. Olkiewicz, M., Fortuny, A., Stüber, F., Fabregat, A., Font, J., Bengoa, C., 2012.
- 620 Evaluation of different sludges from WWTP as a potential source for biodiesel production.
- 621 Procedia Eng. 42, 695–706.
- 622 17. Pastore, C., Lopez, A., Lotito, V., Mascolo, G., 2013. Biodiesel from dewatered
- 623 wastewater sludge: A two-step process for a more advantageous production. Chemosphere 92,
- 624 667–73.
- 18. Peng, D., Robinson, D.B., 1976. A new two-constant equation of state. Ind. Eng. Fundam.1, 59-64.
- 627 19. Perry, R.H., Green, D.W., 1997. Perry's Chemical Engineers' Handbook. MacGraw-Hill,
- 628 7<sup>th</sup> ed. New York.
- 629 20. Pokoo-Aikins, G., Heath, A., Mentzer, R.A., Mannan, M.S., Rogers, W.J., El-Halwagi,
- 630 M.M., 2010. A multi-criteria approach to screening alternatives for converting sewage sludge
- 631 to biodiesel. J. Loss Prevent. Proc. 23, 412–20.
- 632 21. Ríos, S.D., Torres, C.M., Torras, C., Salvadó, J., Mateo-Sanz, J.M., Jiménez, L., 2013.
- 633 Microalgae-based biodiesel: Economic analysis of downstream process realistic scenarios.
- 634 Bioresour. Technol. 136, 617-25.
- 635 22. Siddiquee, M.N., Rohani, S., 2011. Experimental analysis of lipid extraction and biodiesel
- 636 production from wastewater sludge. Fuel Process. Technol. 92, 2241–51.
- 637 23. Siddiquee, M.N., Hossein, K., Rohani, S., 2011. Biodiesel production from the lipid of
- 638 wastewater sludge using an acidic heterogeneous catalyst. Chem. Eng. Technol. 34, 1983–8.
- 639 24. Torres, C.M., Gadalla, M., Mateo-Sanz, J.M., Jiménez, L., 2013a. An automated
- 640 environmental and economic evaluation methodology for the optimization of a sour water
- 641 stripping plant. J. Clean. Prod. 44, 56-8.

- 642 25. Torres, C.M., Ríos, S.D., Torras, C., Salvadó, J., Mateo-Sanz, J.M., Jiménez, L., 2013b.
- 643 Microalgae-based biodiesel: A multicriteria analysis of production process using realistic
- 644 scenarios. Bioresour. Technol. 147, 7-16.
- 645 26. Torres, C.M., Ríos, S.D., Torras, C., Salvadó, J., Mateo-Sanz, J.M., Jiménez, L., 2013c.
- 646 Sustainability analysis of biodiesel production from Cynara Cardunculus crop. Fuel 111, 535-
- 647 42.
- 648 27. Turton, R., Bailie, R.C., Whiting, W.B., Shaeiwitz, J.A., 2003. Analysis, synthesis and
- 649 design of chemical processes. Prentice Hall, 2<sup>nd</sup> ed. New Jersey.
- 650 28. Tyagi, V.K., Lo, S.-L., 2013. Sludge: A waste or renewable source for energy and
- resources recovery? Renew. Sustain. Energy Rev. 25, 708–728.
- 652 29. Willson, R.M., Wiesman, Z., Brenner, A., 2010. Analyzing alternative bio-waste
- 653 feedstocks for potential biodiesel production using time domain (TD)-NMR. Waste Manage.654 30, 1881–8.
- 30. Yi, W., Sha, F., Xiaojuan, B., Jingchan, Z., Siqing, X., 2016. Scum sludge as a potential
- 656 feedstock for biodiesel production from wastewater treatment plants. Waste Manage. 47, 91–
- 657 97.
- 658 31. Zhang, X., Yan, S., Tyagi, R.D., Surampalli, R.Y., 2013. Energy balance and greenhouse
- 659 gas emissions of biodiesel production from oil derived from wastewater and wastewater
- 660 sludge. Renew. Energy 55, 392–403.
- 661

### 662 Figure captions

- **Fig. 1.** Scheme of the current WWTP of Reus (Tarragona, Spain) (a) and the biodiesel
- 664 production process (b).
- 665 **Fig. 2.** Flow diagram of the AspenHysys<sup>®</sup> simulation model for the biodiesel production
- 666 process from liquid primary sludge (wet route).
- **Fig. 3.** Influence of number of stages in the extraction of lipids from liquid primary sewage
- sludge (a) on the biodiesel production and (b) on the break-even price.

|               |                               |    | Mixing time<br>(min) | Fitting          | R <sup>2</sup><br>Concentration |       |       |
|---------------|-------------------------------|----|----------------------|------------------|---------------------------------|-------|-------|
| Configuration | Sludge/hexane<br>volume ratio | рН |                      | Conc             |                                 |       |       |
|               |                               |    |                      | Low              | High                            | Low   | High  |
| CS1           | 1/2                           | 2  | 20                   | y=0.2554x+7e-5   | y=0.0001exp(1483.6x)            | 0.916 | 0.989 |
| CS2           | 1/1                           | 2  | 60                   | y=0.8968x-0.0001 | 0.967                           | 0.990 |       |
| CS3           | 2/1                           | 2  | 60                   | y=3.97           | 0.997                           |       |       |
| CS4           | 1/1                           | 4  | 20                   | y=7e-5e          | 0.954                           |       |       |
| CS5           | 2/1                           | 4  | 20                   | y=-238.81x2-     | 0.980                           |       |       |
| CS6           | 2/1                           | 4  | 60                   | y=3.97           | 0.997                           |       |       |

**Table 1.** Configurations of the lipid extraction process.

| Name     | Equipment            | Purpose  |
|----------|----------------------|--|
| V-100    | Tank                 | Reception of sludge and mixing with acid   |
| CSTR-100 | Mixing unit          | Representation of time and mixing needed for extraction (1st stage)  |
| X-103    | Separation unit      | Representation of the separation controlled by<br>"Equilibrium" (1st stage)  |
| CSTR-101 | Mixing unit          | Representation of time and mixing needed for extraction (2nd stage)  |
| X-104    | Separation unit      | Representation of the separation controlled by<br>"Equilibrium" (2nd stage)  |
| V-101    | Separator            | Flash separation of solvent from lipids after extraction   |
| V-102    | Tank                 | Feed tank of hexane  |
| MIX-102  | Mixing unit          | Collect hexane streams from the recovery equipment   |
| V-103    | Tank                 | Feed tank of methanol  |
| MIX-101  | Mixing unit          | Mixing methanol with sulfuric acid to reach the<br>appropriate reaction conditions (controlled by "Reaction<br>inlet") |
| CRV-100  | Reactor              | Transesterification of lipids with methanol and sulfuric acid  |
| V-104    | 3-phase<br>separator | Separating light and aqueous phase before purification   |
| CRV-102  | Reactor              | Neutralizing acid before recovering methanol from aqueous phase  |
| X-101    | Solid separator      | Separating neutralization product  |
| T-102    | Distillation column  | Recovery of methanol from the aqueous phase  |
| V-107    | Separator            | Purification of light phase removing remaining methanol from FAME  |
| V-106    | Refrigeration tank   | Adjusting temperature of product to achieve winterization  |
| X-102    | Solid separator      | Removing heavy fractions (waxes) from FAME product   |

**Table 2.** List of the main equipments used in the AspenHysys<sup>®</sup> simulation model for the biodiesel production process from liquid primary sludge (Fig. 2)

|          |                                    | Unit (\$)  |          |          |              |         |  |  |  |  |  |
|----------|------------------------------------|------------|----------|----------|--------------|---------|--|--|--|--|--|
|          | Item                               | Extraction | Recovery | Reaction | Purification | Total   |  |  |  |  |  |
|          | Total bare module                  | 2270511    | 410244   | 316032   | 1229418      | 4226204 |  |  |  |  |  |
|          | Centrifugation and drying          | 0          | 0        | 0        | 0            | 0       |  |  |  |  |  |
|          | Reactors                           | 0          | 0        | 129214   | 33843        | 163057  |  |  |  |  |  |
|          | Distillation columns               | 0          | 0        | 0        | 687314       | 687314  |  |  |  |  |  |
| sts (\$) | Flush & other separation equipment | 249299     | 63451    | 0        | 74147        | 386897  |  |  |  |  |  |
| nt co    | Mixing units                       | 942909     | 96465    | 9563     | 13670        | 1062606 |  |  |  |  |  |
| mer      | Heat exchangers                    | 0          | 239677   | 0        | 227158       | 466835  |  |  |  |  |  |
| Ivest    | Pumps                              | 26143      | 10652    | 4035     | 13823        | 54652   |  |  |  |  |  |
| II       | Storage                            | 1052160    | 0        | 173100   | 173340       | 1398600 |  |  |  |  |  |
|          | Fix. cap. costs (bare, cont. aux.) | 3482963    | 629315   | 484793   | 1885927      | 6482997 |  |  |  |  |  |
|          | Working capital                    | 522444     | 94397    | 72719    | 282889       | 972450  |  |  |  |  |  |
|          | Total investment costs             | 4005408    | 723712   | 557512   | 2168816      | 7455447 |  |  |  |  |  |
|          | Raw materials                      | 271648     | 0        | 313255   | 53435        | 638338  |  |  |  |  |  |
|          | Utilities                          | 269715     | 44350    | 60       | 53497        | 367623  |  |  |  |  |  |
| y)       | Steam                              | 0          | 44346    | 0        | 50944        | 95290   |  |  |  |  |  |
| ; (\$/   | Cooling                            | 0          | 0        | 59       | 2378         | 2437    |  |  |  |  |  |
| costs    | Electricity                        | 269715     | 5        | 1        | 1            | 269722  |  |  |  |  |  |
| ing (    | Makeup water                       | 0          | 0        | 0        | 174          | 174     |  |  |  |  |  |
| ctur     | Fuel (spray dryer)                 | 0          | 0        | 0        | 0            | 0       |  |  |  |  |  |
| nufa     | Operation labour                   | 270000     | 40500    | 155250   | 195750       | 661500  |  |  |  |  |  |
| Maı      | Direct manufacturing cost          | 811363     | 84850    | 468565   | 302683       | 1667461 |  |  |  |  |  |
|          | Overhead (fix. & gen. exp. costs)  | 1312587    | 210534   | 440997   | 753629       | 2717746 |  |  |  |  |  |
|          | Total manufacturing costs          | 2123950    | 295384   | 909562   | 1056311      | 4385208 |  |  |  |  |  |

# **Table 3.** Disaggregated results for the optimal configuration CS6<sup>1</sup> as a function of the process steps.

<sup>1</sup> ratio sludge/hexane 2/1, pH 4, 2 mixers and 60 min of mixing

|         |   | Process (\$)    |                             |                              |  |  |  |  |  |
|---------|---|-----------------|-----------------------------|------------------------------|--|--|--|--|--|
|         | Item  | CS6 (wet route) | Dry route<br>(conventional) | Dry route ( <i>in situ</i> ) |  |  |  |  |  |
|         | Total bare module                                       | 4226204         | 9780923                     | 11290902                     |  |  |  |  |  |
|         | Centrifugation and drying                               | 0               | 6400000                     | 6400000                      |  |  |  |  |  |
|         | Reactors  | 163057          | 189218                      | 1196930                      |  |  |  |  |  |
|         | Distillation columns                                    | 687314          | 688299                      | 316706                       |  |  |  |  |  |
| s (\$   | Flush & other separation equipment                      | 386897          | 384994                      | 437067                       |  |  |  |  |  |
| cost    | Mixing units  | 1062606         | 255744                      | 339856                       |  |  |  |  |  |
| nent    | Heat exchangers   | 466835          | 420554                      | 265841                       |  |  |  |  |  |
| estn    | Pumps   | 54652           | 43513                       | 56962                        |  |  |  |  |  |
| Inv     | Storage   | 1398600         | 1398600                     | 2277540                      |  |  |  |  |  |
|         | Fix. cap. costs (bare, cont. aux.)                      | 6482997         | 15003935                    | 17320243                     |  |  |  |  |  |
|         | Working capital   | 972450          | 2250590                     | 2598036                      |  |  |  |  |  |
|         | Total investment costs                                  | 7455447         | 17254525                    | 19918280                     |  |  |  |  |  |
|         | Raw materials   | 638338          | 978234                      | 3910752                      |  |  |  |  |  |
|         | Utilities   | 367623          | 2617825                     | 5263030                      |  |  |  |  |  |
| -       | Steam   | 95290           | 558431                      | 3075539                      |  |  |  |  |  |
| (\$/y)  | Cooling   | 2437            | 28423                       | 156467                       |  |  |  |  |  |
| osts (  | Electricity   | 269722          | 92086                       | 92363                        |  |  |  |  |  |
| ရာ ငင   | Makeup water  | 174             | 245                         | 21                           |  |  |  |  |  |
| acturin | Drying: Fuel (spray dryer) and Electricity (centrifuge) | 0               | 1938641                     | 1938641                      |  |  |  |  |  |
| anuf    | Operation labour  | 661500          | 717750                      | 508500                       |  |  |  |  |  |
| Μ       | Direct manufacturing cost                               | 1667461         | 4313809                     | 9682283                      |  |  |  |  |  |
|         | Overhead (fix. & gen. exp. costs)                       | 2717746         | 5174616                     | 6574965                      |  |  |  |  |  |
|         | Total manufacturing costs                               | 4385208         | 9488425                     | 16257248                     |  |  |  |  |  |
|         | Biodiesel production (t/y)                              | 3546            | 4700                        | 4425                         |  |  |  |  |  |
|         | Break-even price (BEP) (\$/t)                           | 1232            | 2014                        | 3674                         |  |  |  |  |  |

# **Table 4.** Comparison of optimal configuration $CS6^1$ (wet route) with dry routes.

<sup>1</sup> ratio sludge/hexane 2/1, pH 4, 2 mixers and 60 min of mixing





Fig. 2.



|                                | Primary |         |         |         |         |         |        |        |        |        |        |        |        |        | FAME   | Water  | K <sub>2</sub> SO <sub>4</sub> |
|--------------------------------|---------|---------|---------|---------|---------|---------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------------------------------|
| Stream name                    | sludge  | F       | Е       | R       | S       | S012    | S013   | S015   | S017   | S019   | S020   | S025   | S031   | S032   | 98%    | 98%    | 99%                            |
| Temperature (°C)               | 25.0    | 24.9    | 25.9    | 31.7    | 60.6    | 61.8    | 89.3   | 35.0   | 59.6   | 60.0   | 61.3   | 168.9  | 41.4   | 69.7   | 5.0    | 35.0   | 41.4                           |
| Pressure (kPa)                 | 101.3   | 115.0   | 105.0   | 105.0   | 120.0   | 80.0    | 101.3  | 101.0  | 101.0  | 101.0  | 101.0  | 80.0   | 101.0  | 101.0  | 101.3  | 110.0  | 101.0                          |
| Mass Flow (kg/h)               | 60000.0 | 60319.6 | 19925.3 | 59858.6 | 19464.9 | 19454.1 | 588.4  | 501.0  | 292.1  | 880.6  | 618.2  | 117.2  | 610.6  | 224.2  | 406.4  | 386.5  | 11.2                           |
| Components mass fracti         | ons     |         |         |         |         |         |        |        |        |        |        |        |        |        |        |        |                                |
| H <sub>2</sub> O               | 0.9600  | 0.9549  | 0.0000  | 0.9623  | 0.0001  | 0.0001  | 0.0000 | 0.0001 | 0.0360 | 0.0409 | 0.0019 | 0.0098 | 0.6376 | 0.0466 | 0.0001 | 0.9804 | 0.0001                         |
| Palmitic acid                  | 0.0084  | 0.0084  | 0.0183  | 0.0024  | 0.0000  | 0.0000  | 0.6167 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| Triolein                       | 0.0004  | 0.0003  | 0.0008  | 0.0001  | 0.0000  | 0.0000  | 0.0257 | 0.0003 | 0.0000 | 0.0002 | 0.0002 | 0.0000 | 0.0000 | 0.0000 | 0.0004 | 0.0000 | 0.0000                         |
| Rest Biomass                   | 0.0205  | 0.0204  | 0.0000  | 0.0205  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| Ash                            | 0.0086  | 0.0085  | 0.0000  | 0.0086  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| Non-saponifiable lipids        | 0.0022  | 0.0022  | 0.0048  | 0.0006  | 0.0000  | 0.0000  | 0.1609 | 0.1887 | 0.0000 | 0.1075 | 0.1531 | 0.0010 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| HCl                            | 0.0000  | 0.0053  | 0.0000  | 0.0053  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| n-Hexane                       | 0.0000  | 0.0000  | 0.9762  | 0.0000  | 0.9993  | 0.9993  | 0.1967 | 0.0218 | 0.0000 | 0.1314 | 0.1872 | 0.8943 | 0.0000 | 0.0000 | 0.0169 | 0.0000 | 0.0000                         |
| Methanol                       | 0.0000  | 0.0000  | 0.0000  | 0.0001  | 0.0004  | 0.0004  | 0.0000 | 0.0008 | 0.9425 | 0.2593 | 0.0140 | 0.0702 | 0.3598 | 0.9534 | 0.0010 | 0.0156 | 0.0000                         |
| $H_2SO_4$                      | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0215 | 0.0071 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| M-Palmitate                    | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0001  | 0.0001  | 0.0000 | 0.7582 | 0.0000 | 0.4346 | 0.6191 | 0.0244 | 0.0000 | 0.0000 | 0.9446 | 0.0000 | 0.0000                         |
| M-oleate                       | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0001 | 0.0301 | 0.0000 | 0.0171 | 0.0244 | 0.0003 | 0.0000 | 0.0000 | 0.0371 | 0.0000 | 0.0000                         |
| Glycerol                       | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0018 | 0.0000 | 0.0000 | 0.0026 | 0.0000 | 0.0000 | 0.0040 | 0.0000                         |
| КОН                            | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000                         |
| K <sub>2</sub> SO <sub>4</sub> | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000  | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.0000 | 0.9999                         |







**(b)** 

