

Article

Determination of Synthetic Cathinones in Urine and Oral Fluid by Liquid Chromatography High-Resolution Mass Spectrometry and Low-Resolution Mass Spectrometry: A Method Comparison

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Abstract: Synthetic cathinones have become very popular recreational drugs. Therefore, determining them in biological samples is now a matter of concern. In recent years, different methods that have been developed can determine these drugs at low-concentration levels. In general, liquid chromatography mass spectrometry detection plays an important role in these methods and the trend is to use low-resolution and high-resolution mass spectrometry. In this article, for the first time, we compare these two analyzers using an Orbitrap and a triple quadrupole mass spectrometer in order to determine a group of synthetic cathinones in urine and oral fluid samples. For this comparison, we evaluated and compared different parameters: Method detection and quantification limits, linearity, apparent recoveries, matrix effect, repeatability (intra-day), reproducibility (inter-day), and accuracy. Similar results were obtained for the two analyzers for the apparent recoveries and matrix effect. However, triple quadrupole showed higher sensitivity compared to Orbitrap for both urine and oral fluid samples. The quantification limits in urine and the detection limits in saliva were two times lower for triple quadrupole. Finally, when blind samples were analyzed to study the accuracy, similar results were obtained for both analyzers.

Keywords: synthetic cathinones; solid-phase extraction; tandem mass spectrometry; high-resolution mass spectrometry; biological samples

1. Introduction

Synthetic cathinones are an important group among the new psychoactive substances (NPS) that have emerged in recent years as some of the most common drugs around the world [1,2]. These compounds are synthetized from the molecule cathinone, a natural alkaloid that can be found in the *Catha edulis* plant [3]. Although they were first synthetized for medical purposes, they began being used as recreational drugs and have emerged as an important alternative to other worldwide popular drugs such as cocaine, amphetamine, or ecstasy [3]. This is because they have similar effects to some of these drugs, such as paranoia and euphoria, which can be partly attributed to their similarity to amphetamine compounds [2,4]; however, cathinones have an additional ketone group at the β position of the amino chain. Moreover, when cathinones are combined with other drugs such as gamma-hydroxybutyric acid (GHB), gamma-butyrolactone (GBL), or cocaine they can reduce inhibition and enhance sexual activities [2,4]. Synthetic cathinones are sold as "bath salts" and can be found



easily on different websites because drug agencies cannot control all of them due to their continuous structural changes [4,5].

When these substances are consumed, they can be present in different biological samples and they have become very important in toxicological and forensic analyses [6–20]. Although in these samples it is possible to find the parent drug and its metabolites, it is usually the latter that remain longer in the body. However, there are some matrices including blood and oral fluid (OF), in which the parent drug is not as metabolized as in others, such as in urine [21]. Moreover, depending on the biological sample, the detection window can be different and even this could be affected by several factors, such as the frequency and the amount of drug taken by the individual. In general, for short detection windows, breath, blood, or OF are the usual samples, for medium detection windows, urine and sweat are the most convenient, and for long detection windows, hair and meconium are the preferred choices [21–24]. In this work, we focused on urine and OF.

Although the concentration range of cathinones in urine and OF may depend on the amount of the drug consumed, in urine, low levels of these compounds in the range of ng mL⁻¹ 24 h after the consumption are usual [25], while in OF, cathinones can be found between low and high levels of ng mL⁻¹ due to its earlier detection [26]. Therefore, it is important to provide methods with high sensitivity to be able to detect the usual levels at which these substances can be found in biological samples from drug abusers. The instrumentation is, thus, a key factor for achieving the goal of low detection limits. Mass spectrometry (MS) is a promising tool for this purpose. In the literature, there are several examples in which cathinones have been determined mainly using liquid chromatography (LC) due to the polar characteristics of these compounds. The trend is to use MS as a detection system to obtain methods with high sensitivity. Low-resolution MS, such as tandem MS (MS/MS), and high-resolution MS (HRMS) have become the preferred choices in recent years [6-9,11-20]. In the case of MS/MS, there are several examples using triple quadrupole (QqQ) or ion trap MS (QTrap) [6,7,11–16]. For HRMS, Orbitrap, QOrbitrap, and quadrupole time-of-flight (QToF) are the most used detectors [8,9,17–20]. Despite the satisfactory results obtained in these previously reported strategies, neither HRMS nor MS/MS have been compared for determining synthetic cathinones in biological samples as other compounds and matrices have been compared in the literature [27–35]. It has been proven that, in general, HRMS offers some advantages compared to MS/MS such as the target, post-target, and nontarget analyses, the possibility of operating in full-scan mode, and higher specificity. Thus, HRMS allows analyzing data for an unlimited number of compounds, with the possibility of performing targeted and untargeted analysis and to perform retrospective analysis. On the other hand, MS/MS normally offers better sensitivity and wider linear ranges. It also offers short dwell times to obtain very narrow peaks [27,34,36,37]. In particular, there are several examples in which QqQ and Orbitrap have been compared [27,28,30,32,33], and the authors concluded that both detectors can reach a similar sensitivity level. However, there are some examples that find significant differences between the compared MS analyzers. This was the case of Vanhaecke et al. [31], who reported higher sensitivity for the QqQ for determining anabolic steroids in meat, Gómez-Canela et al. [34], who reported higher sensitivity for the Orbitrap for characterizing multiclass cytostatic compounds, Kaufmann et al. [29], who obtained better sensitivity for the QqQ for determining veterinary drugs in honey and bovine kidney samples, and Viaene et al. [35], who achieved better sensitivity for the QToF for determining opioids in plasma. As it can be observed, the type of compound and the matrix in which they are determined are key factors when the sensitivity is evaluated.

Considering all the mentioned studies, the main objective of the present work was to establish, for the first time, an accurate comparison between MS/MS and HRMS in terms of sensitivity, accuracy, and other method validation parameters when synthetic cathinones are determined in urine and OF samples.

2. Materials and Methods

2.1. Standards and Materials

Eleven synthetic cathinones were selected for this study: 2-methylmethcathinone (2-MMC), 3,4-dimethylmethylcathinone (3,4-DMMC), 4-fluoromethcathinone (flephedrone), 4-methylcathinone (4-MEC), 4-methylmethcathinone (mephedrone), alpha-pyrrolidinovalerophenone (alpha-PVP), beta-ethylmethcathinone (pentedrone), buphedrone, butylone, methylenedioxypyrovalerone (MDPV), and N-ethylcathinone. All of them were obtained from LGC Standards (Luckenwalde, Germany).

Different stock standard solutions for each cathinone were prepared in methanol (MeOH) at three concentrations: 100 mg L⁻¹, 1000 mg L⁻¹, and 2000 mg L⁻¹. A working standard mixture solution from the individual standard solutions was prepared in MeOH at 1 mg L⁻¹ and then diluted in water to prepare the final solutions. All the standard solutions were stored in the freezer at -20 °C.

The solvents and reagents used in this study were: Acetonitrile (ACN) for LC-MS from Chem-Lab (Zedelgem, Belgium); water for LC-MS purchased from J.T. Baker (Deventer, The Netherlands); hydrochloric acid (HCl), formic acid (HCOOH), ammonium hydroxide (NH₄OH), sodium dihydrogen phosphate (NaH₂PO₄), and disodium hydrogen phosphate (Na₂HPO₄) from Sigma-Aldrich (St. Louis, MO, USA); and ultrapure water obtained using a water purification system (Merck Millipore, Darmstadt, Germany).

For urine, Oasis MCX (150 mg/6 cc) cartridges from Waters Corp. (Milford, MA, USA) were used. For OF, Salivette[®] devices were purchased from Starstedt (Nümbrecht, Germany) and a Hettich Universal 32R centrifuge machine from Hettich (Tuttlingen, Germany) was used to centrifugate the samples.

2.2. HPLC Separation

The chromatographic conditions were optimized in a previous study in which the same group of cathinones were successfully separated [9].

A Luna Omega 5 μ m Polar C₁₈ (150 mm × 4.6 mm, 5 μ m) with a Security Guard from Phenomenex (Torrance, CA, USA) was used for the chromatographic separation at 35 °C with a mobile phase flow rate of 0.6 mL min⁻¹. The sample injection volume was 20 μ L and the vials were stored in the autosampler at 10 °C. The mobile phase was composed by A: 0.1% HCOOH in H₂O and B: 0.1% HCOOH in ACN in gradient mode starting at 15% B, which was maintained for 5 min, increased to 35% in 5 min, then to 80% in 4 min, and to 100% in 1 min. Finally, it was maintained for 2 min before returning to the initial conditions in 1 min and held for 4 min.

2.3. Orbitrap

A Thermo Scientific Accela 1250 UHPLC system (Bremen, Germany) equipped with an Accela Autosampler automatic injector and an Accela 1250 pump coupled with a Thermo Scientific Exactive OrbitrapTM mass spectrometer were used. The instrument was also equipped with a higher-energy collisional dissociation (HCD) cell and a heated electrospray ionization (HESI) source.

The gas flow rate and temperature parameters were optimized with positive ionization and the following parameters were used: Auxiliary gas, 5 adimensional units (AU); sheath gas, 60 AU; capillary voltage, 30 volts (V); tube lens voltage, 80 V; spray voltage, 2 kV; and skimmer voltage, 24 V. The capillary and heater temperatures were set at 350 °C and 400 °C, respectively. The probe position was set at micrometer 1, side to side 0, and vertical B. Two-time windows were used for data acquisition in positive mode (0–11.75 min and 11.75–15 min). Two scan events in each window were set: Full scan at 50,000 Full Width at Half Maximum (FWHM) with 250 ms of injection time for the first one, and fragmentation scan at 10,000 FWHM with 50 ms of injection time for the second one. The HCD cell was set in "all ion fragmentation" mode with an injection time of 50 ms. The optimum collision voltages were 15 electronvolts (eV) for the first one and 25 eV for the second one. The mass ranges used for the full scan at high resolution were between 60 mass-to-charge ratio (m/z) and 300 m/z.

Table 1 specifies the LC-HRMS protonated molecules and fragments. The ion ratio was calculated by dividing the signal of the protonated molecule by the one of the fragment.

Table 1. Retention time, protonated molecule (quantifier), the two fragment ions (qualifier), and the ratios of the studied cathinones by liquid chromatography high-resolution mass spectrometry (LC-HRMS).

		Protonated [M+	Molecule H] ⁺		Fragment		
Analyte	Rt (min)	Formula	Accurate Mass <i>m/z</i>	Formula	Accurate Mass <i>m/z</i>	Ratio	
Flephedrone	6 17	CtoHtoFNO	182 00757	$C_{10}H_{11}FN$	164.08755	38	
riepneurone	0.17	C1011131110	102.077.57	C ₉ H ₈ FN	149.06408	10	
Ethcathinone	6.29	C11H16NO	H]+ Accurate Mass m/z Formula 182.09757 $C_{10}H_{11}FN$ 178.12264 $C_{11}H_{14}N$ 178.12264 $C_{11}H_{14}N$ 178.12264 $C_{11}H_{14}N$ 178.12264 $C_{11}H_{14}N$ 222.11247 $C_{11}H_{12}NO$ 178.12264 $C_{11}H_{12}NO$ 178.12264 $C_{11}H_{14}N$ 178.12264 $C_{11}H_{14}NO_2$ 178.12264 $C_{11}H_{14}NO_2$ 178.12264 $C_{11}H_{14}NO_2$ 178.12264 $C_{11}H_{14}NO_2$ 178.12264 $C_{11}H_{14}NO_2$ 178.12264 $C_{12}H_{16}NO_2$ 178.12264 $C_{12}H_{16}NO_2$ 178.12264 $C_{12}H_{16}NO_2$ 178.12264 $C_{12}H_{16}NO_2$ 192.13829 $C_{12}H_{16}NO_2$ 192.13829 $C_{12}H_{16}NO_2$ 192.13829 $C_{12}H_{16}NO_2$ 232.16959 $C_{7}H_7$ $C_{7}H_5O_2$ $C_{7}H_5O_2$	$C_{11}H_{14}N$	160.11262	23	
Euleumone	0.29	0111110110	170.12204	$C_9H_{10}N$	132.08132	12	
Buphedrone	7.66	C11H16NO	178 12264	$C_{11}H_{14}N$	160.11262	19	
Dupileurone	7.00	011116100	Mass m/z Formation 182.09757 $C_{10}F$ 178.12264 $C_{9}F$ 178.12264 $C_{11}I$ 178.12264 $C_{10}I$ 178.12264 $C_{10}I$ 222.11247 $C_{11}F$ 178.12264 $C_{10}I$ 178.12264 $C_{10}I$ 222.11247 $C_{11}F$ 178.12264 $C_{10}I$ 192.13829 $C_{12}I$ 192.13829 $C_{12}I$ 192.13829 $C_{12}I$ 192.13829 $C_{12}I$ 192.13829 $C_{12}I$	C ₉ H ₉ N	131.07350	6	
2-MMC	8 97	C11H1/NO	178 12264	$C_{11}H_{14}N$	30		
2-1011010	0.97	011116100	1 ₁₆ NO 178.12264	$C_{10}H_{11}N$	145.08915	16	
Butvlone	0.05	CraHr NOa	222 11247	C ₁₁ H ₁₂ NO	1 ₁₂ NO 174.09189		
Butylone	9.05	C1211161 (C3	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	204.10245	17		
Menhedrone	0.38	C11H1/NO	178 12264	$C_{11}H_{14}N$	160.11262	34	
mephearone	9.50	CIIIII6IVO	Molecule H]+ Accurate Mass m/z Form 182.09757 $C_{10}H_1$ 182.09757 $C_{9}H_8$ 178.12264 $C_{11}H_1$ 178.12264 $C_{11}H_1$ 178.12264 $C_{11}H_1$ 178.12264 $C_{11}H_1$ 222.11247 $C_{11}H_1$ 21247 $C_{12}H_1$ 178.12264 $C_{10}H_1$ 192.13829 $C_{12}H_1$ 192.13829 $C_{12}H_1$ 192.13829 $C_{12}H_1$ 232.16959 C_7H_1 276.15942 C_8H_1	$C_{10}H_{11}N$	145.08915	14	
4 MEC	10 59	CraHaNO	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$C_{12}H_{16}N$	174.12827	30	
4-MEC	10.36	C121118100	192.13629	$C_{10}H_{11}N$	145.08915	9	
Pontodrono	10.08	CraHaNO	102 12820	$C_{12}H_{16}N$	174.12827	16	
remedione	10.90	0121118110	192.13029	$C_9H_{10}N$	132.08132	10	
3.4 DMMC	11.00	C ₁₂ H ₁₈ NO	102 12820	$102 12820$ $C_{12}H_{16}N$ 174.		37	
J,4-DivitviC	11.00		192.13829	$C_{11}H_{13}N$	159.10480	12	
Alpha-PVP	10.01		232.16959 C ₇ H C ₇ H ₅	C_7H_7	91.05478	15	
Alpha-1 VI	12.21	C1511221NO		C7H5O	105.03404	5	
MDDV	10 59	C. H. NO.	076 15040	$C_8H_{16}N$	126.12827	8	
IVIDP V	12.58	$C_{16} I_{22} I O_3$	276.13942	$C_8H_7O_2$	135.04461	9	

2.4. QqQ

The method was validated using an Agilent model 1200 series LC coupled with an Agilent 6460 series triple quadrupole mass spectrometer equipped with an electrospray ionization (ESI) interface from Agilent Technologies (Waldbronn, Germany).

The LC-MS/MS parameters were optimized by injecting individual standards at 1 mg L⁻¹ into a mixture of H₂O:MeOH (50:50, volume/volume (v/v)). Five different windows were created according to their retention times: Flephedrone and ethcathinone; buphedrone; 2-MMC, butylone, and mephedrone; 4-MEC and pentedrone; and 3,4-DMMC, alpha-PVP, and MDPV. The optimized source parameters were: Gas flow rate, 13 L min⁻¹; gas temperature, 350 °C; nebulizer pressure, 60 psi; and capillary voltage 2500 V. The fragmentor was 100 V for most of the compounds and 125 V for 3,4-DMMC, alpha-PVP, and MDPV. The collision energies were between 8 and 25 eV. A multiple reaction monitoring (MRM) mode was used, selecting the two most intensive transitions between the parent ion and the product ions. The optimized fragments are shown in Table 2. For the QqQ, the ion ratio was calculated by dividing the qualifier signal between the quantifier one.

Analyte	Rt (min)	Fragmentor (V)	Precursor Ion	Product Ion	CE (eV)	Ion Ratio
				164	15	
Flephedrone	6.53	100	182.1	149	15	70
	1 0.00 200			123	15	21
				160	12	
Ethcathinone	6.62	100	178.1	132	14	93
				105	18	39
				160	10	
Buphedrone	8.29	100	178.1	131	15	55
				132	15	52
				160	8	
2-MMC	9.98	100	178.1	145	15	71
				119	15	11
				174	14	
Butylone	10.26	100	222.1	204	10	101
				191	10	23
				<u>160</u>	8	
Mephedrone	10.48	100	178.1	145	15	61
				119	15	12
				<u>174</u>	10	
4-MEC	11.81	100	192.1	146	16	21
				145	18	24
				174	10	
Pentedrone	12.25	100	192.1	132	16	49
				91	18	24
				174	10	
3,4-DMMC	13.02	125	192.1	159	15	57
				133	15	7
				<u>91</u>	22	
Alpha-PVP	13.50	125	232.2	105	25	42
				126	25	41
				126	25	
MDPV	13.96	125	276.2	135	25	67
				175	20	52

Table 2. Retention time, fragmentor, precursor, product ions (the quantifier ion is underlined), ion ratios, and collision energies of the studied cathinones by LC-MS/MS.

2.5. Urine Collection and Preparation

To pretreat the urine samples, the same sample pretreatment procedure that was used in a previous study by our research group was applied [9]. Different urine samples from nonaddicted women and men volunteers of different ages were collected and mixed to obtain a pooled urine sample. All the volunteers consented to the study following the *Declaration of Helsinki* normative. The samples were collected in polypropylene tubes and kept in the freezer at -20 °C. Before their extraction, the samples were diluted with a phosphate buffer at pH 6 (0.15 molar (M) NaH₂PO₄ and 0.05 M Na₂HPO₄), 50:50 (*v*/*v*). The Oasis MCX cartridge was activated with 5 mL of MeOH, then conditioned with 5 mL of phosphate buffer solution (pH 6), and loaded with 5 mL of the mixture of urine:buffer 50:50 (*v*/*v*). After the loading, the cartridge was washed with 2 mL of MeOH and the analytes were eluted with 2 mL of 5% NH₄OH in MeOH. Finally, 100 µL of 1% HCl in MeOH were added prior to the evaporation to dryness under a gentle stream of N₂. Then, the obtained residue was reconstituted with 1 mL of mobile phase, filtered through a 0.45 µm polytetrafluoroethylene (PTFE) syringe filter, and transferred to a vial for its chromatographic analysis.

2.6. Saliva Collection and Preparation

The saliva was pretreated using the same procedure as in a previous study by our research group [11]. Salivette[®] was used to collect the OF samples from nonaddicted volunteers of different

ages following the *Declaration of Helsinki* normative. The cotton swab of the Salivette[®] was chewed and rolled around in the mouth for 1 min. This swab was returned to the tube and it was centrifuged for 1 min at 8000 rpm ($10,800 \times g$). The saliva fell to the bottom of the device and was discarded. Then 1 mL of MeOH was added to the swab and was centrifuged again at the same conditions. The centrifuge was collected, this procedure was repeated, and the two methanolic solutions were combined. Then, 100 µL of 1% HCl in MeOH were added to the solution before evaporating it to dryness under a gentle stream of nitrogen. The residue was then reconstituted with 500 µL of mobile phase under initial chromatographic conditions, filtered through a 0.45-µm PTFE filter, and the final solution was injected into the chromatograph.

2.7. Validation

The methods were validated following the guidance for the validation of analytical methodology and calibration equipment used for testing illicit drugs in seized materials and biological specimens and the European guidelines for workplace drug testing in oral fluid [38,39]. The developed methods were compared in terms of apparent recoveries (R_{app}), matrix effects (ME), method detection and quantification limits (MDLs and MQLs), linearity, and accuracy. The retention time in the analyte confirmation had to correspond to a tolerance of ±2.5% and the ion ratio to ±20%. Moreover, in the case of Orbitrap, the error mass had to correspond to a maximum tolerance of 5 ppm [40].

To evaluate the mentioned parameters, urine was spiked before the extraction procedure and OF was spiked in the cotton swab after it had been chewed. The R_{app} in both cases was the recovery of all the methods, calculated from the ratio of the cathinone concentration of a sample spiked prior to the procedure and the concentration for the same cathinone in a standard sample. Both of these were obtained from the instrumental calibration curve.

The percentage of ME was calculated according to the following Equation (1):

$$%ME = [(C_{spiked}/C_{STD}) \times 100] - 100$$
 (1)

 C_{STD} is the concentration of a cathinone in a standard sample and C_{spiked} is the concentration of the same cathinone in urine or in an OF extract spiked at the same concentration as C_{STD} after the procedure.

In the case of Orbitrap, the MDLs were defined as the lowest detectable concentrations with a signal intensity higher than $\times 10^3$ for the protonated molecule and for the fragments, while the MQLs were considered the lowest point in the calibration curve. However, for the QqQ, the MDLs were the minimum concentrations with a signal to noise (S/N) ratio greater than 3 and the MQLs were the lowest point in the calibration curve with a S/N ratio higher than 10.

The method linearities were based on a matrix-matched calibration curve by evaluating the determination coefficient (r^2). Two concentration levels were used to study the R_{app} and ME: 1 ng g⁻¹ (low) and 40 ng g⁻¹ (high) for OF and 1 ng mL⁻¹ (low) and 40 ng mL⁻¹ (high) for urine. In the case of OF, the concentrations were expressed as nanograms (ng) of cathinone per gram of OF weighted (ng g⁻¹), and ng was used for the amount of the cathinone in OF in the calibration curve. In the pretreatment, the Salivette[®] was weighed before and after the sampling to determine the amount of saliva sampled. Thus, to express the concentration of a cathinone in a sample, the ng found was divided by the grams of saliva weighed.

Finally, to determine the accuracy of the methods, we analyzed different blind samples spiked with all the cathinones by a laboratory staff member at different concentrations. The concentration found in these samples was calculated, compared to the spiked value, and the accuracy in terms of error was calculated for each compound.

3. Results and Discussion

In order to evaluate both LC-MS/MS and LC-HRMS methodologies, the methods for urine and OF samples for the two detectors were validated to compare their performance in terms of linearity, MDLs, MQLs, R_{app}, ME, and accuracy of the methods.

3.1. Urine Samples

3.1.1. Method Recoveries and Matrix Effects

As it was mentioned above, $\[Mapp]$ was evaluated at low and high-concentration levels (1 ng mL⁻¹ and 40 ng mL⁻¹). For the low concentration tested, the values of $\[Mapp]$ for the different cathinones ranged from 73 to 125% for Orbitrap and from 53 to 98% for QqQ, while for the higher concentration value the recovery values ranged between 69 and 88% and 72 and 87%, respectively (Figure 1).



Figure 1. The percentage of apparent recovery ($\[mathcar{R}_{app}\]$) at low (1 ng mL⁻¹) and high (40 ng mL⁻¹) levels of concentrations for Orbitrap and triple quadrupole (QqQ) in the analysis of synthetic cathinones in urine.

In the case of ME (Figure 2), for Orbitrap at 1 ng mL⁻¹, suppression values were between -8 and -28% for most of the compounds. Enhancements of 12 and 21% were obtained for 4-MEC and alpha-PVP, respectively. At 40 ng mL⁻¹, suppressions between -9 and -22% were obtained for all the compounds. However, for QqQ at the low-concentration level (1 ng mL⁻¹), values from -14 to -33% were obtained for most of the compounds except for flephedrone and 3,4-DMMC, which had enhancements of 12 and 10%, respectively. At the high level (40 ng mL⁻¹), ME values ranged between -5 and -20% for QqQ.

In order to statistically compare both instruments in terms of R_{app} , a two-way ANOVA was performed. At the low-concentration level, LC-HRMS achieved slightly better results compared to LC-QqQ. The results of the statistical test proved that there were significant differences between both MS systems (critical Fisher (F)-value: 4.96; F-obtained: 7.85). On the contrary, at the high concentration, no significant differences between both instruments were found (critical F-value: 4.96; F-obtained: 2.88).





Figure 2. The percentage of matrix effect (%ME) at low (1 ng mL⁻¹) and high (40 ng mL⁻¹) levels of concentrations for Orbitrap and QqQ in the analysis of synthetic cathinones in urine.

3.1.2. Linearity, MDLs, and MQLs

Matrix-matched calibration curves were used for quantification in the two methods to compensate the ME and the possible losses in the sample pretreatment. Table 3 shows the linearity range and the MDLs and MQLs obtained when the two MS analyzers were used to analyze cathinones in urine. In the case of Orbitrap, the linear range was from 0.050 to 4 ng mL⁻¹ and from 4 to 200 ng mL⁻¹ for most of the cathinones and from 0.200 to 4 ng mL⁻¹ and from 4 to 200 ng mL⁻¹ for ethcathinone, buphedrone, and 4-MEC. For QqQ, the linear range was between the MQL and 4 ng mL⁻¹ and from 4 to 200 ng mL⁻¹. The linear ranges for the two analyzers encompassed the usual concentrations at which these compounds can be found in urine samples from drug abusers. The MQLs for QqQ ranged from 0.020 to 0.070 ng mL⁻¹ and for Orbitrap they were 0.050 ng mL⁻¹ for most analytes and 0.200 ng mL⁻¹ for ethcathinone, buphedrone, and 4-MEC. The MDLs were between 0.005 and 0.035 ng mL⁻¹ for Orbitrap and between 0.040 and 0.160 ng mL⁻¹ for QqQ.

Table 3. Linearity, method detection limits (MDLs) and method quantification limits (MQLs) for Orbitrap and QqQ in ng mL⁻¹ for the determination of synthetic cathinones in urine.

	MDL Orbitrap	MDL QqQ	MQL Orbitrap	MQL QqQ	Linear Range Orbitrap	Linear Range QqQ
Flephedrone	0.080	0.025	0.200	0.050	0.200-4···4-200	$0.050 - 4 \cdots 4 - 200$
Ethcathinone	0.100	0.030	0.050	0.050	$0.050 - 4 \cdot \cdot \cdot 4 - 200$	$0.050 - 4 \cdot \cdot \cdot 4 - 200$
Buphedrone	0.100	0.035	0.200	0.050	$0.200 - 4 \cdots 4 - 200$	$0.050 - 4 \cdots 4 - 200$
2-MMC	0.100	0.025	0.050	0.050	$0.050 - 4 \cdots 4 - 200$	$0.050 - 4 \cdots 4 - 200$
Butylone	0.120	0.030	0.050	0.050	$0.050 - 4 \cdots 4 - 200$	$0.050 - 4 \cdot \cdot \cdot 4 - 200$
Mephedrone	0.080	0.005	0.050	0.030	$0.050 - 4 \cdots 4 - 200$	$0.030 - 4 \cdot \cdot \cdot 4 - 200$
4-MEC	0.040	0.010	0.200	0.030	$0.200 - 4 \cdots 4 - 200$	$0.030 - 4 \cdots 4 - 200$
Pentedrone	0.040	0.025	0.050	0.030	$0.050 - 4 \cdots 4 - 200$	$0.030 - 4 \cdot \cdot \cdot 4 - 200$
3,4-DMMC	0.040	0.010	0.050	0.030	$0.050 - 4 \cdots 4 - 200$	$0.030 - 4 \cdot \cdot \cdot 4 - 200$
Alpha-PVP	0.160	0.005	0.050	0.020	$0.050 - 4 \cdots 4 - 200$	$0.020 - 4 \cdot \cdot \cdot 4 - 200$
MDPV	0.060	0.005	0.050	0.030	$0.050 - 4 \cdots 4 - 200$	$0.030 - 4 \cdot \cdot \cdot 4 - 200$

Comparing all these values, it was possible to conclude that with QqQ lower MDLs and MQLs were obtained and, therefore, the method based on LC-MS/MS has better sensitivity for the analyzed cathinones in urine samples than the one based on HRMS. In different studies reported in the literature focused on the comparison of different MS analyzers, the general conclusion was that no significant differences between LC-HRMS and LC-MS/MS in terms of sensitivity were observed [27–35]. However,

there are some examples in which the results are coincident with those obtained in the present study, as in the case of Vanhaecke et al. [31] and Kaufmann et al. [29], who also obtained better sensitivity for QqQ, although it is important to point out that these authors analyzed different compounds and the matrices were also different to the ones analyzed in this study.

3.1.3. Accuracy

To study the accuracy of the method, three different blind urine samples (A, B, and C) spiked by a laboratory staff member were analyzed using the two methods, LC-Orbitrap and LC-MS/MS. The three samples were spiked at different concentrations: 2 ng mL⁻¹ (A), 60 ng mL⁻¹ (B), and 125 ng mL⁻¹ (C). The accuracy was calculated in terms of percentage of error between the concentration obtained for each cathinone by analyzing these blind samples and the spiked concentration for each analyte. Finally, the average percentage of error was calculated for each sample. In the case of Orbitrap, values of 18% (A), 6% (B), and 15% (C) were obtained and for QqQ the values were 16% (A), 1% (B), and 15% (C). According to the validation guide, the results have to fall within ±20% at the lower concentration and ±15% at the higher concentration [36]. From these results it was possible to observe that all the values obtained were within the range mentioned and, therefore, they were considered very good results for the method accuracy. Moreover, no significant differences between LC-HRMS and LC-MS/MS were observed in terms of accuracy.

3.2. OF Samples

3.2.1. Method Recoveries and Matrix Effects

The $\[Mapp]\]$ and ME for the OF samples were also evaluated, as for the urine samples, at the same two concentration levels: 1 ng g⁻¹ and 40 ng g⁻¹. For both MS analyzers higher $\[Mapp]\]$ was obtained at the high-concentration level evaluated and higher ME were obtained at the low-concentration level. R_{app} at the low level ranged from 41 to 53% for Orbitrap and 50 to 60% for QqQ, while at high levels it ranged from 50 to 55% and from 61 to 66%, respectively (Figure 3). The ME values observed at the low level were between -7 and -20% for Orbitrap and between -14 and -21% for QqQ. ME at the high level ranged from -1 to -9% for Orbitrap and from -5 to -10% for QqQ (Figure 4).



Figure 3. The $%R_{app}$ at low (1 ng g⁻¹) and high (40 ng g⁻¹) levels of concentrations for Orbitrap and QqQ in the analysis of synthetic cathinones in oral fluid (OF).



Figure 4. The %ME at low (1 ng g^{-1}) and high (40 ng g^{-1}) levels of concentrations for Orbitrap and QqQ in the analysis of synthetic cathinones in OF.

In the same case as urine, a two-way ANOVA was used to compare both detectors in OF. At both levels of concentration, significant differences could be observed, obtaining better results for the LC-QqQ (critical F-value: 4.96; F-obtained: 5.98 at low concentration and critical F-value: 4.96; F-obtained: 265.38 at high concentration).

3.2.2. Linearity, MDLs, and MQLs

Similarly to the urine samples, a matrix-matched calibration curve for the OF samples was used. Table 4 shows the linearity range and MDL and MQL parameters studied for analyzing cathinones in OF. Similar linear ranges were obtained for Orbitrap and QqQ. For QqQ, the calibration curves were adjusted to two linear ranges, from 0.075 to 4 ng and from 4 to 75 ng of cathinone in OF for all the compounds. The Orbitrap calibration curves were also adjusted to the two same linear ranges for most of the compounds except for buphedrone and 4-MEC, which were from 0.100 to 4 ng and from 4 to 75 ng of cathinone in OF. These linear ranges covered all the usual concentrations at which cathinones can be found in OF from drug abusers.

Table 4. Linearity in ng of cathinones, MDLs, and MQLs in ng g^{-1} for Orbitrap and QqQ for the determination of synthetic cathinones in OF.

	MDL Orbitrap	MDL QqQ	MQL Orbitrap	MQL QqQ	Linear Range Orbitrap	Linear Range QqQ
Flephedrone	0.020	0.010	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Ethcathinone	0.035	0.030	0.075	0.075	$0.075 - 4 \cdot \cdot \cdot 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Buphedrone	0.025	0.030	0.100	0.075	$0.100 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
2-MMC	0.015	0.015	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Butylone	0.020	0.010	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Mephedrone	0.010	0.010	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
4-MEC	0.025	0.020	0.100	0.075	$0.100 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Pentedrone	0.020	0.020	0.075	0.075	$0.075 - 4 \cdot \cdot \cdot 4 - 75$	$0.075 - 4 \cdots 4 - 75$
3,4-DMMC	0.015	0.010	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$
Alpha-PVP	0.010	0.005	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdots 4 - 75$
MDPV	0.015	0.003	0.075	0.075	$0.075 - 4 \cdots 4 - 75$	$0.075 - 4 \cdot \cdot \cdot 4 - 75$

Thus, the MQL, which was considered as the lowest point in the calibration curve, was 0.075 ng g^{-1} for all the compounds in the case of QqQ. For Orbitrap, they were 0.075 ng g^{-1} for most of the compounds except for buphedrone and 4-MEC, for which they were 0.100 ng g^{-1} . The MDLs were slightly lower in the case of QqQ, between 0.003 and 0.030 ng g⁻¹, and they were between 0.010 and 0.035 ng g⁻¹ in the case of Orbitrap.

Comparing both the MDLs and MQLs for the two analyzers, lower values were obtained, in general, when LC-MS/MS was used. Therefore, better sensitivity was obtained with this analyzer, despite only slight differences.

3.2.3. Accuracy

To study the accuracy of the method for OF samples, three different blind samples spiked by a member of the laboratory at different concentrations (0.45 ng g⁻¹ (A), 6.2 ng g⁻¹ (B), and 58 ng g⁻¹ (C)) were analyzed with the two methods. The % errors obtained for Orbitrap were 16% (A), 10% (B), and 15% (C), while for QqQ the % errors were 18% (A), 9% (B), and 6% (C). These values were within the acceptable limits according to the guide and are considered excellent results in both cases [36]. The results show that there were no observable differences and in both cases good results were achieved.

4. Conclusions

Two analytical methods for determining synthetic cathinones in urine and OF by LC-HRMS and LC-MS/MS were compared in terms of linearity, MDLs, MQLS, %R_{app}, %ME, and accuracy. For both biological samples, better sensitivity was observed in the case of LC-MS/MS and no significant differences were found when %R_{app} and %ME were compared. Finally, the two methodologies demonstrated good accuracy values and, although similar values were obtained for urine and OF samples, the latter showed slightly better results for LC-MS/MS. For these reasons, in general terms, we can say that LC-MS/MS can provide better sensitivity and accuracy when synthetic cathinones are determined in urine and OF. Nevertheless, it was also observed that the two methodologies (based on either HRMS or MS/MS) both have good potential for forensic and toxicological analyses. With these methods, low concentrations of these substances can be found in urine and OF samples, which is interesting from the juridical point of view. The developed research can be considered an important starting point to apply the different methods for urine and OF samples in monitoring campaigns, which can be focused in the study of the trends of the cathinone consumption in different regions, and to establish possible relationships between the consumption and the age of the consumers.

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Abbreviations

OF, oral fluid; MS, mass spectrometry; MS/MS, tandem mass spectrometry; HRMS, high-resolution mass spectrometry; QqQ, triple quadrupole; R_{app}, apparent recovery; ME, matrix effect; MDL, method detection limits; MQL, method quantification limits.

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