

# Practical Synthesis of Chiral Ferrocenylphosphino-Gold(I) Catalysts and NEST Analysis of the Enantioinduction

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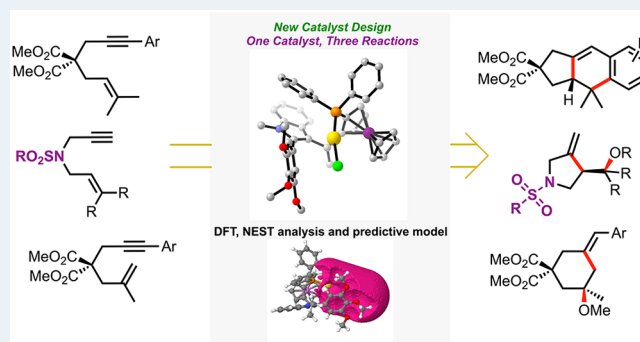
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**ABSTRACT:** The concise modular synthesis of a family of monodentate 1,2-disubstituted ferrocene ligands containing a diaryl phosphine and a 2-aryl indole is described. Their gold(I) complexes were applied to the enantioselective gold(I)-catalyzed formal [4 + 2] cycloaddition of 1,6-arylenynes, the enyne cyclization/nucleophile addition of N-tethered 1,6-enynes, and the methoxycyclization of 1,6-arylenynes with high levels of enantioselectivity in all cases. Crystallographic and computational studies highlighted the relevant role of noncovalent interactions within the ligand scaffold and between the ligand and substrate in the modes of enantioinduction in the cyclization of unsaturated substrates. Our recently developed open-source tool NEST was applied to analyze the chiral pockets of the catalysts, which in combination with RDKit allowed us to understand the enantioselectivity in these reactions, paving the way for a predictive-based approach toward the rational development of chiral ligands for enantioselective Au(I) catalysis.

**KEYWORDS:** gold catalysis, enynes, ferrocenyl ligand, cycloaddition, alkoxy cyclization



## INTRODUCTION

The design of new chiral ligands plays an important role in the progress of asymmetric catalysis.<sup>1</sup> Ferrocenyl phosphines are arguably one of the most relevant and versatile classes of ligands for asymmetric metal-catalyzed transformations.<sup>2</sup> Since the pioneering work of Hayashi,<sup>3</sup> many other ligands based on ferrocene, such as Josiphos,<sup>4</sup> TaniaPhos,<sup>5</sup> Walphos,<sup>6</sup> Bophos,<sup>7</sup> Trifler,<sup>8</sup> and Zhaophos,<sup>9</sup> have been applied in a broad range of asymmetric metal-catalyzed transformations. Interestingly, the first example of enantioselective gold(I) catalysis relied on the use of a bifunctional gold(I) ferrocenyl phosphine catalyst in the asymmetric aldol-type reaction between aldehydes and isocynoacetates.<sup>10</sup> Despite this pioneering work, the field of enantioselective gold(I) catalysis has been largely dominated by axially chiral binuclear gold(I) complexes<sup>11–16</sup> and monodentate phosphoramidite ligands.<sup>17–20</sup> Alternative strategies based on complexes with chiral counteranions<sup>21–25</sup> and/or hydrogen-bonding<sup>26–29</sup> have also been developed. Only recently, gold(I) ferrocene-based complexes have emerged as promising candidates for asymmetric gold(I) catalysis.<sup>30–33</sup>

Our group has reported the development of novel ligand designs tailored to meet the requirements imposed by the geometry of gold(I) (Figure 1).<sup>30,32,34–37</sup> Despite this considerable progress, ligand preparation is often lengthy<sup>32,38</sup> and, on occasion, chiral HPLC separation is required,<sup>39,40</sup> thus limiting expedient modifications of the catalyst structure.

Inspired by the work of Togni and Hu on heterocycle-containing ferrocenyl ligands,<sup>41–44</sup> we envisioned a new ligand scaffold based on a ferrocenyl phosphine containing an N-heterocycle. Here, we report the readily modular preparation of chiral gold(I) complexes with 1,2-disubstituted ferrocene ligands in only two steps from [1-(dimethylamino)ethyl]-2-(diphenylphosphino)ferrocene (PPFA), which is commercially available in either enantiomeric form, and their application in enantioselective gold(I) catalysis. This allowed us to test with this new catalyst design our recently developed open-source tool NEST,<sup>36</sup> based on DFT calculations and X-ray structural data, in combination with RDKit, to understand the enantioselectivity achieved in gold(I)-catalyzed cyclization reactions.

## RESULTS AND DISCUSSION

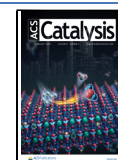
A new library of gold(I) complexes has been built using a modular synthetic sequence based on 2-aryl indoles **3a–o** and commercially available (*R,S*<sub>p</sub>)- or (*S,R*<sub>p</sub>)-PPFA (**1**), which can also be easily obtained from Ugi's amine<sup>45</sup> (Scheme 1).

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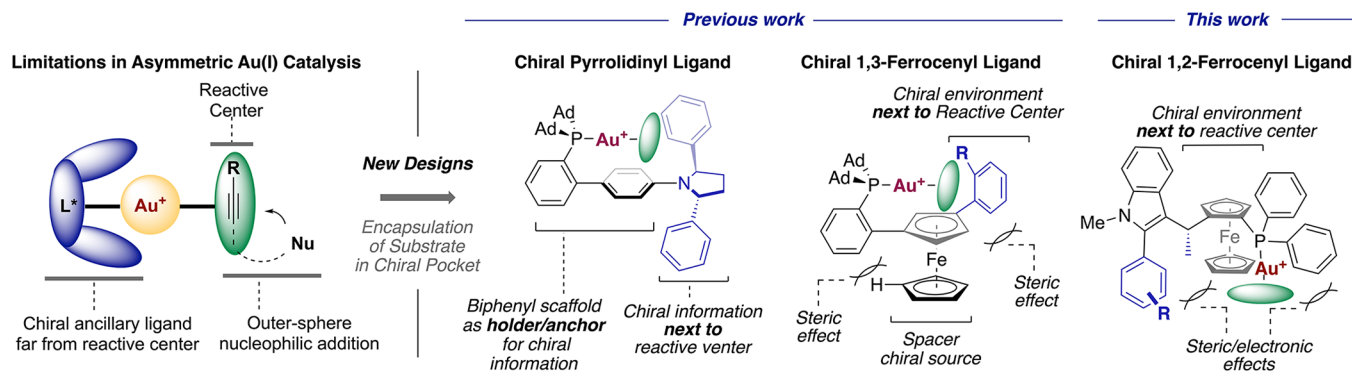
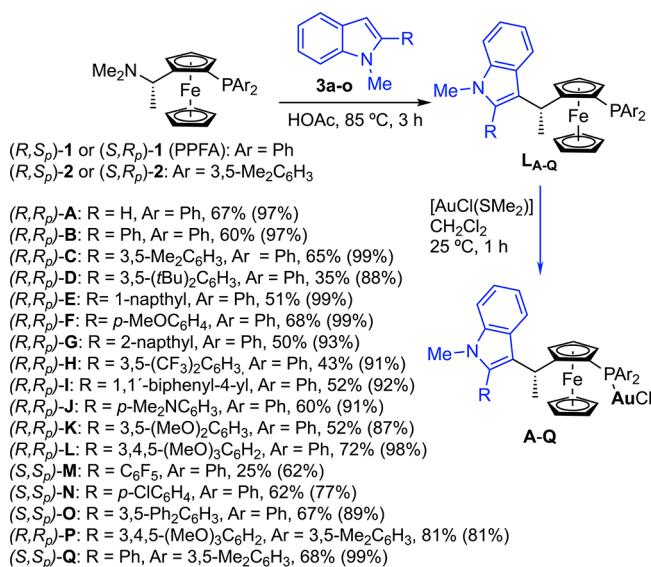


Figure 1. Design of chiral monodentate 1,2-ferrocene-based ligands for asymmetric gold(I) catalysis.

### Scheme 1. Synthesis of Gold(I) Complexes A–Q from Ferrocenes 1 or 2<sup>a</sup>



<sup>a</sup>Yields for the complexation with ([AuCl(SMe<sub>2</sub>)] in parentheses.

The reaction of PPFA with 2-substituted *N*-methylindoles afforded the chiral ligands L<sub>A–O</sub>, which upon complexation with [AuCl(SMe<sub>2</sub>)] led to complexes A–O in moderate to good yields, with the exception of D, H, and M in which the corresponding ligands bearing very bulky or electron-withdrawing R groups were obtained in 25–43% yields. Complexes P and Q were similarly prepared from 2 (Ar = 3,5-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>). The structures of complexes A–B, G–H, L, and P were confirmed by X-ray diffraction (Figure 2).

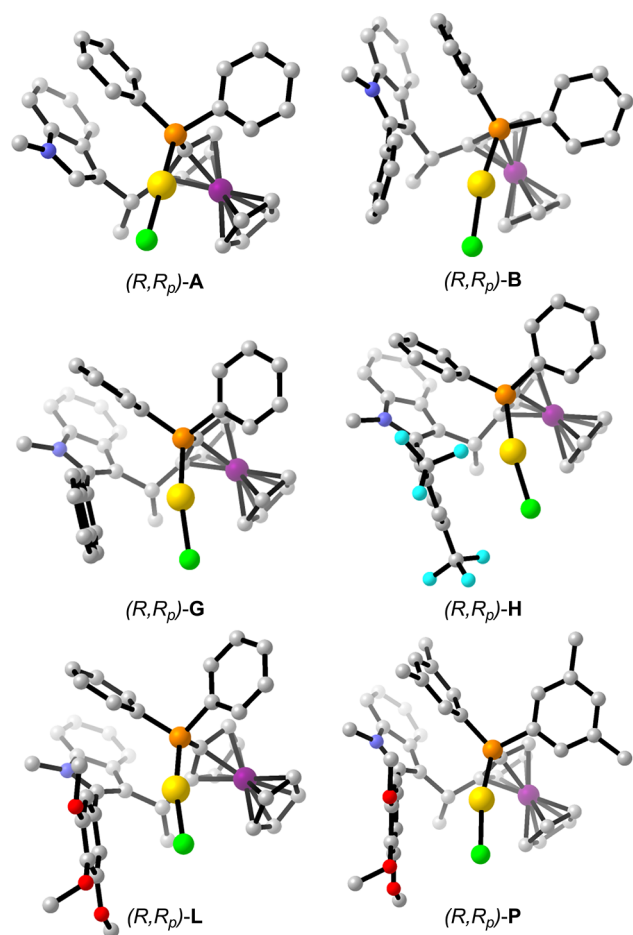
We first prepared catalyst (*R,R*<sub>p</sub>)-A derived from *N*-methylindole (3a) (Scheme 1) and examined its activity in the formal [4 + 2] intramolecular cycloaddition of 1,6-enyne 4a to yield cycloadduct 5a<sup>46,47</sup> (Table 1), a reaction that we have used to benchmark the performance of chiral gold(I) catalysts.<sup>24,26,32,34,36</sup> Although gold(I) complex A was active in the reaction, the cycloadduct was obtained in a nearly racemic form (Table 1, entry 1). Examination of the X-ray structure of (*R,R*<sub>p</sub>)-A suggested that the lack of enantioinduction could be ascribed to insufficient shielding of the metal center by the 2-methylindole group (Figure 2). Therefore, a phenyl group at the C-2 position of the indole was introduced, leading to a new gold(I) complex (*R,R*<sub>p</sub>)-B, whose performance resulted in a

significant increase in enantioselectivity from 52:48 to 82:18 *er* (Table 1, entries 1–2).

Analysis of the crystallographic structure of complex (*R,R*<sub>p</sub>)-B confirmed that the introduction of the pseudobiaryl fragment in the ligand blocks one side of the scaffold, creating a better defined chiral pocket (Figure 2). In the solid state, shorter distances were observed between the aryl ring at C-3 of the indole and the gold atom in complex L (3.413 Å) than in B (3.824 Å), G (3.714 Å), and H (3.657 Å). The X-ray crystal structure of the cationic complex (*R,R*<sub>p</sub>)-L<sub>1</sub>Au(MeCN)NTf<sub>2</sub>, prepared by the reaction of (*R,R*<sub>p</sub>)-L with AgNTf<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> in the presence of MeCN, revealed a reduction in the dihedral angle in the indole–aryl axis from 74.2° to 42.6° with respect to neutral (*R,R*<sub>p</sub>)-L. The distance centroid–Au in the cationic complex is 3.59 Å, much shorter than that in the neutral complex (4.16 Å). This crystallographic information suggests the presence of weak Au–arene interactions, which are also present in other gold(I) complexes.<sup>34,48–50</sup> The torsion angle in the indole–aryl axis was significantly larger for complex L (−74.2°) than for the rest of this family of complexes.<sup>51</sup>

The introduction of two methyl groups at the 3,5-position of the aryl at C-2 of the indole in complex C led to a modest improvement in the enantioselectivity (Table 1, entry 3). However, increasing the steric bulk with two *tert*-butyl groups in D proved to be detrimental (Table 1, entry 4). Results with E and F (Table 1, entries 5 and 6) were very similar to those obtained with B. The presence of a 2-naphthyl group in G or a 3,5-(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub> aryl in H led to 5a in moderate yields and enantioselectivities (Table 1, entries 7 and 8). Product 5a was obtained in poor yield and moderate enantioselectivity, with complex I bearing a biphenyl fragment at C-2 of the indole (Table 1, entry 9). On the other hand, whereas the presence of a *p*-Me<sub>2</sub>N in J resulted in nearly racemic 5a (Table 1, entry 10), complexes K and L with 3,5-dimethoxyphenyl or 3,4,5-trimethoxyphenyl rings, respectively, yielded 5a in good yields and enantioselectivities (Table 1, entries 11 and 12). Cationic complex (*R,R*<sub>p</sub>)-L<sub>1</sub>Au(MeCN)NTf<sub>2</sub> exhibited a performance similar to that of L.<sup>51</sup>

Catalyst M with a C<sub>6</sub>F<sub>5</sub> group at the C-2 of the indole led to lower enantioselectivity than B with a Ph group, whereas a *p*-ClC<sub>6</sub>H<sub>5</sub> in N gave the same enantioselectivity (Table 1, compare entry 2 with entries 13 and 14). Replacement of the *tert*-butyl groups in D by Ph groups in O led to a decrease in the yield of 5a, but with an improved enantiomeric excess (Table 1, entries 4 and 15). Finally, we found that changing the phenyl groups at the phosphine to 3,5-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub> groups in



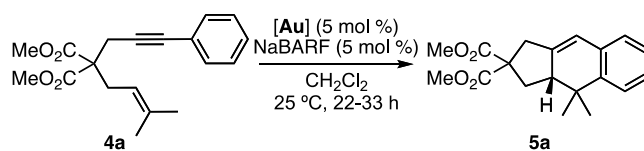
	A	B	G	H	L	P
$d(\text{Au}-\text{C}_3)$	4.152	3.824	3.714	3.657	3.413	3.434
$\wedge(1-2-3-4)$	-	-52.55	-55.26	-68.8	-74.2	-49.8

**Figure 2.** Selected X-ray crystal structures of complexes A, B, G, H, L, and P. Distances are expressed in Å and angles in degrees. Hydrogen atoms are omitted for clarity.

P and Q had a minor effect on the enantioselectivity (Table 1, compare entry 12 with entry 16 and entry 2 with entry 17).

The scope of the formal [4 + 2] cycloaddition of aryl 1,6-enynes **4a–m** was then explored using  $(R,R_p)$ -L (5 mol %) and  $\text{AgNTf}_2$  (5 mol %) in  $\text{CH}_2\text{Cl}_2$  at 25 °C (Scheme 2). Good to excellent enantioselectivities were obtained for **5a–e** (Scheme 2). Cyclization of **4f–g** bearing *o*-substituents at the aryl led to tricyclic adducts **5f–g** in good yields, although the enantioselectivity decreased slightly in the case of **5g**. Good to excellent yields and enantioselectivities were also obtained in the cyclization of substrates **4h–m** with different aryl or heteroaryl groups, alkene substitutions, or substituents at the 1,6-enyne tether.

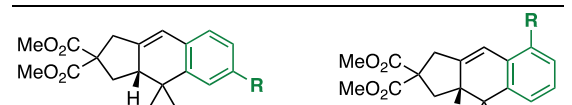
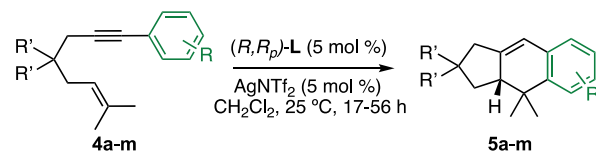
**Table 1.** [4 + 2] Cycloaddition of Enyne **4a** with Gold(I) Complexes A–Q



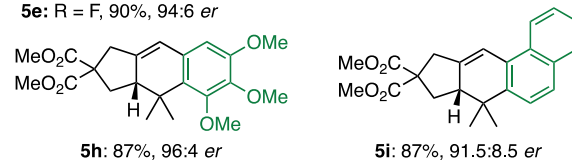
entry	Au catalyst	yield (%) <sup>a</sup>	er <sup>b</sup>
1	$(R,R_p)$ -A	53	52:48
2	$(R,R_p)$ -B	68	82:18
3	$(R,R_p)$ -C	63	85:15
4	$(R,R_p)$ -D	40	63:37
5	$(R,R_p)$ -E	63	86:14
6	$(R,R_p)$ -F	46	87:13
7	$(R,R_p)$ -G	53	79:21
8	$(R,R_p)$ -H	42	72:28
9	$(R,R_p)$ -I	13	78:22
10	$(R,R_p)$ -J	48	55:45
11	$(R,R_p)$ -K	79	91.5:8.5
12	$(R,R_p)$ -L	83	95.5:4.5
13	$(S,S_p)$ -M	74	37:63
14	$(S,S_p)$ -N	77	18:82
15	$(S,S_p)$ -O	22	13.5:86.5
16	$(R,R_p)$ -P	60	92:8
17	$(S,S_p)$ -Q	75	20:80

<sup>a</sup>Determined by <sup>1</sup>H NMR against internal standard. <sup>b</sup>Determined by SFC on the chiral stationary phase.

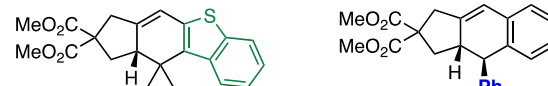
**Scheme 2.** Enantioselective [4 + 2] Cycloaddition of Aryl-Substituted 1,6-Enynes **4a–m** Catalyzed by  $(R,R_p)$ -L



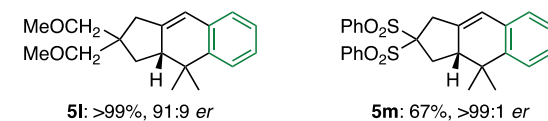
**5a:** R = H, 63%, >99:1 *er*  
**5b:** R = OMe, 66%, 96.5:4.5 *er*  
**5c:** R = NO<sub>2</sub>, 62%, >99:1 *er*  
**5d:** R = Me, 99%, 96:4 *er*  
**5e:** R = F, 90%, 94:6 *er*



**5h:** 87%, 96:4 *er*  
**5i:** 87%, 91.5:8.5 *er*



**5j:** 87%, 96:4 *er*  
**5k:** 74%, 95:5 *er*

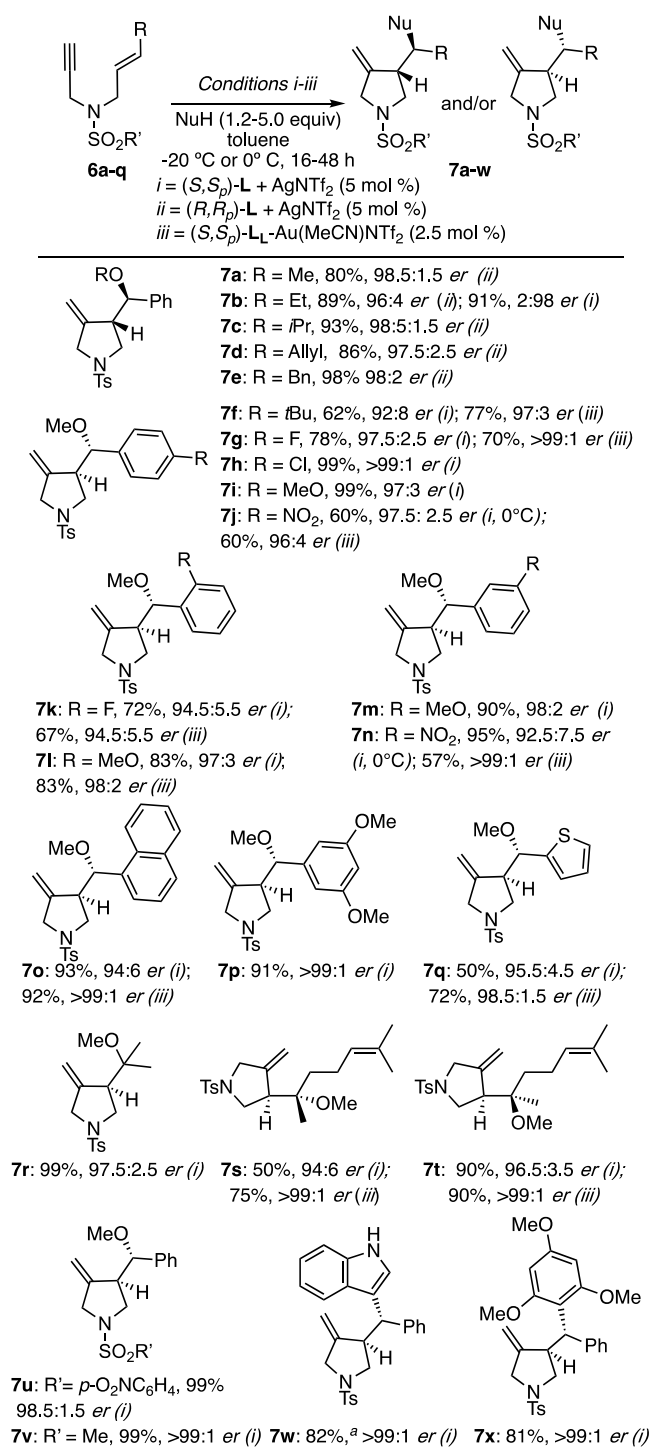


**5l:** >99%, 91:9 *er*  
**5m:** 67%, >99:1 *er*

Complexes  $(R,R_p)$ -L,  $(S,S_p)$ -L, or  $(S,S_p)$ -L<sub>1</sub>Au(MeCN)NTf<sub>2</sub> were also found to be optimal<sup>51</sup> for the cyclization/nucleophile addition of *N*-tethered 1,6-enynes **6**, providing chiral

pyrrolidines in good yields and high enantioselectivities (Scheme 3). Thus, alkoxy cyclization products **7a–v** were obtained in good yields and good to excellent enantioselectivities from 1,6-enynes **6a–q** and alcohols. As expected,<sup>26,52,53</sup> diastereomeric products **7s** and **7t** were obtained from (*E*)- and (*Z*)-configured starting materials, respectively. The reaction was also extended to electron-rich C-nucleophiles

**Scheme 3. Enantioselective Cyclization/Nucleophile Addition of *N*-Tethered 1,6-Enynes **6a–q****

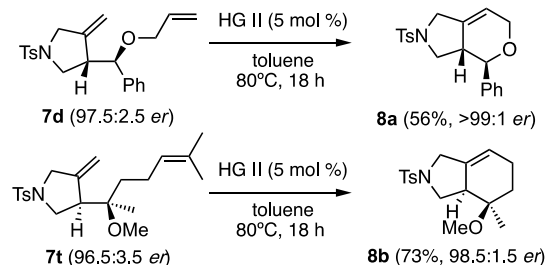


<sup>a</sup>3:1 ratio together with the product of indole attack at the carbene of the intermediate cyclopropyl gold(I) carbene.<sup>51</sup>

such as indole and 1,3,5-trimethoxybenzene to provide enantiopure compounds **7w** and **7x**. In general, slightly better enantiomeric excesses were obtained with the more reactive (*S,S\_p*)-L<sub>1</sub>Au(MeCN)NTf<sub>2</sub>, which also allowed us to lower the catalyst loading from 5 to 2.5 mol %.

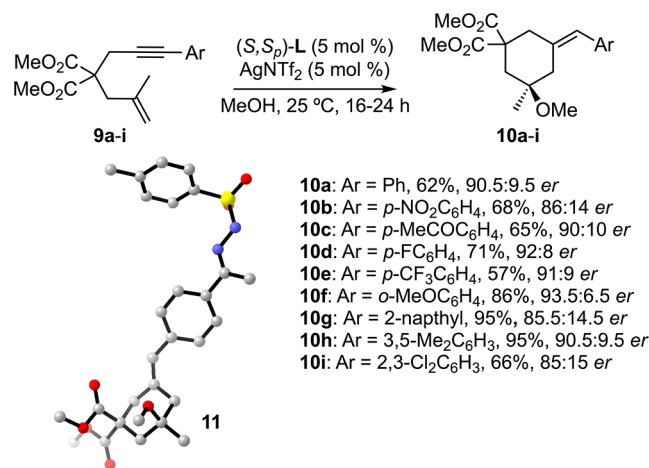
Alkoxy cyclization products **7d** and **7t** were readily converted to bicyclic products **8a** and **8b**, respectively, by ring-closing metathesis (Scheme 4).

**Scheme 4. Ring-Closing Metathesis of **7d** and **7t****



Similarly, the methoxycyclization of malonate-tethered aryl-substituted 1,6-aryl enynes **9a–i** yielded 1-methoxy-1-methyl-3-methylenecyclohexanes **10a–i** in good to excellent yields and enantioselectivities, which ranged from 85:15 to 93.5:6.5 *er* (Scheme 5). The absolute configuration of the resulting products was confirmed by X-ray diffraction of tosyl hydrazone **11**, obtained from **10c**.

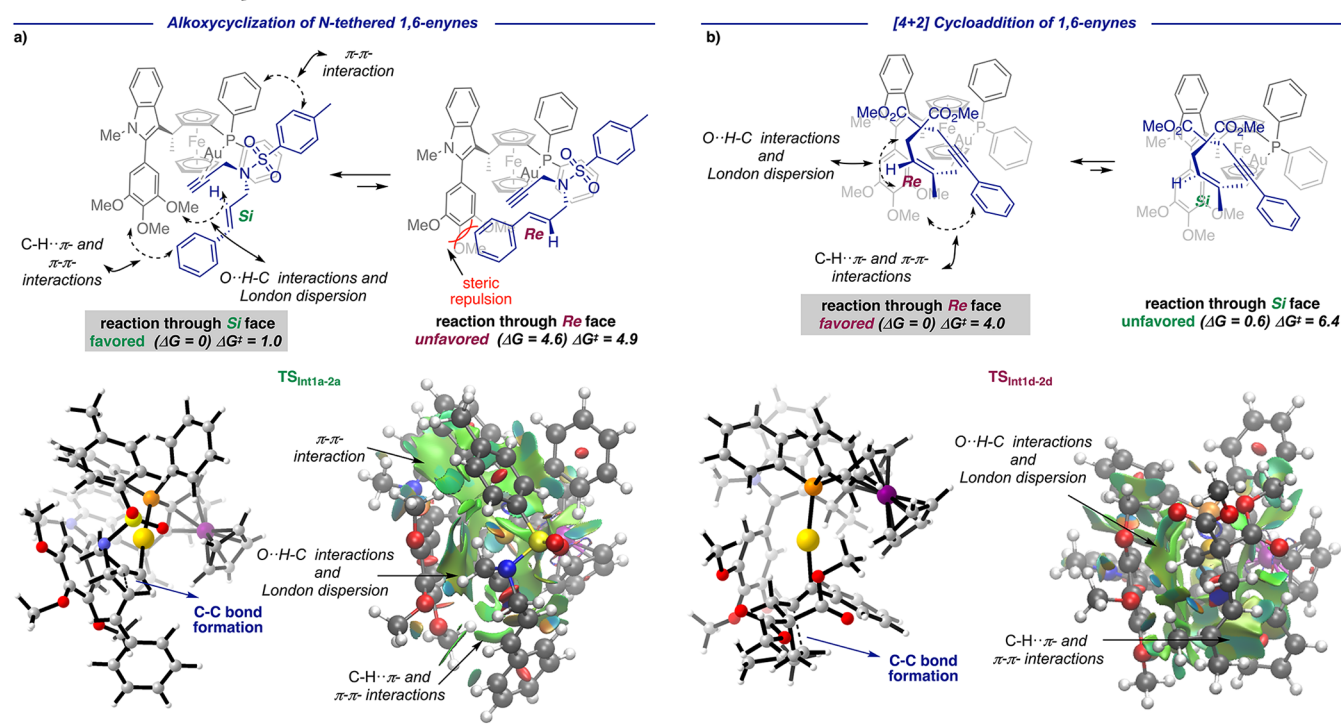
**Scheme 5. Enantioselective Methoxycyclization of Aryl-Substituted 1,6-Enynes **9a–i****



## MECHANISTIC INVESTIGATIONS

To unveil the mode of enantioinduction of these novel ferrocenyl-based phosphine gold(I) complexes, we performed DFT calculations. BP86-D3/6-31G(d) (C, H, P, N, O, S) and SDD (Au, Fe) PCM = CH<sub>2</sub>Cl<sub>2</sub> levels of theory were chosen due to their efficiency proved in our recent benchmark of DFT functionals using similar systems (Scheme 6).<sup>54,55</sup> We started our investigations by computing the reaction coordinate for the alkoxy cyclization of substrate **6a** (Scheme 6a) with complex (*R,R\_p*)-L. We calculated two possible minima resulting from the substrate coordinated to gold(I) through the alkyne [L<sub>1</sub>Au(*h*<sup>2</sup>-alkyne)]<sup>+</sup> (**Int1a-b**) and the reaction of the two enantiotopic *Re* and *Si* faces of the alkene (*R* and *S* pathways)

**Scheme 6. Schematic Representation of the Lowest-Energy Intermediates and Molecular Structures and NCI Plots of the Transition States<sup>a</sup> for the (a) Alkoxy cyclization of *N*-Tethered 1,6-Enyne **6a** and (b) [4 + 2] Cycloaddition of Aryl 1,6-Enyne **4a** with Catalyst (*R*,*R*<sub>p</sub>)-**L****



<sup>a</sup>(S) pathways are depicted in green and (R) pathways in purple. CYLview representations and NCI plots of the two possible transition states  $TS_{Int1a-2a}$  and  $TS_{Int1d-2d}$  are displayed. Hydrogens are omitted for clarity. Color-filled RDG isosurface: strong attractive interactions are indicated in blue (C–C or C–N bond formation), weak attractive interactions are shown in green (noncovalent interactions), and strong repulsive interactions are shown in red. Color code: P, orange; Au, yellow; N, blue; O, red; C, gray; Fe, purple; and H, white. Energy values are given in kcal/mol relative to **Int1a** and **Int1d**, respectively.

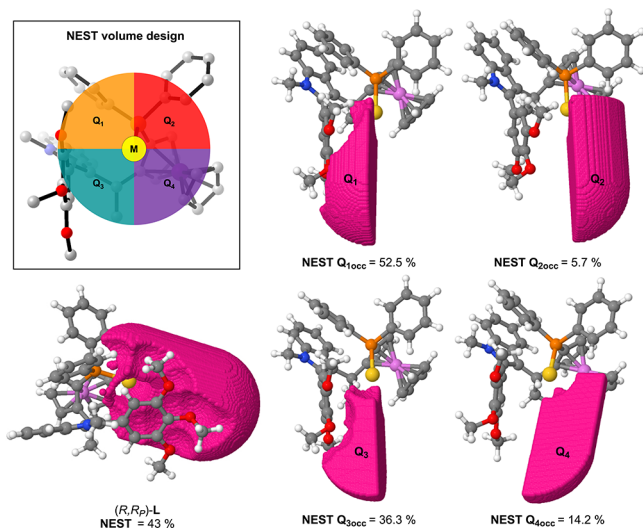
in the intramolecular electrophilic addition as the enantio-determining step of the transformation. We examined the evolution of the two possible gold(I) complexes **Int1a–b** following an exocyclic pathway, with **Int1a** being the most stable intermediate and the activation energy to reach  $TS_{Int1a-2a}$  being lower than that of the other possible pathway ( $TS_{Int1b-2b}$ ) by at least  $\Delta G^\ddagger_{R-S} = 3.9$  kcal/mol. Therefore, in agreement with the experimental results obtained with catalyst **L** (98.5:1.5 *er*) (Scheme 3), cyclopropyl gold(I) carbene **Int2a**, which leads to product **7a** with an *S* absolute configuration at C-3 of the pyrrolidine, is preferentially formed through the *Si* face of the alkene. As revealed by NCI plots,<sup>56</sup> attractive noncovalent interactions<sup>57</sup> between the aryl ring of the phosphine and the aryl ring of the tosyl moiety in the substrate and C–H– $\pi$ <sup>58</sup> and London dispersion<sup>59</sup> interactions between the trimethoxy-substituted aryl ring of the indole and the aryl ring of the substrate play a major role in the chiral folding of the substrate and in the stabilization of the transition state ( $TS_{Int1a-2a}$ ).

In the [4 + 2] cycloaddition of 1,6-enyne **4a** with the catalyst (*R*,*R*<sub>p</sub>)-**L**, the reaction proceeds from the most stable [LAu(*h*<sup>2</sup>-alkyne)]<sup>+</sup> intermediate **Int1d** through the lowest-energy transition state  $TS_{Int1d-2d}$  (favored by 2.4 kcal/mol compared to the *S* pathway,  $TS_{Int1c-2c}$ ) to yield product **5a** with *R* configuration, by the reaction of the alkyne through the *Re* prochiral face of the alkene (Scheme 6b). Attractive CH– $\pi$  and  $\pi$ – $\pi$  interactions between the aryl ring of the enyne and

the trimethoxy-substituted aryl ring of the ligand are also stereocontrolling elements.

## DATA ANALYSIS

The above presented experimental data on the enantiomeric ratio (*er*) for the [4 + 2] cycloaddition of enyne **4a** (Table 1) and the methoxycyclization of *N*-tethered 1,6-enyne **6a** (Tables S9 and S10 in the Supporting Information) for catalysts **B–I** and **K–Q** have been analyzed by constructing (multi)linear regression models with one or two chemical descriptors. This data analysis has a double aim: on the one hand, the elucidation of the determinant factors for enantioselectivity and on the other the evaluation of the prediction ability of the current approach as well as the performance of our recently developed descriptor NEST (see below). The present methodological strategy is similar to the one we used in previous studies.<sup>36,60</sup> The *er* data have been regressed in front of a total of 311 standardized descriptors: (i) derived from NEST,<sup>36</sup> a web tool designed on purpose by some of us for the description of quadrant volume occupation of catalysts bearing elongated ligands (Figure 3); (ii) obtained from DFT-optimized precatalyst (L–Au–Cl) structures, such as geometrical parameters or charges; (iii) based on the structure, such as Hammett parameters; (iv) automatically obtained from RDKit<sup>61</sup> (for a full list of descriptors, see Table S12 in the Supporting Information). The data analysis was carried out while results of *er* for catalysts **A–N** were available but not those for catalyst **O–Q**; hence, these were used to test



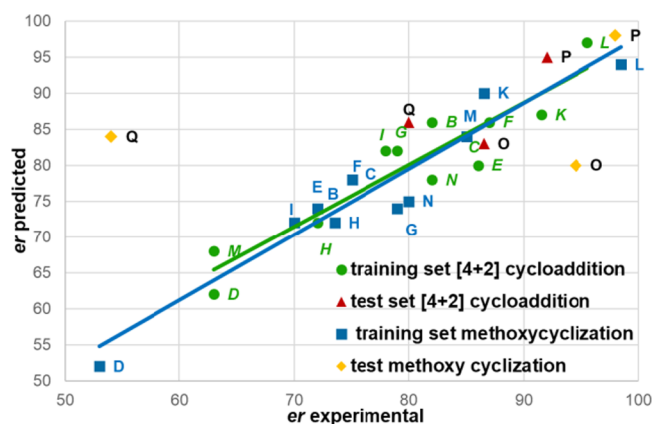
**Figure 3.** NEST volume design and an example of the NEST occupied volume of complex **L** ( $a = 5 \text{ \AA}$ ).

the validity and the scope of the model. Catalyst **A** was left out of the analysis due structural differences from the rest of catalysts (lack of phenyl group at the C-2 position of indole) (Figure 1), and **J** was excluded due to the lack of reactivity for methoxycyclization.<sup>51</sup>

In the data analysis of the  $er$  obtained for each catalyst in the [4 + 2] cycloaddition of enyne **4a**, we observed that a single descriptor was not enough to reproduce the  $er$ , due to low coefficients of determination ( $r^2 \leq 0.45$ ) and high mean squared error ( $MSE \geq 5$ ) and maximum absolute error ( $MAE \geq 13$ ) for the training set.<sup>51</sup> When two descriptors were used instead, the prediction ability was reached with several pairs of descriptors; with  $r^2$  up to 0.89 and MAE down to 5.7  $er$  for the training set, all pairs of descriptors with the best statistical results were carefully analyzed, also taking into account their chemical interpretability.<sup>51</sup> From this analysis, we selected the pair “NEST Min. horizontal 5” (descriptor 164) and “H bond and C–H  $\pi$  bond acceptor” (descriptor 226), which has  $r^2 = 0.86$ ,  $MSE = 3.1$ , and  $MAE = 6.4$   $er$  for the training set. The combination of these two descriptors explained the experimental  $er$  and, at the same time, revealed the origin of the enantioselectivity. The “NEST Min. horizontal 5” is a steric measure of the space available in the catalyst; more precisely, it is the NEST<sup>36</sup> minimum horizontal occupation, the lowest value for the addition of the occupation of the two top quadrants ( $Q1 + Q2$ ) or bottom ( $Q3 + Q4$ ), with the phosphine oriented with the ferrocene carbon on the  $z$  axis and a NEST length of 5.

The descriptor “H bond and C–H  $\pi$  bond acceptor” measures the ability of the substituents on the phenyl ring at the C2 position of the indole to set stabilizing interactions with the substrate by either accepting hydrogen bonds or C–H  $\pi$  interactions. This pair of descriptors applied to catalysts **O–Q** (test set) led to a very good agreement with an MAE of 6.5  $er$ . The results are summarized in Table 2. Considering that catalysts in the test set (**O–Q**) present important structural changes with respect to the ones in the data set (**A–N**), the quality of the predictions is remarkable. These results support the idea that the pocket for the substrate in the catalyst and the ability to accept hydrogen bonds or C–H  $\pi$  interactions of the ligand govern the enantioselectivity.

**Table 2.** Results of Statistical Analysis



cat.	$er$ [4 + 2] <sup>a</sup> exp. <sup>b</sup>	$er$ [4 + 2] <sup>a</sup> pred. <sup>b</sup>	$er$ methox. <sup>c</sup> exp. <sup>b</sup>	$er$ methox. <sup>c</sup> pred. <sup>b</sup>
B	82:18	86:14	72:28	74:26
C	85:15	84:26	75:25	78:22
D	63:27	62:28	53:47	52:48
E	86:14	80:20	72:28	74:26
F	87:13	86:14	75:25	78:22
G	79:21	82:18	79:21	74:26
H	72:28	72:28	73.5	72:28
I	78:22	82:18	70:30	72:28
K	91.5:8.5	87:13	86.5:13.5	90:10
L	95.5:4.5	97:3	98.5:1.5	94:6
M	63:37 <sup>d</sup>	68:32	85:15	84:16
N	82:18 <sup>d</sup>	78:22	80:20	75:25
O	86.5:13.5 <sup>d</sup>	83:27	94.5:5.5	80:20 <sup>e</sup>
P	92:8	95:5	98:2	98:2
Q	80:20 <sup>d</sup>	86:14	54:46	84:16

<sup>a</sup>[4 + 2] Cycloaddition of enyne **4a**. <sup>b</sup>Exp. indicates experimental values (measurements); pred. indicates predicted values within the statistical model. <sup>c</sup>Methoxycyclization of enyne **6a**. <sup>d</sup> $er$  for the opposite catalyst enantiomer (see Data Treatment in the Supporting Information). <sup>e</sup>Taking not the lowest-energy conformer of the precatalyst.

We then performed the analysis for the  $er$  obtained in the methoxycyclization of  $N$ -tethered 1,6-enyne **6a** (Scheme 3).<sup>51</sup> We found that, using one descriptor, the ability of the model to explain the results was slightly better than for the [4 + 2] cycloaddition of **4a** but still exhibited low  $r^2 \leq 0.76$ ,  $MSE \geq 4$ , and  $MAE \geq 14$  for the training set.<sup>51</sup> When two descriptors were used, results improved up to  $r^2 \leq 0.92$ ,  $MSE \geq 2.5$ , and  $MAE \geq 4.3$  for the training set. All best pairs of descriptors were carefully analyzed and the pair of descriptors “NEST Min. vertical 5” (descriptor 161) and “H bond and C–H  $\pi$  bond acceptor” (descriptor 226) was selected, with  $r^2 = 0.91$ ,  $MSE = 2.8$ , and  $MAE = 5.0$   $er$  for the training set (see Table 2 and the Supporting Information for further details). The NEST minimum vertical occupation is the lowest value for the addition of the occupation of the two right quadrants ( $Q2 + Q4$ ) or left ( $Q1 + Q3$ ), with the same phosphine orientation and NEST length as for the [4 + 2] cycloaddition.<sup>51</sup> Note that the selected pair in this case is similar to the one for the [4 + 2] cycloaddition: the same descriptor for the ability of accepting hydrogen and C–H- $\pi$  interactions and a similar NEST descriptor with different orientations of the two consecutive quadrants, horizontal in one and vertical in the other. This indicates that enyne **4a** and  $N$ -tethered 1,6-enyne **6a**

coordinate in slightly different ways, and steric effects appear in different regions of the catalyst structure. When the test set was applied to this reaction, results were very good for catalyst **P** (98:2 *er exp* vs 98:2 *er pred*) but poor for catalysts **O** (94.5:5.5 *er exp* vs 70:30 *er pred*) and **Q** (54:46 *er exp* vs 84:16 *er pred*). For **O**, we suggest that the lowest-energy geometry of the catalyst is likely not the reactive one, as the large phenyl substituents block the coordination of the substrate. These phenyl groups can easily rotate and accommodate to a conformation with a slightly higher energy (1.3 kcal mol<sup>-1</sup>); using this conformer, the prediction improved (94.5:5.5 *er exp* vs 80:20 *er pred*). For catalyst **Q**, we cannot quantitatively reproduce the observed enantioselectivity, but, considering that the enantiomeric ratio was low (similar to **A**) and knowing the bulkiness of the *N*-tosyl group, the reactive species for catalyst **Q** might need to undergo significant conformational changes for the coordination of enyne **6a**. Note that this does not occur in the [4 + 2] cycloaddition of **4a**, where the same catalysts are able to induce enantioselectivity.

## CONCLUSIONS

In conclusion, we designed a new family of chiral gold(I) ferrocenyl phosphine complexes and applied them to the gold(I)-catalyzed enantioselective cyclization of three structurally different 1,6-enynes with excellent enantioselectivities. Noncovalent interactions were found to be key elements in the efficient folding of enynes by a combination of experimental, crystallographic, and computational methods. NEST analysis was applied to our system, which allowed for the validation and reliable prediction of enantioselectivities, highlighting the importance of the size and position of the free space for substrate coordination and the importance of attractive weak interactions. This work provides a platform for predicting and advancing rational ligand design for asymmetric gold(I) catalysis while streamlining the design of novel chiral ligand scaffolds.

## ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acscatal.4c07495>.

Experimental procedures, characterization data, NMR spectra, SFC and HPLC trace chromatograms, computational details, NEST analysis results, DFT calculations, and crystallographic data (PDF)

### Accession Codes

CCDC 2406055–2406061 and 2406222 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: + 44 1223 336033.

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## Notes

The authors declare no competing financial interest.

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