



Growth, spectroscopy and laser operation of disordered Tm,Ho:NaGd(MoO₄)₂ crystal

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ABSTRACT

We report on the growth, structure refinement, Raman and optical spectroscopy, and first laser operation of Tm³⁺,Ho³⁺-codoped disordered sodium gadolinium double molybdate. A single crystal of 5.55 at% Tm, 0.43 at% Ho:NaGd(MoO₄)₂ was grown by the Czochralski method in argon atmosphere. Its structure was refined by the Rietveld method (sp. gr. *I*4₁/*a*, lattice constants: *a* = 5.237(5) Å and *c* = 11.443(6) Å). The Raman spectroscopy revealed a dominant high-frequency mode at 883 cm⁻¹ due to symmetric stretching vibrations of the [MoO₄]²⁻ groups. The stimulated-emission cross-section for the ⁵I₇ → ⁵I₈ transition of the Ho³⁺ ion is 1.84 × 10⁻²⁰ cm² at 2043 nm for π-polarization and the emission bandwidth is as broad as 66 nm for σ-polarized light. The parameters of the Tm³⁺ → Ho³⁺ energy transfer were quantified revealing a thermal equilibrium decay time of the ³F₄(Tm³⁺) and ⁵I₇(Ho³⁺) multiplets of 3.8 ms. The continuous-wave Tm,Ho-laser generated 548 mW at 2045 – 2055 nm with a slope efficiency of 32.4 %, a laser threshold of 98 mW and a liner polarization (π). Due to its inhomogeneously broadened emission bands, the Tm,Ho:NaGd(MoO₄)₂ crystal is promising for mode-locked lasers.

1. Introduction

Current trends in laser engineering include the research on ultrafast sources. The development of laser oscillators capable of generating ultrashort (few optical cycle) pulses is strongly dependent on material engineering capable of tailoring the gain bandwidth of active media [1]. Broad, smooth and nearly structureless gain profiles are desirable. They are readily achieved in rare-earth doped glasses [2]. However, glassy gain media suffer from poor thermal properties. An alternative approach consists of utilizing crystalline materials with structure disorder. Such disordered crystals, e.g., garnets or aluminates, have recently gained a lot of attention for the generation of ultrashort pulses from mode-locked

lasers [3,4]. The structure disorder can induce a significant inhomogeneous spectral line broadening leading to a “glassy-like” spectroscopic behavior at higher thermal conductivity [5].

Rare-earth-doped tetragonal scheelite-type double tungstate and double molybdate crystals with a common general formula MRE(TO₄)₂, where M represents an alkali metal (Li or Na), RE denotes a rare-earth (La, Gd, Y, Lu) or Bi, and T is W or Mo, respectively, are known for applications as laser gain media, including ultrafast lasers [6–9]. They adopt a centrosymmetric space group *I*4₁/*a* or lower space groups (*I*4, *I*/4) depending on the particular character of a nearly random distribution of M⁺ and RE³⁺ cations over two non-equivalent lattice sites (Wyckoff: 2*b* and 2*d*) with *S*₄ symmetry [6,10–12]. The local disorder is

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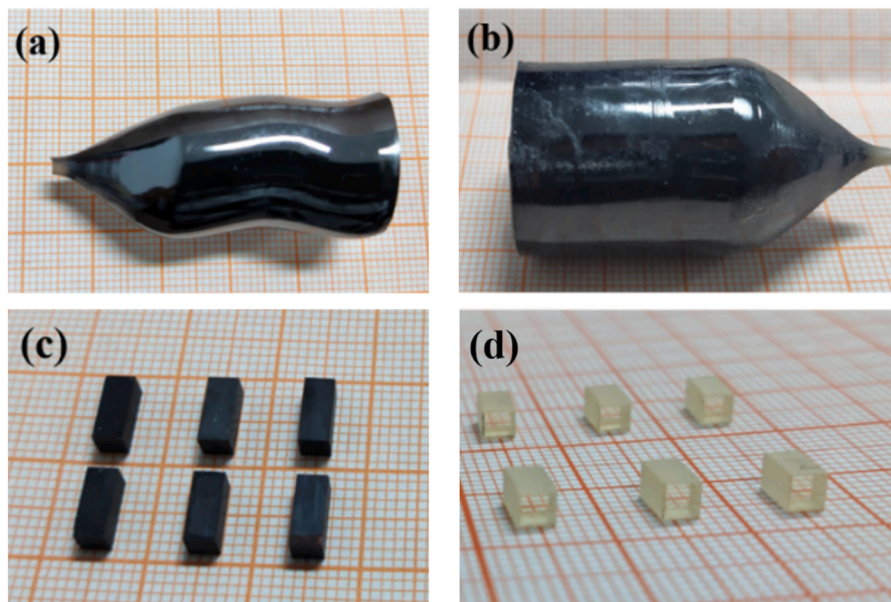


Fig. 1. Photographs of the as-grown Tm,Ho:NaGd(MoO₄)₂ crystals: growth (a) using a Pt wire; (b) using a single-crystalline seed oriented along the [001] axis; (c,d) parallelepipedic crystal samples: (c) before annealing, (d) after annealing and polishing.

explained by the short-range M⁺/RE³⁺ distribution, i.e., the second coordination sphere, around each of the 2b and 2d sites. In this way, multiple cationic environments around the dopant ions replacing for the host-forming RE³⁺ cations can be expected, adopting symmetries from S₄ (when all the neighbor cations are the same) to C₂ or even C₁ (for “mixed” environments). This leads to a strong spectral line broadening even at cryogenic temperatures. The highest degree of disorder is associated with equal occupancy factors for both M⁺ and RE³⁺ cations.

The structure disorder of MRE(TO₄)₂ crystals is manifested in broad and nearly structureless absorption and emission bands for polarized light (these crystals are optically uniaxial [13]). Other advantages of these compounds include relatively easy doping due to a substitutional rare-earth site, and a weak concentration quenching of luminescence. The considered crystals are also known as efficient Raman frequency shifters [14]. The MRE(TO₄)₂ compounds can be grown by the Czochralski (Cz) or Top-Seeded solution Growth methods. Despite their relatively low thermal conductivity ($\kappa_a = 1.10$ and $\kappa_c = 1.24 \text{ Wm}^{-1}\text{K}^{-1}$ for NaGd(WO₄)₂ [15]), owing to their overall thermo-optic behavior [16], rare-earth doped MRE(TO₄)₂ crystals outperform glasses.

For solid-state laser development targeting the eye-safe spectral range of 2 μm , Thulium (Tm³⁺) and Holmium (Ho³⁺) dopant ions are usually considered. Their emission originates from the spectrally overlapping ³F₄ → ³H₆ and ⁵I₇ → ⁵I₈ transitions, respectively. The Ho³⁺ ion emits above 2 μm , in a spectral range free of water absorption lines in the air, which represent a limiting factor for femtosecond pulse generation. To excite Ho³⁺ ions, often, the Tm³⁺, Ho³⁺ codoping scheme is employed [17–19], as Tm³⁺ ions act as donors (D) by absorbing the pump radiation around 0.8 μm and transferring it non-radiatively to Ho³⁺ ones (acceptors, A), ³F₄(Tm³⁺) → ⁵I₇(Ho³⁺). Apart from the simplified pump configuration which can rely on commercial and power scalable AlGaAs laser diodes, the codoped materials benefit from combined gain bandwidths of the two ions, contributing to the pulse shortening in mode-locked lasers [20,21]. Recently, sub-100 fs lasers emitting around 2 μm have been demonstrated employing Tm³⁺, Ho³⁺-codoped disordered gain media [3,4,20,21]. Ultrafast 2- μm lasers find their application niches in frequency down-conversion towards the mid-infrared spectral range, THz and high-harmonic generation and time-resolved pump-probe spectroscopy [22]. Tunable 2- μm lasers based on Thulium and/or Holmium ion emission are in general interesting not only for spectroscopic research, but also important for

applications in medicine (laser surgery), environmental monitoring (e.g., H₂O, CO₂) and material processing (welding and cutting of plastics).

MRE(TO₄)₂ type crystals have been studied for single Tm³⁺ [23–26] and Ho³⁺ [27,28] doping, as well as Tm³⁺, Ho³⁺ codoping [7,29,30], regarding their spectroscopic properties and laser operation. Han *et al.* reported on continuous-wave and tunable operation of Tm³⁺, Ho³⁺-codoped NaY(WO₄)₂ and NaLu(WO₄)₂ crystals: with the former material, an output power of 265 mW at 2.05 μm was extracted corresponding to a slope efficiency of 43 % and a broadband wavelength tuning across 1825 – 2080 nm was obtained [31]. Tang *et al.* explored the Tm,Ho:LiGd(MoO₄)₂ crystal delivering only 25 mW at 2.05 μm with a slope efficiency of 20 % [30].

In contrast to potassium containing double tungstate and molybdate compounds, which undergo polymorphic transformations upon cooling after growth, for Na-based tungstates and molybdates, the tetragonal phase is stable in the whole temperature range from 300 K to the melting point, allowing for the Czochralski growth (except for NaLu(MoO₄)₂ and NaLu(WO₄)₂, which melt incongruently). Tetragonal double molybdates offer better spectroscopic properties as compared to their tungstate counterparts, namely, higher absorption and emission cross-sections [8]. Among them, sodium gadolinium double molybdate, NaGd(MoO₄)₂ (space group: *I*4₁/*a*, abbreviated: NGM), attracts a lot of attention. The spectroscopy and (in certain cases) laser operation of NaGd(MoO₄)₂ doped with Nd³⁺ [32], Yb³⁺ [33], Tm³⁺ [6,23,34], and Ho³⁺ [35,36] ions were reported. The authors of [37] performed a comprehensive study of spectroscopic characteristics of a series of Tm³⁺-doped NaLa_{1-x}Gd_x(MoO₄)₂ mixed crystals for the entire range of *x* = 0 – 1 including the end-member of this series, the Tm:NaGd(MoO₄)₂ crystal. Han *et al.* presented a Tm:NaGd(MoO₄)₂ laser delivering 850 mW at 1.90 μm with a slope efficiency of 45 % and a laser threshold of 180 mW, as well as a wavelength tuning from 1875 to 1975 nm [24]. Bol’shchikov *et al.* reported on diode-pumped laser operation of a “mixed” Tm:NaLa_{1/3}Gd_{2/3}(MoO₄)₂ crystal delivering 210 mW at 1.91 μm with a slope efficiency of 27 % and a laser threshold of 550 mW, as well as a wavelength tuning from 1860 to 1955 nm, with the beam quality being close to the diffraction-limited one [38]. Wang *et al.* studied the thermal and unpolarized spectroscopic properties of a Tm,Ho:NaGd(MoO₄)₂ crystal [29].

In the present work, we report on the Cz growth, structure refinement, polarized Raman and optical spectroscopy and first continuous-

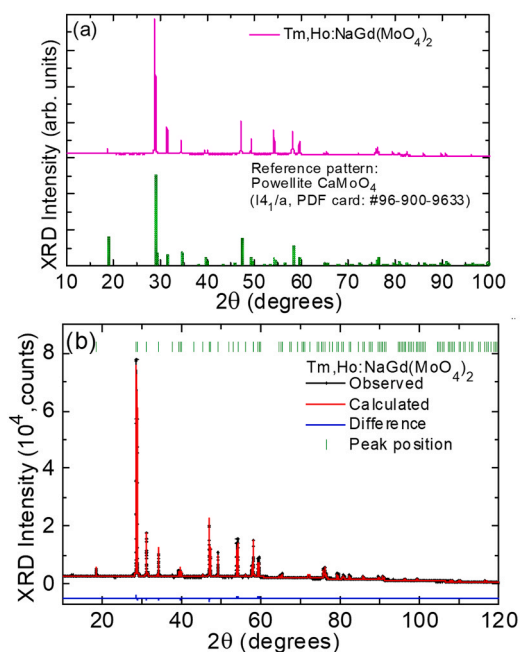
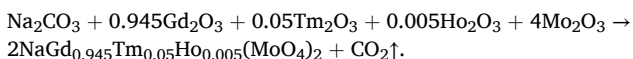


Fig. 2. (a) X-ray powder diffraction (XRD) pattern of $\text{Tm,Ho:NaGd}(\text{MoO}_4)_2$ crystal (magenta) and the reference pattern of powellite CaMoO_4 adopting the sp. gr. $I4_1/a$ (green bars), PDF card #96–900–9633, COD database; (b) Rietveld refinement: observed (black), calculated (red), and residual (blue) patterns, green dashes - Bragg reflections.

wave laser operation of $\text{Tm}^{3+}, \text{Ho}^{3+}$ -codoped disordered $\text{NaGd}(\text{MoO}_4)_2$ crystals with the goal of developing novel broadly emitting materials for ultrafast lasers around $2 \mu\text{m}$.

2. Crystal growth

A $\text{NaGd}(\text{MoO}_4)_2$ single crystal co-doped with 5 at% Tm^{3+} and 0.5 at% Ho^{3+} (in the melt) was grown by the standard Cz method in argon atmosphere using an iridium crucible. The starting materials were Na_2CO_3 (purity: 4N), Gd_2O_3 , Tm_2O_3 , Ho_2O_3 , and Mo_2O_3 (5N). They were accurately weighed according to the chemical formula $\text{NaGd}_{0.945}\text{Tm}_{0.05}\text{Ho}_{0.005}(\text{MoO}_4)_2$. To counteract the volatilization of Mo_2O_3 during the synthesis of the polycrystalline material and the crystal growth, an excess of 2.0 wt% Mo_2O_3 was added. The equation of the solid-state chemical reaction is expressed as follows:



The mixture was subsequently put together, ground, and heated at 650°C for 10 h (h) in a platinum crucible to decompose Na_2CO_3 . After cooling to room temperature (RT, 293 K), the mixture was pressed into pellets and reheated at 950°C for 15 h to form the tetragonal molybdate phase via solid-state reaction.

The synthesized polycrystalline material was placed in an iridium crucible and melted using an intermediate-frequency heater. Typically, to eliminate bubbles from the melt and prevent the formation of polycrystals during growth, the melt was maintained at a temperature $30\text{--}50^\circ\text{C}$ above the melting point for about 2–3 h before cooling it to the melting point. During the crystal growth, the pulling rate varied from 0.5 to 1.0 mm/h, and the crystal rotation speed was 8–15 revolutions per minute (rpm). An automated system was used to regulate the boule diameter. Following completion of the growth, the crystal was cooled to RT at a stepped rate of $15\text{--}25^\circ\text{C/h}$.

Initially, a Pt wire was used as a seed for the crystal growth, resulting in a crystal boule with a nearly black coloration and spiral growth, Fig. 1 (a). Subsequently, a seed oriented along the crystallographic [001] axis

was cut from this first crystal, yielding a crystal boule with a larger size and minimal spiral growth, Fig. 1(b). The observed spiral growth could be attributed to poor heat dissipation due to the low thermal conductivity of this double molybdate. In addition, non-symmetrical breakdown of the facet formation might contribute to this phenomenon. Both crystal boules were nearly black, primarily due to the presence of color centers associated with oxygen vacancies but annealing in air significantly improved their transparency. Small crystal samples were cut from the as-grown crystal boule, Fig. 1(c), and after annealing at 900°C for 72 h in air, the coloration was almost completely removed, resulting in nearly colorless samples, Fig. 1(d). When observing the internal quality of the polished samples, Fig. 1(d), with a He-Ne laser, no scattering centers were found, indicating an excellent optical quality.

Using X-ray fluorescence, the actual Tm^{3+} and Ho^{3+} doping levels in the $\text{NaGd}(\text{MoO}_4)_2$ crystal were measured to be 5.55 at% (ion density: $N_{\text{Tm}} = 3.54 \times 10^{20} \text{ at/cm}^3$) and 0.43 at% ($N_{\text{Ho}} = 2.74 \times 10^{19} \text{ at/cm}^3$), respectively, and the segregation coefficients of the dopant ions were evaluated, $K_{\text{RE}} = C_{\text{crystal}}/C_{\text{melt}}$, yielding $K_{\text{Tm}} = 1.11$ and $K_{\text{Ho}} = 0.86$. The obtained value of K_{Tm} appeared to be much higher than that determined earlier, $K_{\text{Tm}} = 0.54$ [37].

3. Experimental

The X-ray powder diffraction (XRD) pattern was measured using a Bruker D2 Phaser diffractometer for diffraction angles 2θ in the range of $5\text{--}120^\circ$ with a step of 0.02° using $\text{Cu K}\alpha 1$ radiation.

The polarized Raman spectra were measured for an *a*-cut $\text{Tm,Ho:NaGd}(\text{MoO}_4)_2$ crystal using a Renishaw inVia confocal Raman microscope equipped with a $\times 50$ Leica objective, an Ar⁺ ion laser ($\lambda = 488 \text{ nm}$), and a set of polarizing optics. Porto's notations were used, *a* (*ij*) \bar{a} , where "a" and " \bar{a} " represent the propagation directions of the incident and scattered light, respectively, and "i" and "j" signify the corresponding polarization states (π or σ).

The polarized absorption spectra were measured using a Varian CARY 5000 spectrophotometer and a calcite Glan-Taylor polarizer.

The luminescence spectra were measured using an optical spectrum analyzer (OSA, Yokogawa AQ6376) purged by N_2 gas to mitigate the influence of the structured water vapor absorption in the air around $2 \mu\text{m}$. The emission was collected using a CaF_2 lens and a large mode area ZrF_4 fiber. A metal-grid polarizer was also employed. The excitation was provided by a Ti:Sapphire laser at 795 nm (excitation to the $^3\text{H}_4$ multiplet of Tm^{3+}).

The luminescence decay was studied employing a nanosecond optical parametric oscillator (Horizon, Continuum), a 1/4 m monochromator (Oriel 77200), an InGaAs detector, and an 8 GHz digital oscilloscope (DSA70804B, Tektronix). To mitigate the impact of radiation trapping, the samples were ground into fine powders.

The low-temperature measurements were enabled by a cryostat (Oxford Instruments, model SU 12) equipped with helium-gas close-cycle flow.

4. Crystal structure and Raman spectra

4.1. X-ray diffraction

The crystal structure of tetragonal scheelite-type double tungstate and double molybdates was a subject of detailed investigations by multiple groups, see for example Cano-Torres *et al.* [6]. We employed the powder XRD analysis and the Rietveld method for refining the unit-cell parameters and atomic coordinates of the $\text{Tm}^{3+}, \text{Ho}^{3+}$ -codoped $\text{NaGd}(\text{MoO}_4)_2$ crystal. All the observed diffraction peaks demonstrated a close correspondence with the reference powellite phase, CaMoO_4 (tetragonal class, space group $I4_1/a - C_4^2_h$, No. 88), the Crystallography Open Database (COD) card #96–900–9633, Fig. 2(a). The lack of diffraction peaks related to impurity and secondary phases indicated a single-phase nature of the grown crystal.

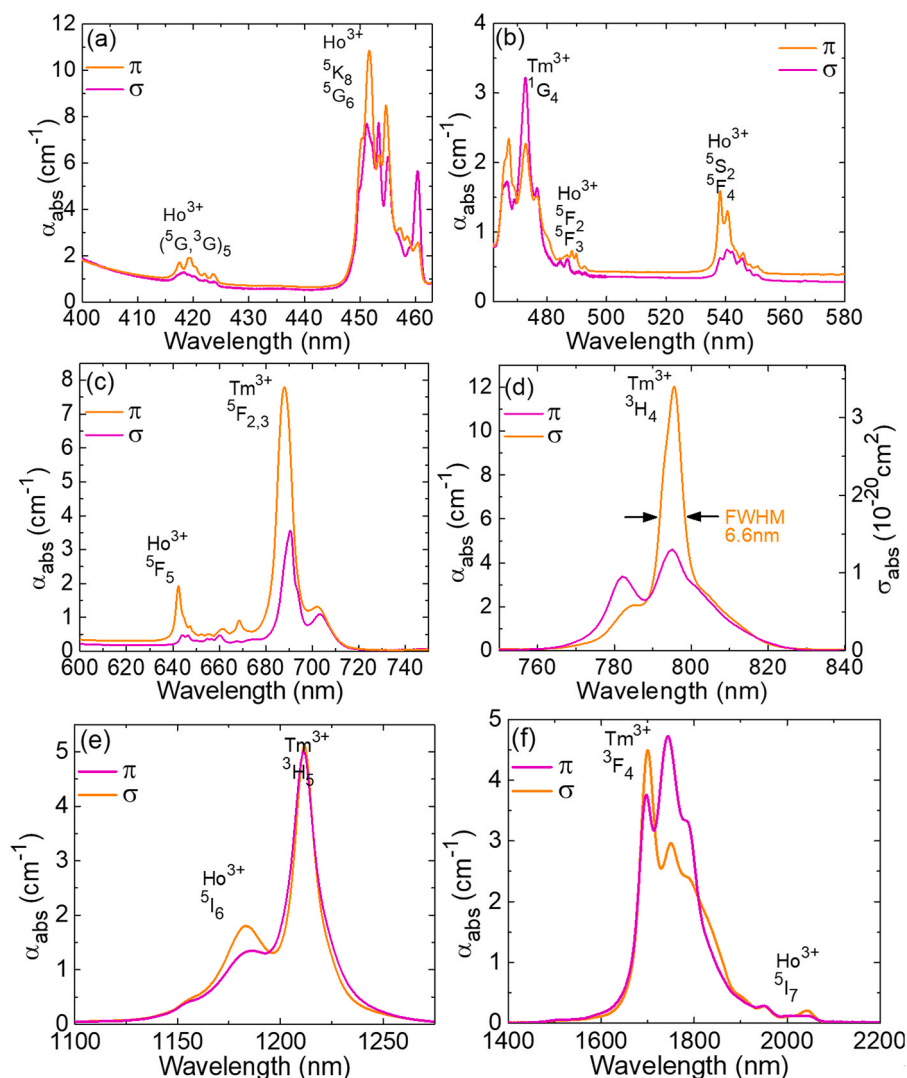


Fig. 5. (a-f) Polarized absorption spectra of the 5.55 at% Tm, 0.43 at% Ho:NaGd(MoO₄)₂ crystal, light polarizations: π and σ .

interatomic distance is along the vector $[u,v,w] = [1/2,0,1/4]$, measuring 3.8784(9) Å. Notably, this distance extends to 5.2375(3) Å along the a -axis in the a - c plane.

4.2. Raman spectroscopy

The polarized Raman spectra of the a -cut Tm,Ho:NaGd(MoO₄)₂ crystal are shown in Fig. 4. The spectra consist of broadened, mutually overlapping bands. This is a typical situation for structurally disordered crystals. The spectra can be divided into two frequency regions: a low frequency region at 103–417 cm⁻¹, where the modes are due to the “skeletal” (external) motions, and a high-frequency region, between 748 and 931 cm⁻¹, for which the modes originate from the internal vibrations of the [MoO₄]²⁻ groups. No Raman bands between these two ranges are observed due to the lack of double oxygen bridge vibrations in contrast to monoclinic double tungstates and molybdates. The strongest high-frequency Raman peak at 883 cm⁻¹ corresponds to symmetric stretching vibrations of the [MoO₄]²⁻ anion, the $\nu_1(A_g)$ mode according to the scheelite $I4_1/a$ symmetry [42].

5. Optical spectroscopy

The tetragonal NaGd(MoO₄)₂ crystal is optically uniaxial, and its optical axis is parallel to the crystallographic c -axis. The two principal

light polarizations are $E \parallel c$ (π) and $E \perp c$ (σ), and the corresponding refractive indices are n_e and n_o . At $\sim 2 \mu\text{m}$, their values are $n_e = 1.967$ and $n_o = 1.960$ (indicating an optically positive crystal). An a -cut crystal was used for the spectroscopic measurements giving access to both principal light polarizations.

5.1. Optical absorption

The polarized absorption spectra are shown in Fig. 5(a–f). They contain the absorption bands of both Tm³⁺ and Ho³⁺ ions. For Tm³⁺ dopant ions, the transitions in absorption originate from the ³H₆ ground state and terminate at various excited states spanning from ³F₄ to ¹G₄. For Ho³⁺ ions, the ground state is ⁵I₈ and the observed transitions in absorption terminate at the excited states from ⁵I₇ to (⁵G, ³G)₅. The multiplet assignment follows the work of Carnall et al. [43]. The UV absorption edge for the studied crystal is at 387 nm (optical bandgap: 3.20 eV). The bandgap of an isostructural NaY(MoO₄)₂ crystal (undoped) E_g is 3.28 eV (determined from the absorption edge) [44]. Nearly the same UV absorption edge had been observed earlier for scheelite-like NaLa(MoO₄)₂ crystal [45]. However, when the same crystal was grown using a starting reagent of MoO₃ which passed through a special procedure of very deep purification, the UV absorption edge of the crystal was further blue shifted by ~ 15 nm. Therefore, the optical absorption in the range of at least 370–390 nm is caused by accidental impurities.

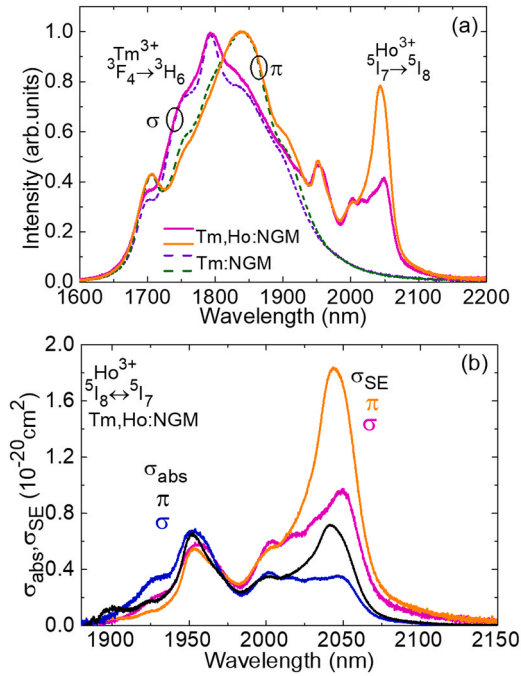


Fig. 6. Emission properties of Tm^{3+} and Ho^{3+} ions in the $\text{NaGd}(\text{MoO}_4)_2$ crystal: (a) luminescence spectra of a codoped crystal (solid curves), the spectra for a singly Tm^{3+} -doped one (dashed curves) are given for comparison; (b) absorption, σ_{abs} , and stimulated-emission (SE) cross-sections, σ_{SE} , for the $^5\text{I}_7 \leftrightarrow ^5\text{I}_8$ Ho^{3+} transition. Light polarizations: π and σ .

In the visible, a weak and broad absorption band spanning from 400 to 650 nm is due to the residual color center absorption being typical for tetragonal double molybdates [46].

The polarized absorption cross-sections, σ_{abs} , corresponding to the $^3\text{H}_6 \rightarrow ^3\text{H}_4$ transition of Tm^{3+} ions, were calculated from the measured absorption coefficient (α_{abs}), $\sigma_{\text{abs}} = \alpha_{\text{abs}}/N_{\text{Tm}}$. This spectral band well matches the emission of commercial high-power spatially multimode fiber-coupled AlGaAs laser diodes used as pump sources of Nd^{3+} -doped crystals. The maximum σ_{abs} is $3.41 \times 10^{-20} \text{ cm}^2$ at 795.5 nm, corresponding to an absorption bandwidth (full width at half maximum, FWHM) of 6.6 nm for light polarization σ . For π -polarized light, the peak σ_{abs} is much lower, $1.31 \times 10^{-20} \text{ cm}^2$ at 795.1 nm, but a broader absorption bandwidth of 15.7 nm is observed. In the previous studies [34, 37], the peak absorption cross-sections of Tm^{3+} ions in the $\text{NaGd}(\text{MoO}_4)_2$ crystal were found to lie in the range of $(3.85\text{--}5.2) \times 10^{-20} \text{ cm}^2$ for σ -polarized light and $(1.39\text{--}2.3) \times 10^{-20} \text{ cm}^2$ for π -polarization. The relatively broad absorption profile of Tm^{3+} ions around 0.8 μm is expected to mitigate the impact of temperature-induced drift of the pump diode emission wavelength.

The studied double molybdate crystal features higher absorption

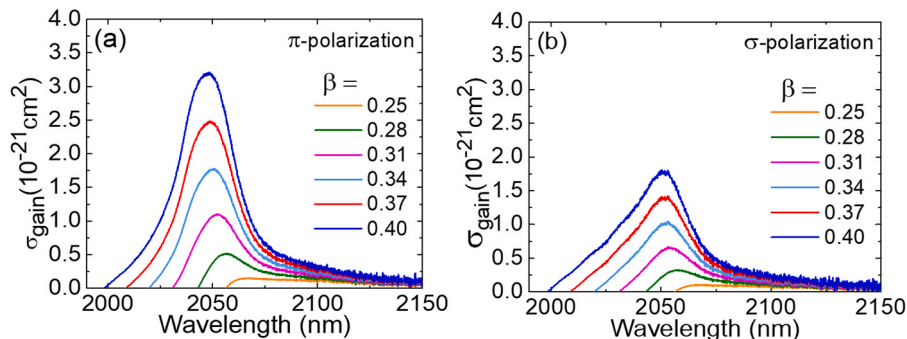


Fig. 7. Gain cross-sections, σ_{gain} , of the Ho^{3+} ion in the $\text{NaGd}(\text{MoO}_4)_2$ crystal: (a) light polarization π ; (b) light polarization σ .

cross-sections for Tm^{3+} ions as compared to its double tungstate counterpart, $\text{NaGd}(\text{WO}_4)_2$, in which the maximum σ_{abs} for the $^3\text{H}_6 \rightarrow ^3\text{H}_4$ transition is $2.9 \times 10^{-20} \text{ cm}^2$ at 795 nm for light polarization σ (FWHM ~ 8 nm); at the same wavelength, σ_{abs} amounts to $1.18 \times 10^{-20} \text{ cm}^2$ for π -polarized light [25].

5.2. Luminescence (spectra and lifetimes)

Fig. 6(a) depicts the polarized luminescence spectra of the $\text{Tm,Ho}:\text{NaGd}(\text{MoO}_4)_2$ crystal for light polarizations π and σ . For comparison, we also show the spectra measured for a singly Tm^{3+} -doped crystal. The broad emission band at 1.6–2.2 μm originates from two overlapping transitions: $^3\text{F}_4 \rightarrow ^3\text{H}_6$ Tm^{3+} and $^5\text{I}_7 \rightarrow ^5\text{I}_8$ Ho^{3+} . The studied crystal exhibits a strong anisotropy of emission properties, with higher intensity at 2.1 μm (Ho^{3+} emission) for π -polarized light. A strong spectral line broadening is observed, associated with the structure disorder of the host matrix. This results in smooth and broad emission profiles.

Fig. 6(b) illustrates the stimulated emission (SE) cross-sections, σ_{SE} , for the $^5\text{I}_7 \rightarrow ^5\text{I}_8$ Ho^{3+} transition calculated using the Füchtbauer-Ladensburg equation [47]:

$$\sigma_{\text{SE}}^i(\lambda) = \frac{\lambda^5}{8\pi(n)^2\tau_{\text{rad}}c} \frac{W_i(\lambda)}{1 / \left(3 \sum_{j=2\sigma+\pi} \int \lambda' W_j(\lambda') d\lambda' \right)}, \quad (1)$$

where λ is the luminescence wavelength, $\langle n \rangle$ is the averaged (over polarization) refractive index at the central emission wavelength, τ_{rad} is the radiative lifetime of the $^5\text{I}_7$ Ho^{3+} state, c is the speed of light, and the indices i and $j = \sigma, \pi$ indicate the polarization state. In this calculation, we used the value of τ_{rad} of 6.0 ± 0.5 ms which provided the best agreement with the data obtained using the reciprocity method [48] based on the measured absorption spectrum. This estimate agrees well with the value of 6.61 ms determined using the Judd-Ofelt theory for an

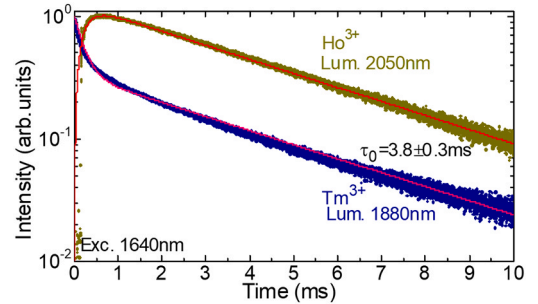


Fig. 8. Luminescence decay of Tm^{3+} and Ho^{3+} ions in $\text{NaGd}(\text{MoO}_4)_2$: a codoped crystal, decay from the $^3\text{F}_4$ Tm^{3+} and $^5\text{I}_7$ Ho^{3+} manifolds measured under resonant Tm^{3+} excitation: circles – experimental data, curves – their fits using Eq. (2).

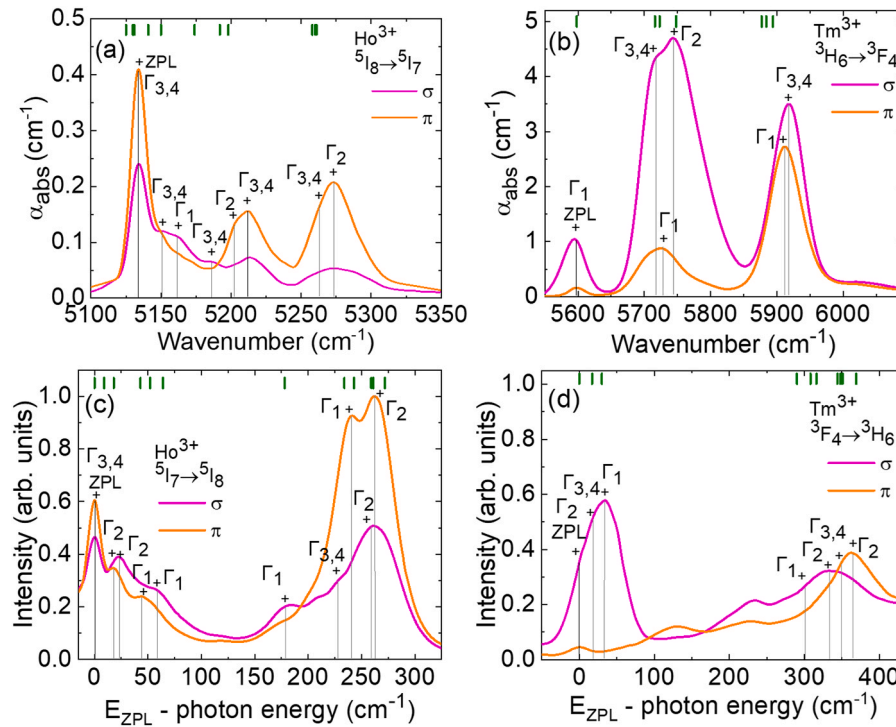


Fig. 9. Low-temperature (12 K) spectroscopy of Tm^{3+} and Ho^{3+} ions in $\text{NaGd}(\text{MoO}_4)_2$: (a,b) absorption: (a) the $^5\text{I}_8 \rightarrow ^5\text{I}_7$ Ho^{3+} transition and (b) the $^3\text{H}_6 \rightarrow ^3\text{F}_4$ Tm^{3+} transition; (c,d) luminescence: (c) the $^5\text{I}_7 \rightarrow ^5\text{I}_8$ Ho^{3+} transition and (d) the $^3\text{F}_4 \rightarrow ^3\text{H}_6$ Tm^{3+} transition. Light polarizations: π and σ . Vertical dashes – crystal-field calculations for isostructural crystals [27,51], vertical lines – assigned electronic transitions.

Table 4

Experimental crystal-field splitting (E_{exp} , in cm^{-1}) of selected Tm^{3+} and Ho^{3+} multiplets in $\text{NaGd}(\text{MoO}_4)_2$ in comparison with that for some scheelite-type molybdate crystals (E_{calc} – crystal-field calculation, Γ – irreducible representation).

Ho^{3+} $2S+1L_J$	$\text{NaGd}(\text{MoO}_4)_2$ E_{exp}	$\text{NaBi}(\text{MoO}_4)_2$ [27]			Tm^{3+} $2S+1L_J$	$\text{NaGd}(\text{MoO}_4)_2$ E_{exp}	$\text{NaLa}(\text{MoO}_4)_2$ [51]		
		E_{exp}	E_{calc}	Γ			E_{exp}	E_{calc}	Γ
$^5\text{I}_8$	0	0	0	$\Gamma_{3,4}$	$^3\text{H}_6$	0	0	3	Γ_2
	18	9	9	Γ_2		19	-	17	$\Gamma_{3,4}$
	23	22	18	Γ_2		34	41	30	Γ_1
	44	45	43	Γ_1		-	280	290	Γ_2
	59	-	52	Γ_1		302	309	308	Γ_1
	-	72	64	$\Gamma_{3,4}$		-	-	316	$\Gamma_{3,4}$
	179	179	178	Γ_1		335	-	344	Γ_2
	228	233	234	$\Gamma_{3,4}$		349	-	348	$\Gamma_{3,4}$
	241	-	243	Γ_1		-	-	351	Γ_1
	259	-	259	Γ_2		364	-	369	Γ_2
	262	260	262	Γ_2		-	-	-	-
	-	-	272	Γ_1		-	-	-	-
	-	-	-	Γ_2		-	-	-	-
	-	-	-	$\Gamma_{3,4}$		-	-	-	-
$^5\text{I}_7$	-	-	5125	Γ_2	$^3\text{F}_4$	5595	5595	5598	Γ_1
	5133	5130	5130	$\Gamma_{3,4}$		5716	5707	5716	$\Gamma_{3,4}$
	-	-	5131	Γ_2		5728	5739	5723	Γ_1
	-	-	5141	Γ_1		5744	-	5748	Γ_2
	5150	-	5150	$\Gamma_{3,4}$		-	-	5877	Γ_2
	5162	5165	5174	Γ_1		-	-	5884	Γ_1
	5187	5193	5192	$\Gamma_{3,4}$		5918	5892	5894	$\Gamma_{3,4}$
	5202	5201	5198	Γ_2		-	-	-	-
	5212	5255	5258	$\Gamma_{3,4}$		-	-	-	-
	5263	5255	5260	Γ_2		-	-	-	-
	5273	-	5261	Γ_1		-	-	-	-

isostructural $\text{Ho}:\text{NaLa}(\text{MoO}_4)_2$ crystal [28].

The maximum SE cross-section σ_{SE} is $1.84 \times 10^{-20} \text{ cm}^2$ at 2043 nm for π -polarized light (the corresponding emission bandwidth is 36 nm), and for σ -polarization, σ_{SE} is almost two times lower, $0.97 \times 10^{-20} \text{ cm}^2$ at 2049 nm, while corresponding to a much broader emission profile (FWHM: 66 nm). The significant anisotropy of stimulated-emission properties suggests linearly polarized laser radiation in *a*-cut crystals. As mentioned above, one of the key advantages of double molybdates

consists of notably higher transition cross-sections for the dopant ions (as compared to their double tungstate counterparts). Sun *et al.* reported on polarized spectroscopy of a $\text{Ho}:\text{NaY}(\text{WO}_4)_2$ crystal yielding σ_{SE} of $1.43 \times 10^{-20} \text{ cm}^2$ at 2047 nm for π -polarization supporting this statement [49] (note that the data for $\text{Ho}:\text{NaGd}(\text{WO}_4)_2$ are not available in the literature).

According to the quasi-three-level nature of the $^5\text{I}_7 \rightarrow ^5\text{I}_8$ Ho^{3+} laser transition with reabsorption at the laser wavelength, we calculated the

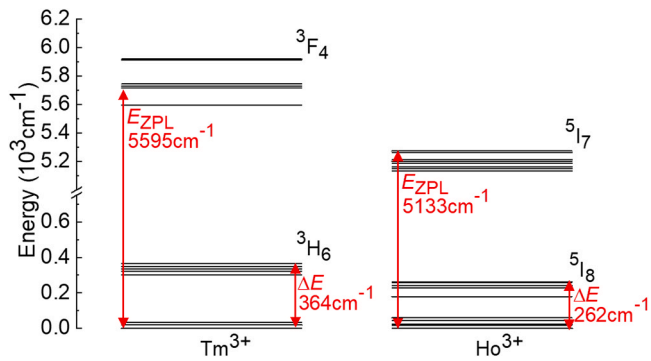


Fig. 10. Experimental crystal-field splitting of the two lowest multiplets of Tm^{3+} and Ho^{3+} ions in the $\text{NaGd}(\text{MoO}_4)_2$ crystal. E_{ZPL} – zero-phonon line energy.

gain cross-sections, $\sigma_{\text{gain}(\text{Ho})} = \beta\sigma_{\text{SE}(\text{Ho})} - (1 - \beta)\sigma_{\text{abs}(\text{Ho})}$, for light polarizations π and σ , for the $\text{NaGd}(\text{MoO}_4)_2$ crystal, see Fig. 7. Here, β is the population inversion ratio, $\beta = N_2/N_{\text{Ho}}$ where N_2 is the density of ions

excited to the upper laser level ($^5\text{I}_7$) and N_{Ho} is the total Ho^{3+} ion density. The gain profiles are nearly structureless which is a consequence of the inhomogeneous spectral line broadening inherent to this disordered compound. Higher gain is expected for π polarization. For $\beta > 0.25$, a local peak at $2.05 \mu\text{m}$ dominates in the spectra. When β reaches 0.40, the gain bandwidth (FWHM) is 31 nm, and the gain profile is peaking at 2048 nm. For σ -polarized light, at the same inversion level, the gain

Table 5

Output performance^a of Tm,Ho lasers reported recently.

Crystal	P_{out} , mW	η , %	P_{th} , mW	λ_L , nm	Polariz.	Ref.
$\text{KLu}(\text{WO}_4)_2$	451	31	450	2081	$E \parallel N_m$	[18]
CaGdAlO_4	1010	32	155	2081	π	[3]
CNGG	425	22.9	136	2084	Unpol.	[4]
GdScO_3	1160	50.5	184	2067, 2099	$E \parallel c$	[19]
$\text{NaY}(\text{WO}_4)_2$	265	43	~50	2050	π	[31]
$\text{NaGd}(\text{MoO}_4)_2$	548	32.4	98	2045–2055	π	This work

^a P_{out} – output power, η – slope efficiency, P_{th} – laser threshold, λ_L – laser wavelength.

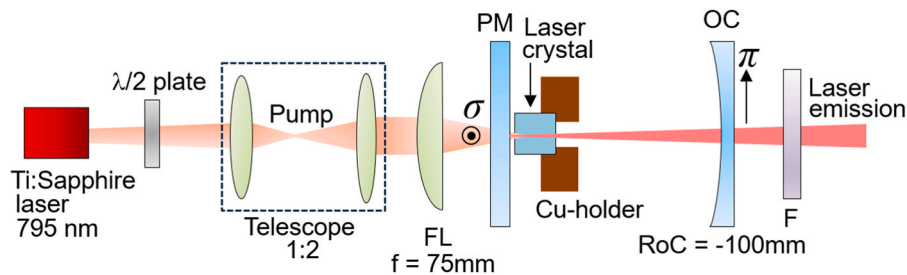


Fig. 11. Scheme of the Tm,Ho:NaGd(MoO₄)₂ laser: $\lambda/2$ – half-wave plate, FL – focusing lens, PM – pump mirror, OC – output coupler, F – long-pass filter.

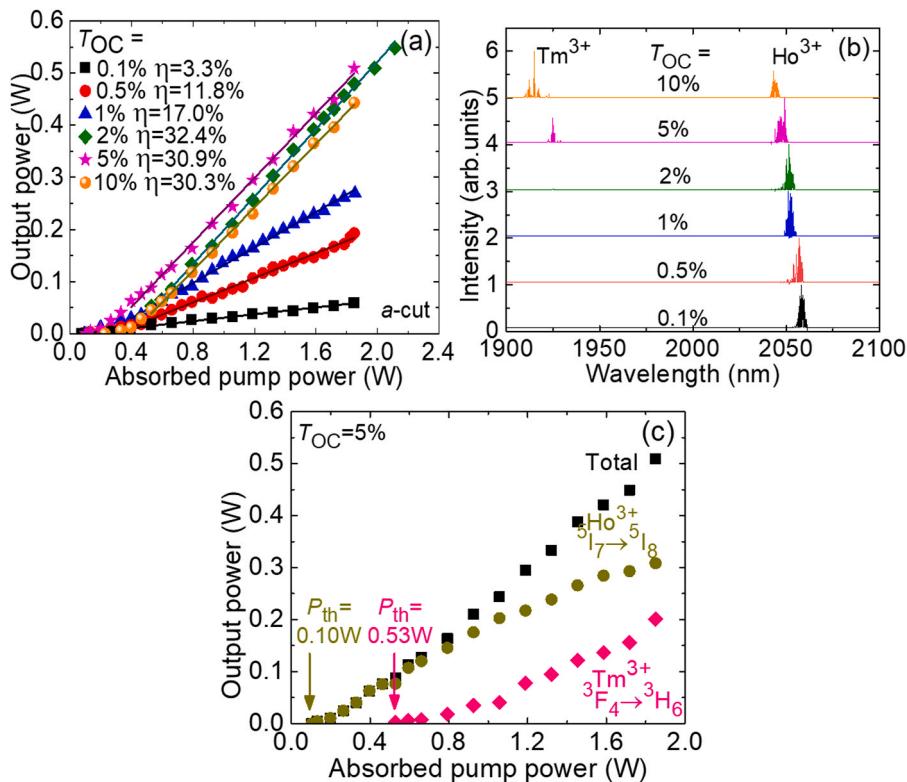


Fig. 12. Tm,Ho:NaGd(MoO₄)₂ laser: (a) input-output dependences for different output coupling rates, η – slope efficiency; (b) laser emission spectra measured well above the laser threshold, π -polarization; (c) Tm^{3+} and Ho^{3+} colasing for $T_{\text{OC}} = 5\%$. a-cut crystal.

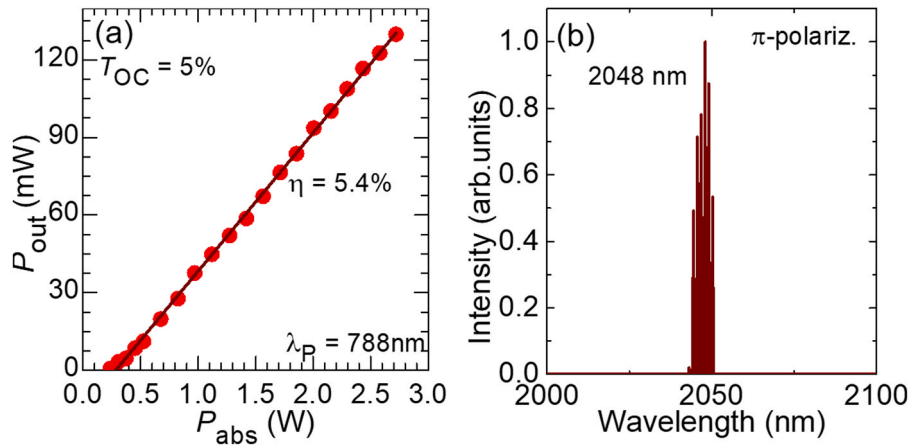


Fig. 13. Diode-pumped Tm,Ho:NaGd(MoO₄)₂ laser: (a) input-output dependences, η – slope efficiency; (b) a typical laser emission spectrum, π -polarization. $T_{OC} = 5\%$.

maximum is found at 2051 nm and the gain profile is broader (FWHM: 35 nm), offering better possibilities for wavelength tuning and ultra-short pulse generation.

The luminescence dynamics from the excited states of Tm³⁺ and Ho³⁺ ions in NaGd(MoO₄)₂, which are responsible for emission around 2 μ m, was studied under resonant excitation of Tm³⁺ (at 1640 nm, to the ³F₄ state). The luminescence was detected at two wavelengths corresponding to almost pure Tm³⁺ (1880 nm) and Ho³⁺ (2050 nm) emission. The luminescence decay curves from the ³F₄ Tm³⁺ and ⁵I₇ Ho³⁺ manifolds are plotted in a semi-log scale in Fig. 8. Shortly after the excitation pulse, a rapid rise of the Ho³⁺ emission intensity is observed and at the same time, a fast drop in the intensity of Tm³⁺ luminescence is evident. These phenomena are indicative of a direct energy transfer from Tm³⁺ (donor, D) to Ho³⁺ (acceptor, A). At longer time scales, the decay rates from the ³F₄ Tm³⁺ and ⁵I₇ Ho³⁺ manifolds become identical as a thermal equilibrium is reached between them, balancing the direct and back energy transfer.

The parameters of the bidirectional ³F₄(Tm³⁺) \leftrightarrow ⁵I₇(Ho³⁺) energy-transfer were determined using the dynamical model developed by Walsh *et al.* [50]. The luminescence decay curves were fitted using the following equations:

$$\frac{n_2(t)}{n_2(0)} = \frac{\beta}{\alpha + \beta} \exp\left(-\frac{t}{\tau}\right) + \frac{\alpha}{\alpha + \beta} \exp\left(-(\alpha + \beta)t\right), \quad (2a)$$

$$\frac{n_7(t)}{n_7(0)} = \frac{\beta}{\alpha + \beta} \exp\left(-\frac{t}{\tau}\right) - \frac{\alpha}{\alpha + \beta} \exp\left(-(\alpha + \beta)t\right), \quad (2b)$$

where t is time after a short-pulse excitation, τ is the thermal equilibrium decay time, n_2 and n_7 are the populations of the ³F₄ Tm³⁺ and ⁵I₇ Ho³⁺ manifolds, respectively (here, we assume that the luminescence intensity is proportional to electronic populations), $\alpha = P_{28}N_{Ho}$ and $\beta = P_{71}N_{Tm}$ are the transfer rates, P_{28} is a parameter of a direct non-radiative energy transfer, Tm³⁺ \rightarrow Ho³⁺, and P_{71} is a parameter of the back nonradiative energy transfer, Ho³⁺ \rightarrow Tm³⁺. The best-fit curves are shown in Fig. 8. The best-fit parameters are $\tau = 3.80 \pm 0.3$ ms, $P_{28} = 1.09 \pm 0.05 \times 10^{-22}$ cm³ μ s⁻¹ and $P_{71} = 0.42 \pm 0.05 \times 10^{-23}$ cm³ μ s⁻¹. The ratio of the energy-transfer parameters, $\Theta = P_{71}/P_{28}$, also known as the equilibrium constant, is relatively small, 0.038. This value confirms the predominantly direct Tm³⁺ \rightarrow Ho³⁺ energy transfer.

The luminescence decay curve for a singly doped 5 at% Tm:NaGd(MoO₄)₂ crystal was also studied for comparison yielding a luminescence lifetime of the ³F₄ and ³H₄ Tm³⁺ states τ_{lum} of 1.41 ms and 30 μ s, respectively. The measured ³F₄ lifetime is longer than that determined in the previous studies (1.2 ms) [34,37], as well as the previously reported radiative lifetime, calculated using the Judd-Ofelt theory (0.9 ms) [37], which is probably due to the residual effect of reabsorption (radiation

trapping). The relatively short luminescence lifetime of the ³H₄ state indicates an efficient cross-relaxation among adjacent ions contributing to enhanced pump quantum efficiency.

5.3. Low temperature spectroscopy

The polarized absorption and emission spectra corresponding to the ³H₆ \leftrightarrow ³F₄ Tm³⁺ and ⁵I₈ \leftrightarrow ⁵I₇ Ho³⁺ transitions in the studied crystal were acquired at low temperature (LT, 12 K). The results are shown in Fig. 9. The LT absorption spectra gave access to the crystal-field splitting of excited-states, and they were expressed in photon energy (in cm⁻¹). The LT emission spectra were analyzed to resolve the Stark sub-level energies of the ground-states, plotted as a function of (E_{ZPL} - photon energy), where E_{ZPL} is the zero-phonon line (ZPL) energy. Here, ZPL indicates the transitions between the lowest-lying Stark sub-levels of both multiplets.

The trivalent dopant ions in the NaGd(MoO₄)₂ crystal reside in the (Na⁺ + Gd³⁺) site of the scheelite structure with S₄ symmetry (Wyckoff: 4b). For the tetragonal crystal field, any multiplet ^{2S+1}L_J with an integer total angular momentum J of 4, 6, 7 and 8 will be split into 7, 10, 11 and 13 Stark components, respectively. In this study, we determined the crystal-field splitting of the two lowest multiplets of Tm³⁺ (³H₆ and ³F₄) and Ho³⁺ (⁵I₈ and ⁵I₇) in NaGd(MoO₄)₂, relevant for describing their laser emission properties around 2 μ m. For assignment of electronic transitions, we followed the crystal-field data reported previously for Tm³⁺ in NaLa(MoO₄)₂ [51] and Ho³⁺ in NaBi(MoO₄)₂ [27] (both crystals adopt the same tetragonal structure as NaGd(MoO₄)₂ studied in the present work). The electronic transitions for these reference crystals are marked with vertical dashes in Fig. 9.

Table 4 lists the experimental energies of Stark sub-levels of (³H₆ and ³F₄) Tm³⁺ and (⁵I₈ and ⁵I₇) Ho³⁺ multiplets in NaGd(MoO₄)₂. Fig. 10 illustrates the corresponding energy-level schemes. The ZPL transition for both ions is indicated. For Tm³⁺, it occurs at 1787.3 nm, corresponding to an energy E_{ZPL} of 5595 cm⁻¹, whereas for Ho³⁺, it is expected at 1948 nm ($E_{ZPL} = 5133$ cm⁻¹). The total Stark splitting of the ground-state amounts to $\Delta E(^3H_6) = 364$ cm⁻¹ for Tm³⁺ and $\Delta E(^5I_8) = 262$ cm⁻¹ for Ho³⁺.

6. Laser operation

The scheme of the laser cavity is depicted in Fig. 11. The rectangular laser element was cut from the annealed Tm,Ho:NaGd(MoO₄)₂ crystal for light propagation along the a -axis (a -cut) and it was 3.0 mm-thick with an aperture of 3.0(c) \times 3.0 mm². Both its input and output surfaces were polished to laser grade quality with good parallelism and antireflection (AR) coated for both the pump and laser wavelengths. The

crystal was mounted on a passively cooled Cu-holder using a silver paint for better heat management. The hemispherical cavity was formed by a plane pump mirror (PM) coated for high transmission (HT) at 0.8 μm and high reflection (HR, $R > 99.9\%$) at 1.85 – 2.3 μm , and a set of concave output couplers (OCs) having a radius of curvature (RoC) of –100 mm and a transmission at the laser wavelength T_{OC} ranging from 0.1 % to 10 %. The crystal was placed near the PM leaving only a small air gap (<1 mm) and the geometrical cavity length was about 99 mm. The pump source was a Ti:Sapphire laser (3900S, Spectra Physics) delivering up to 3.2 W at 795 nm with a nearly diffraction limited beam ($M^2 \approx 1$). The pump polarization was adjusted using an AR-coated half-wave plate to correspond to σ in the crystal. The pump beam was focused into the crystal through the PM using an AR-coated achromatic lens (focal length: $f = 75$ mm). The single-pass pump absorption efficiency under lasing conditions was weakly dependent on the output coupling, amounting to 96 %. The spectra of laser emission were measured by the same optical spectrum analyzer and a multimode ZrF₄ fiber.

The continuous-wave Tm,Ho:NaGd(MoO₄)₂ laser generated a maximum output power of 548 mW at 2045 – 2055 nm corresponding to a slope efficiency η of 32.4 % (relative to the absorbed pump power), and a laser threshold of 98 mW, see Fig. 12(a). This performance was achieved through an intermediate output coupling ($T_{OC} = 2\%$). The laser output corresponded to pure Ho³⁺ emission (the ⁵I₇ → ⁵I₈ transition). For higher output coupling, the laser slope efficiency slightly decreased (to 30.9 % for 5 % OC and then to 30.3 % for 10 % OC). On increasing the output coupling from 0.1 % to 10 %, the laser threshold progressively increased from 74 mW to 227 mW. Well above the laser threshold, the input-output dependencies were linear. Up to the absorbed pump power of about 2 W, neither signs of undesired thermal effects nor crystal fracture were observed. Further power scaling was limited by the power provided by the Ti:Sapphire laser.

Fig. 12(b) illustrates typical laser emission spectra measured for all the studied OCs well above the laser threshold. For low output coupling ($T_{OC} \leq 2\%$), the Tm,Ho-laser operated solely on the ⁵I₇ → ⁵I₈ Ho³⁺ transition without Tm³⁺ colasing. Conversely, higher output coupling ($T_{OC} \geq 5\%$) facilitated colasing at about 1.92 μm (³F₄ → ³H₆, Tm³⁺) and 2.04 μm (⁵I₇ → ⁵I₈, Ho³⁺). The laser emission in all cases was linearly polarized (π), and the polarization state was determined by the anisotropy of the gain. Regarding Ho³⁺ emission, a notable blue-shift in the emission spectra was observed with increasing the output coupling, attributed to the quasi-three-level nature of the ⁵I₇ → ⁵I₈ Ho³⁺ laser scheme involving reabsorption from the terminal laser level.

Fig. 12(c) shows the power redistribution between Ho³⁺ and Tm³⁺ emissions with increasing the absorbed pump power under high output coupling ($T_{OC} = 5\%$). First, the Ho-laser starts to operate with a threshold of 100 mW. Above 530 mW of absorbed pump power, the Tm laser emission appears in addition and the slope efficiency of the Ho-laser decreases. The power fractions of Ho³⁺ and Tm³⁺ emissions almost equalize at high pump powers.

The laser performance achieved for the Tm,Ho:NaGd(MoO₄)₂ crystal in the present work is superior to the previous results for Tm,Ho-lasers based on scheelite-type tetragonal tungstate and molybdate crystals. Han *et al.* reported on a continuous-wave Tm,Ho:NaY(WO₄)₂ laser delivering 265 mW at 2050 nm with a slope efficiency of 43 % and π -polarized emission [31]. The same authors explored a Tm,Ho:NaLu(WO₄)₂ crystal generating only 50 mW at 2060 nm for σ -polarization [31]. Dunaeva *et al.* developed a diode-pumped Tm,Ho:SrMoO₄ laser generating 1.63 mW at 2060 nm with η of 3 % and a laser threshold of 40 mW [52].

The output characteristics of Tm,Ho lasers with Ti:Sapphire laser pumping reported recently are compared in Table 5. The studied laser features a reasonably high slope efficiency (~30 %) comparable to those observed previously for state-of-the-art laser materials such as Tm³⁺, Ho³⁺-codoped KLu(WO₄)₂ and CaGdAlO₄, and benefits from a very low laser threshold (~100 mW). The latter is assigned to the relatively long thermal equilibrium luminescence lifetime, as well as low intracavity

losses in part due to the AR coatings applied on the crystal.

We further applied a fiber-coupled (fiber core diameter: 105 μm , N. A. = 0.22) AlGaAs laser diode (DILAS) emitting at 788 nm for pumping the Tm,Ho:NaGd(MoO₄)₂ crystal. The diode output was reimaged into the crystal using a pair of AR-coated achromatic lenses ($f = 40$ mm and 50 mm, respectively). The above-described laser cavity was used. The absorption efficiency of unpolarized pump was about 48 %. The output characteristics of the diode-pumped Tm,Ho-laser are presented in Fig. 13. For $T_{OC} = 5\%$, we achieved a maximum output power of 130 mW at 2048 nm with a slope efficiency of 5.4 % and a laser threshold of 233 mW. These results highlight the potential of the developed material for diode-pumped laser systems.

7. Conclusion

To conclude, Tm³⁺,Ho³⁺-codoped sodium gadolinium double molybdate (NaGd(MoO₄)₂) crystals hold promise for ultrashort pulse and broadly tunable laser sources emitting in the eye-safe spectral range, above 2 μm . The main feature of this compound belonging to the crystal family of tetragonal scheelite-type double tungstates / molybdates is a strong inhomogeneous spectral line broadening in the absorption and emission spectra stemming from the structure disorder. It mitigates the requirements for wavelength stabilization of pump sources (e.g., laser diodes), as well as enables smooth and broad gain profiles around 2 μm (gain bandwidth: 66 nm for σ -polarized light) supporting generation of sub-100 fs pulses from mode-locked lasers, as well as broadband tuning of laser emission. This compound also benefits from a strong polarization anisotropy of emission properties enabling linearly polarized laser radiation. The Tm³⁺ → Ho³⁺ energy transfer in this double molybdate crystal is very efficient, alongside a relatively long thermal equilibrium decay time leading to low laser thresholds. As compared to tetragonal double tungstates, the studied crystal features higher transition cross-sections for the dopant ions. The aforementioned appealing spectroscopic properties allowed us to demonstrate efficient and low-threshold continuous-wave laser action in Tm:Ho:NaGd(MoO₄)₂.

CRedit authorship contribution statement

Camy Patrice: Supervision, Funding acquisition. **Slimi Sami:** Investigation. **Braud Alain:** Methodology. **Aguiló Magdalena:** Supervision, Funding acquisition. **Pan Zhongben:** Writing – original draft, Investigation. **Solé Rosa Maria:** Methodology. **Zin Elabedine Ghassen:** Writing – original draft, Investigation. **Chen Weidong:** Conceptualization. **Chu Hongwei:** Investigation. **Díaz Francesc:** Supervision, Funding acquisition. **Loiko Pavel:** Writing – original draft, Methodology, Conceptualization. **Petrov Valentin:** Writing – review & editing, Methodology. **Eremeev Kirill:** Investigation. **Griebner Uwe:** Writing – review & editing. **Li Dechun:** Investigation. **Pavlov Sergei:** Investigation. **Mateos Xavier:** Writing – review & editing, Supervision, Funding acquisition. **Subbotin Kirill:** Resources, Investigation.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Data availability

Data will be made available on request.

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