



Research Papers

Spectroscopy of Yb³⁺ ions in multisite MgMoO₄ crystal

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ABSTRACT

Yb³⁺-doped magnesium monomolybdate crystal (Yb:MgMoO₄) is grown by the Czochralski method. The crystal structure, thermal expansion, vibronic properties, polarized room- and low-temperature spectroscopy of this new compound were studied. According to the Rietveld refinement, Yb:MgMoO₄ belongs to the monoclinic class (sp. gr. C2/m, lattice constants: $a = 10.276(9)\text{Å}$, $b = 9.289(6)\text{Å}$, $c = 7.027(6)\text{Å}$ and $\beta = 106.903(8)^\circ$). The highest-energy Raman mode of Yb:MgMoO₄ is found at 969 cm⁻¹ and assigned to symmetric stretching vibrations of [MoO₄]²⁻ tetrahedra. Yb³⁺ ions in MgMoO₄ exhibit intense and broad absorption at 978 nm, relatively broad and smooth emission bands above 1 μm (stimulated-emission cross-section: $\sigma_{SE} = 0.76 \times 10^{-20} \text{ cm}^2$ at 1031 nm for light polarization $E \parallel c$) and a long luminescence lifetime (0.63 ms). The low-temperature (12 K) spectroscopic studies evidenced a multi-site behavior for Yb³⁺ ions in MgMoO₄ revealing at least 8 non-equivalent optical centers which are assigned to different charge compensation mechanisms. This induces a strong inhomogeneous spectral line broadening. For envisioned laser applications, the problem of low Yb³⁺ segregation coefficient ($K_{Yb} \sim 0.029$) needs to be solved.

1. Introduction

The development of ultrashort-pulse lasers around 1 μm based on Ytterbium (Yb³⁺) ions requires the identification of promising gain media featuring both broadband emission properties and good thermal behavior. The spectral line broadening could be reached in crystals with a structure and / or compositional disorder however often at the expense of deteriorated thermal conductivity. It is also desired that laser gain materials provide linearly polarized emission to avoid the detrimental depolarization losses. The latter property could be attained for low symmetry matrices, e.g., monoclinic crystals. Indeed, Yb³⁺-doped ordered monoclinic (sp. gr. C2/c) double tungstate crystals KRE(WO₄)₂ (where RE stands for Y, Lu or Gd), are widely recognized for femtosecond mode-locked lasers and amplifiers at 1 μm [1,2]. Despite the appealing spectroscopic properties of these compounds, their main

drawback stems from the relatively low thermal conductivity ($\sim 3 \text{ Wm}^{-1}\text{K}^{-1}$) [3] and impossibility to grow single crystals directly from melt [2].

Recently, ordered monoclinic (sp. gr. P2/c) magnesium and zinc monotungstate crystals, MgWO₄ [4–7] and ZnWO₄ [8–10], have been identified as interesting laser host matrices for doping with rare-earth ions (RE³⁺), in particular, for broadly tunable and femtosecond mode-locked solid-state lasers [11,12]. These compounds combine i) rather good thermal conductivity [13]; ii) high birefringence [8]; iii) intense absorption and emission bands of RE³⁺ ions for polarized light [4,8]; iv) strong crystal fields resulting in large Stark splitting of RE³⁺ manifolds [14]; and v) efficient electron-phonon interaction [14] and Raman activity [15,16]. In particular, the room-temperature thermal conductivity of MgWO₄ amounts to 8.7 Wm⁻¹K⁻¹ [13]. Loiko et al. reported on a diode-pumped Yb:MgWO₄ laser delivering 18.2 W at ~ 1056

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nm with a slope efficiency of 89 % approaching the Stokes limit and a linearly polarized output [4]. Lin et al. developed an Yb:MgWO₄ oscillator mode-locked by a semiconductor saturable absorber mirror generating 125 fs pulses at 1065 nm at a repetition rate of 117 MHz and continuous wavelength tuning over 1019 – 1057 nm [11]. Recently, efficient laser operation of an Yb³⁺,Li⁺ codoped ZnWO₄ crystal was also obtained: a continuous-wave Yb-laser generated 2.41 W at 1055–1058 nm with a slope efficiency of 76.4 % [17].

The origin of broadband emission properties of RE³⁺ dopant ions in M²⁺WO₄ crystals (M = Mg, Zn) remained unclear: the authors claimed that it may originate from the complex doping mechanism involving univalent alkali-metal charge compensators, i.e., Na⁺ ions entering from the flux used for the growth of MgWO₄ [7] or Li⁺ ions introduced intentionally in the case of ZnWO₄ [8,18]. In our recent study of Yb³⁺-doped zinc monotonogtate, we showed for the first time that even despite the ordered nature of this crystal (Zn²⁺ ions occupy a single 2e Wyckoff site with C₂ symmetry and VI-fold oxygen coordination), it presents a *multi-site* behavior as evidenced by the splitting of the Yb³⁺ absorption and emission zero-phonon lines observed at cryogenic temperatures [17]. This unexpected behavior was assigned to a variety of Yb³⁺ → Zn²⁺ substitutions and the associated local charge compensation, e.g., involving vacancies or univalent alkali metal cations. Multi-site behavior has been recently evidenced for Er³⁺ ions in ZnWO₄ by laser site selective spectroscopy [19].

MgWO₄ melts incongruently, and, consequently, it is grown from the flux [7,20] resulting in relatively small crystals and tolerated optical quality. There exists its monomolybdate counterpart, MgMoO₄ [21–23] which melts congruently at 1322 °C [24] and can be grown by the standard Czochralski (Cz) method (from the melt) [25]. This compound crystallizes in the monoclinic class (sp. gr. C₂/m, unit-cell parameters: *a* = 10.273(3)Å, *b* = 9.288(3)Å, *c* = 7.025(2)Å and β = 106.96° [26]) and exhibits a structure disorder due to existence of two non-equivalent sites for Mg²⁺ ions, labelled Mg1 (Wyckoff: 4g_i) and Mg2 (Wyckoff: 4i). Upon doping with RE³⁺ ions, it was suggested that they tend to replace the Mg²⁺ cations predominantly in one type of sites (Mg1) owing to the stronger distortion of the [Mg1O₆] octahedra and longer Mg1 – O interatomic distances [27]. However, this crystallographic behavior is expected to further promote the diversity of optical centers and induce additional inhomogeneous spectral line broadening, an effect being favorable for generation of ultrashort pulses.

MgMoO₄ was studied for scintillators as it exhibits broadband luminescence under irradiation by electron beam or α-particles [24,28]. Recently, this molybdate also attracted attention as a host matrix for doping with both transition-metal ions, notably Cr³⁺ [22] and rare-earth ions [23,29]. Recently, we reported on the growth, thermal and spectroscopic properties of Tm³⁺:MgMoO₄ for emission at ~2 μm [23]. However, due to the complexity of crystal-field splitting of Tm³⁺ multiples, we were unable to conclude about the possible multi-site behavior of this compound.

In the present work, we report on the Czochralski growth, crystal structure refinement, Raman properties, and polarized room- and low-temperature spectroscopy of an Yb³⁺-doped MgMoO₄ crystal with the goal of exploring promising structurally disordered compounds with broadband emission properties which are of interest for ultrafast lasers around 1 μm. Moreover, we present the first experimental evidence of multi-site behavior of RE³⁺-doped magnesium monomolybdate.

2. Crystal growth and characterization techniques

2.1. Crystal growth

The starting materials for the growth of Yb-doped magnesium monomolybdate comprised MgO (purity: 4 N, Omskreactiv, Russia), MoO₃ (4 N, Lankhit Ltd., Russia), and Yb₂O₃ (YbO-D, 4 N, Industry Branch Standard 48–206–81, p/b M-5649 USSR). The Yb³⁺ concentration in the starting melt for Yb:MgMoO₄ growth was set to 7 at. %

(considering the substitution of Mg²⁺ ions). First, the starting materials were thoroughly mixed and further calcined at a temperature of 700 °C during 5 h for solid-phase synthesis of the magnesium monomolybdate phase. Since the saturated vapor pressure of unbound MoO₃ at the melting point of MgMoO₄ exceeds 1 atm, a significant part of MoO₃ could evaporate from the melt if no calcining is applied prior to melting.

The single-crystal of Yb:MgMoO₄ was grown by the Cz method using a “Kristall-2” growth station (USSR) in air atmosphere. A Pt/Rh crucible having a diameter of 30 mm and a height of 30 mm was used. The seed for the crystal growth was cut from undoped MgMoO₄ along the crystallographic axis [001]. During the growth, the pulling rate was 1 mm/h and the rotation speed was 6 revolutions per minute (r.p.m.). After completing the growth, the crystal was removed from the melt surface and slowly cooled down to room temperature (RT) using a rate of 8 °C/h with the goal of preventing the risk of cracking. The as-grown crystal was further annealed in air at 800 °C for 2 weeks to remove the stresses formed during the growth and improve its transparency.

2.2. Experimental

The actual Yb³⁺ concentration in the crystal was measured by the energy-dispersive X-ray spectroscopy (EDXS) (AZtecENERGY analytical systems; Oxford Instruments; JSM5910-LV, JEOL), at an accelerating voltage of 20 kV. The polished surfaces of the sample for the EDXS analysis was coated with a conducting carbon film. The analytical errors did not exceed 0.05, 0.12 and 0.10 wt % for Mg, Mo and Yb, respectively.

The phase purity of the investigated crystal was determined using X-ray diffraction (XRD) measurements in the θ - θ Bragg-Brentano configuration. The measurements were performed using a Bruker-AXS D8-Advance diffractometer and CuK_α radiation (λ = 0.15406 nm) at 40 kV and 30 mA, under room temperature conditions (293 K). The step scan rate was 3 s per step, with a step size of 0.05° over a 2θ range of 10 to 70 °. The XRD patterns of the sample were compared with the reference standard; the XRD data for undoped MgMoO₄ were represented by PDF card No. 72–2153.

High quality data are essential for crystal structure refinement, so we have performed additional XRD measurements using the variable-counting-time (VCT) technique to collect the data used for the Rietveld refinement. These measurements were made using the same Bruker-AXS D8-Advance diffractometer used in previous measurements, with a vertical θ - θ goniometer, incident- and diffracted-beam Soller slits of 2.5°, a fixed 0.5° receiving slit and an automatic air scattering knife on the sample surface. The angular 2θ range was between 12 and 90°. The data were collected with an angular step of 0.02° with a variable counting time from 1 to 14 s. The CuK_α radiation was obtained from a copper X-ray tube operated at 40 kV and 40 mA. The diffracted X-rays were detected with a position sensitive detector (PSD) LynxEye-XE-T with an opening angle of 2.94°

The X-ray diffraction pattern of the sample measured using the VCT technique over a 2θ diffraction angle range of 12–0° was analyzed and refined using the Rietveld method with the Topas V6 software. For the Rietveld refinement, the crystal structure of undoped MgMoO₄ (represented by PDF card No. 72-2153) was used as the starting model.

The linear thermal expansion tensor of Yb:MgMoO₄ was calculated taking into account the evolution of its unit-cell parameters with temperature. For this study, powder X-ray diffractograms from room temperature to 600 °C were recorded in the 2θ range from 5 to 80° with a temperature step of 50 °C, using the same diffractometer than in the previous measurements, but equipped with a modular temperature chamber (MCT-HIGHTEMP, Bruker). The heating rate was 10 °C/min and a delay time of 300 s was applied before doing the measurements. The diffraction data were recorded with an angular step size of 0.02° and a step time of 1 s. The unit-cell parameters were refined using the Pawley method with Topas V6 software.

For spectroscopic studies, the crystals were oriented with respect to

the crystallographic axes by means of single-crystal diffraction. The crystal orientation was performed using a Bruker-AXS D8-Discover diffractometer, with a parallel incident beam, a vertical $\theta - \theta$ goniometer and a 3D motorized stage. A general area diffraction system (HI STAR detector) located at 15 cm from the sample was used. The measurements were carried out in the 2θ range from 8 to 92° .

The polarized Raman spectra were acquired using a Renishaw inVia confocal Raman microscope employing an $\times 50$ Leica objective, an edge filter, and an Ar⁺ ion laser ($\lambda_{\text{exc}} = 514$ nm).

The absorption spectra for polarized light were recorded using a spectrophotometer (Lambda 1050, Perkin Elmer) and a Glan-Taylor polarizer. The luminescence of Yb³⁺ ions was excited by a tunable Ti:Sapphire laser and detected using a CaF₂ lens, a ZrF₄ fiber and an optical spectrum analyzer (OSA, Ando AQ-6315A). Low-temperature (LT, 12 K) spectroscopic studies were conducted using a APD DE-202 closed-cycle cryo-cooler equipped with an APD HC 2 Helium vacuum cryo-compressor and a Laceshore 330 temperature controller.

To study luminescence dynamics, we used a ns optical parametric oscillator (Horizon, Continuum), a 1/4 m monochromator (Oriel 77200), a fast InGaAs detector and an 8 GHz digital oscilloscope (DSA70804B, Tektronix).

3. Results and discussion

3.1. Results on the crystal growth

Fig. 1 presents a photograph of an as-grown Yb:MgMoO₄ single-crystal. The crystal boule has an amber-like coloration. The actual Yb³⁺ doping level in the crystal was determined to be 0.2 at. % (according to Yb³⁺ → Mg²⁺ substitution), and the ion density N_{Yb} was calculated to be 0.63×10^{19} at/cm³. The segregation coefficient of Yb³⁺ ions, $K_{\text{Yb}} = C_{\text{crystal}}/C_{\text{melt}}$, where C_{crystal} is the actual doping level in the crystal and C_{melt} is its nominal concentration in the melt, was estimated to be only ~ 0.029 .

The entrance of trivalent rare-earth ions into the lattices of Mg²⁺-based crystals is often very difficult. Indeed, for the Mg₂SiO₄ crystal, the segregation coefficient of Yb³⁺ ions was found to be as low as 0.001 [30]. Comparing the isostructural host crystals MgWO₄ and ZnWO₄, the Ho³⁺ distribution coefficient between the Ho:MgWO₄ crystal and the flux was found to be 0.086 [6], whereas the Yb³⁺ distribution coefficient between the related magnesium free Yb:ZnWO₄ crystal and the melt could reach the value of 0.5 [18]. In our previous studies focusing on magnesium monomolybdate doped with Tm³⁺ and Nd³⁺ ions, the corresponding segregation coefficients amounted to $K_{\text{Tm}} = 0.02$ [23], and $K_{\text{Nd}} < 0.01$ [25]. The reason for such behavior is not clear so far. It can be caused by substantial difference in the chemical nature between Mg²⁺ and trivalent lanthanides.

3.2. Crystal structure

The measured X-ray powder diffraction (XRD) pattern of the 0.2 at. % Yb:MgMoO₄ crystal together with the results of the Rietveld



Fig. 1. A photograph showing an as-grown 0.2 at. % Yb³⁺-doped MgMoO₄ crystal (actual doping). The growth direction is along the [001] axis.

refinement are shown in Fig. 2. The results of the analysis indicate that all the observed diffraction peaks are consistent with the reference standard and no additional peaks indicative of impurities are observed, demonstrating the phase purity of the grown crystal. The refinement results show that Yb³⁺:MgMoO₄ crystallizes in the monoclinic class. This compound belongs to the space group *C2/m* (point group *2/m*) with a general multiplicity *Z* of 2. The refinement quality is evaluated using several reliability factors, including $R_p = 8.8\%$, $R_{wp} = 11.71\%$, and $R_{exp} = 7.86\%$, with a reduced chi-squared value of $\chi^2 = (R_{wp}/R_{exp})^2 = 2.22$. The lattice constants of the Yb³⁺:MgMoO₄ crystal are determined to be $a = 10.276(9)\text{\AA}$, $b = 9.289(6)\text{\AA}$, $c = 7.027(6)\text{\AA}$ and $\beta = 106.903(8)^\circ$, the unit cell volume $V = 641.932(7)\text{\AA}^3$. The determined lattice constants are slightly exceeding those reported previously for undoped MgMoO₄ ($a = 10.273\text{\AA}$, $b = 9.288\text{\AA}$, $c = 7.025\text{\AA}$ and $\beta = 106.96^\circ$, $V = 641.14\text{\AA}^3$), and are close to those for the Tm³⁺:MgMoO₄ crystal [23]. This difference between undoped and RE³⁺-doped MgMoO₄ crystals occurs, obviously, due to the difference in the ionic radii between the host-forming cations, Mg²⁺ (0.72 Å) and the dopant ions, Yb³⁺ (0.868 Å), or Tm³⁺ (0.88 Å), respectively.

The calculated pattern demonstrates good agreement with the experimental data, as shown in Fig. 2. The calculated density of the crystal is $\rho_{\text{calc}} = 3.816\text{ g/cm}^3$. It is lower than that of Tm:MgMoO₄ (3.888 g/cm³) [23]. The obtained fractional atomic coordinates (*x*, *y*, *z*), site occupancy factors (O.F.), and isotropic displacement parameters (*B*_{iso}) are listed in Table 1.

The crystal structure of Yb:MgMoO₄ is depicted in Fig. 3(a,b), which presents the projections of the structure along the *b*- and *c*-axes, based on the atomic coordinates determined through Rietveld refinement. The structure of Yb:MgMoO₄ can be described as a two-layered composition of MgO₆ polyhedra layers separated by MoO₄ polyhedra layers, as shown in Fig. 3(a). The distinct structural feature of Yb:MgMoO₄ is the criss-cross unit site, depicted in Fig. 4, in which the coordination environment of Mg(1) and Mg(2) ions is shown. The criss-cross unit site is formed by the interconnected MgO₆ polyhedra through common edges. The dopant Yb³⁺ ions preferentially replace the host-forming cations Mg²⁺ at one of the lattice sites, namely the Mg(1) site [31]. The structures of Yb³⁺ and Tm³⁺ doped MgWO₄ crystals are similar.

3.3. Tensor of linear thermal expansion

The tensor of linear thermal expansion for the Yb:MgMoO₄ crystal in the crystallophysical system (with *X*₁ axis parallel to *a*-axis, *X*₂ parallel to *b*-axis and *X*₃ parallel to *c**-axis) was characterized by monitoring the change of the unit-cell parameters with temperature. The tensor coefficients were calculated from the slope of the linear fits of the evolution of the normalized unit-cell parameters with temperature, see Fig. 5, according to the expression: $\alpha = (1/L_{30}^\circ\text{C})(\Delta L/\Delta T)$, where $L = a, b, c$ and $c \times \cos(\beta - 90^\circ)$.

Thus, the linear thermal expansion tensor in the crystallophysical system is:

$$(a_{ij}) = \begin{pmatrix} 7.365 & 0 & 0.226 \\ 0 & 14.00 & 0 \\ 0.226 & 0 & 3.869 \end{pmatrix} \times 10^{-6} \text{ K}^{-1}. \quad (1)$$

By diagonalizing this tensor, the thermal expansion tensor in the eigen-frame is:

$$(a'_{ij}) = \begin{pmatrix} 7.380 & 0 & 0 \\ 0 & 14.00 & 0 \\ 0 & 0 & 3.854 \end{pmatrix} \times 10^{-6} \text{ K}^{-1}, \quad (2)$$

with the principal axis *X*_{1'} found at 3.69° from the *a*-axis (clockwise), *X*_{2'} being parallel to the *b*-axis and *X*_{3'} found at 13.21° from the *c*-axis (anticlockwise). In this setting, the *b*-axis is pointing towards the observer.

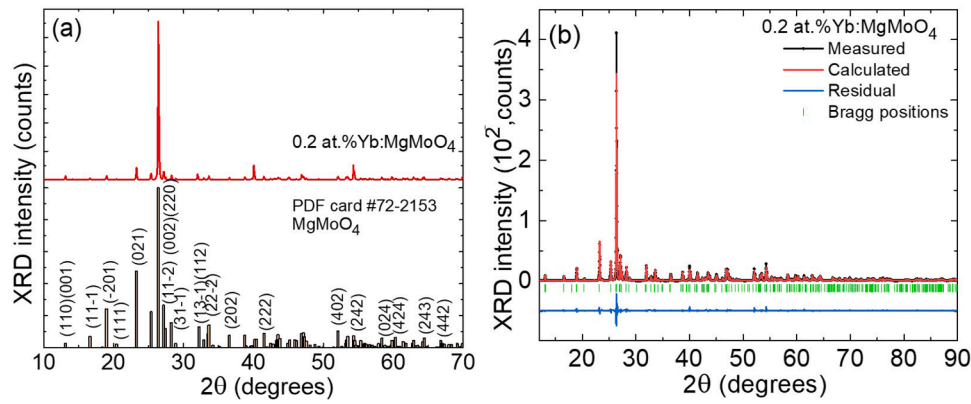


Fig. 2. X-ray powder diffraction (XRD) pattern for the 0.2 at. % Yb:MgMoO₄ crystal: (a) the measured XRD pattern, for comparison, the theoretical pattern for undoped magnesium monomolybdate (ICSD card No. 72-2153 [26]) is shown by vertical bars, *numbers* denote the Miller's indices (*hkl*) for the most intense reflections; (b) output of the Rietveld structure refinement: the measured (*black*), calculated (*red*) and residual (*blue*) profiles, *vertical dashes* mark the Bragg positions.

Table 1

Atomic coordinates, sites, occupancy factors (O.F.) and isotropic displacement parameters $B_{iso}(\text{Å}^2)$ for 0.2 at. % Yb:MgMoO₄ crystal.

| Atoms | x/a | y/b | z/c | O.F. | $B_{iso}, \text{Å}^2$ | Wyckoff | Symmetry |
|-------|-----------|-----------|-----------|-------|-----------------------|---------|----------|
| Mg1 | 1/2 | 0.1781 | 0 | 0.998 | 0.989(7) | 4g | 2 |
| Yb | 1/2 | 0.1781 | 0 | 0.002 | 0.989(7) | 4g | 2 |
| Mg2 | 0.7973 | 1/2 | 0.6396 | 1 | 0.290(5) | 4i | <i>m</i> |
| Mo1 | 1/2 | 0.2517(3) | 1/2 | 1 | 0.962(2) | 4h | 2 |
| Mo2 | 0.7290(5) | 1/2 | 0.0956 | 1 | 0.947(3) | 4i | <i>m</i> |
| O1 | 0.5420(7) | 0.1538(4) | 0.3010(6) | 1 | 0.284(3) | 8j | 1 |
| O2 | 0.3613(2) | 0.3560(6) | 0.3917(8) | 1 | 1.042 | 8j | 1 |
| O3 | 0.8563(8) | 1/2 | 0.9609(4) | 1 | 1.089 | 4i | <i>m</i> |
| O4 | 0.6336(8) | 0.3449(4) | 0.0252(9) | 1 | 0.507(1) | 8j | 1 |
| O5 | 0.2961(2) | 0 | 0.3554(6) | 1 | 1.943 | 4i | <i>m</i> |

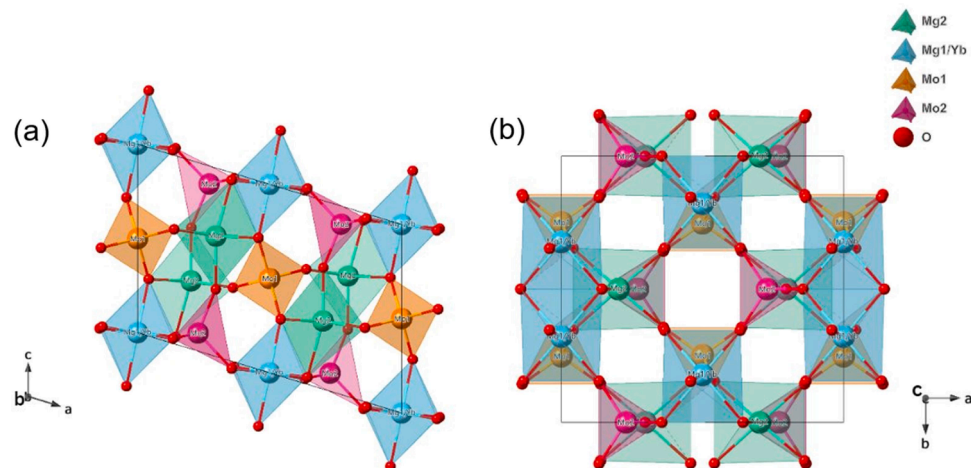


Fig. 3. (a,b) The structure of 0.2 at. % Yb:MgMoO₄: (a) projection in the *a-c* plane; (b) projection in the *a-b* plane; *black lines* – unit-cell.

3.4. Raman spectra

Yb:MgMoO₄ crystallizes in a monoclinic structure, space group $C2/m$, with 8 molecules per unit cell. The factor group theory analysis predicts that the optical modes are distributed among the irreducible representations of the factor group C_{2h} at the center of the Brillouin zone $\Gamma(\mathbf{k} = 0)$ as following: $19A_g + 17B_g + 14A_u + 19B_u$ [32,33]. According to the selection rules, there are 36 Raman-active modes ($19A_g + 17B_g$) and 33 IR-active ones ($14A_u + 19B_u$). Among the Raman-active modes, 18 ones ($11A_g + 7B_g$) are the internal ones, corresponding to the vibrations of the $[\text{MoO}_4]^{2-}$ tetrahedra: symmetric stretching ($\nu_1, 2A_g$), anti-symmetric stretching ($\nu_3, 3A_g + 3B_g$), symmetric bending ($\nu_2, 3A_g +$

$1B_g$), and anti-symmetric stretching ($\nu_4, 3A_g + 3B_g$).

We have measured the polarized Raman spectra for three orientations of Yb:MgMoO₄ when cutting it along the *a*-, *b*- and *c*-axes, as presented in Fig. 6. For labeling the geometry, we employed Porto's notations, $m(nl)k$, where *m* and *k* stand for the direction of propagation of the excitation and scattered light, respectively, and *n* and *l* are the corresponding polarization states. One can see that the Raman spectra of Yb:MgMoO₄ feature intense peaks located within two ranges of distinct wavenumbers, at 115 – 450 cm⁻¹ and 750 – 1050 cm⁻¹, which is typical for simple and complex tungstate and molybdate crystals [34-36]. Within the low-wavenumber range, the bands arise from the lattice vibrations and some bending modes (ν_2 and ν_4) mixed with the former

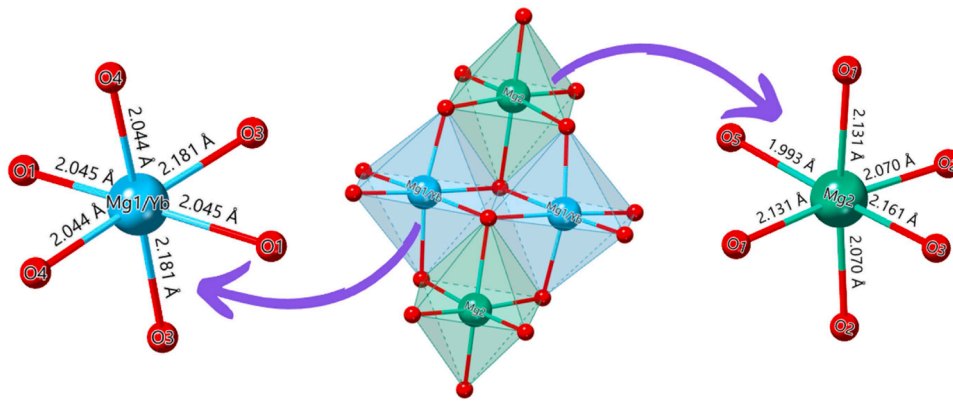


Fig. 4. The criss-cross structural unit in Yb:MgMoO₄ and the anionic coordination environment of Mg(1)|Yb and Mg(2) ions.



Fig. 5. Thermal evolution of the normalized unit-cell parameters of Yb:MgMoO₄, with $L = a, b, c$ and $c \times \cos(\beta - 90^\circ)$.

ones. The Raman peaks at high wavenumbers are mainly due to the stretching (ν_1 and ν_3) vibrations of the [MoO₄]²⁻ tetrahedra. The mode at 957 cm⁻¹ dominates in the spectra while the highest phonon energy mode is found at 969 cm⁻¹. These peaks are assigned to symmetric stretching ($\nu_1, 2A_g$) vibrations. The Raman peak linewidth (FWHM) at 957 cm⁻¹ reaches ~ 7 cm⁻¹. The peak positions and linewidths are consistent with those for Tm:MgMoO₄ [23]. The Raman spectrum of undoped MgMoO₄ contains the same peaks in the range of 115–450 cm⁻¹, except for the fact that the peak at ~ 325 cm⁻¹ is split into three components [32].

3.5. Optical spectroscopy of Yb³⁺ ions

The unpolarized overview absorption spectrum of the 0.2 at. % Yb:MgMoO₄ crystal is shown in Fig. 7(a). One can note a broad and intense absorption band with a maximum at ~ 500 nm which is causing the amber-like coloration of the crystal. It has been shown earlier [23] that the corresponding color centers are most likely related to molybdenum vacancies. This conclusion was made from the fact that the coloration of MgMoO₄ crystals almost completely disappears after prolonged annealing in air atmosphere containing pronounced amounts of MoO₃

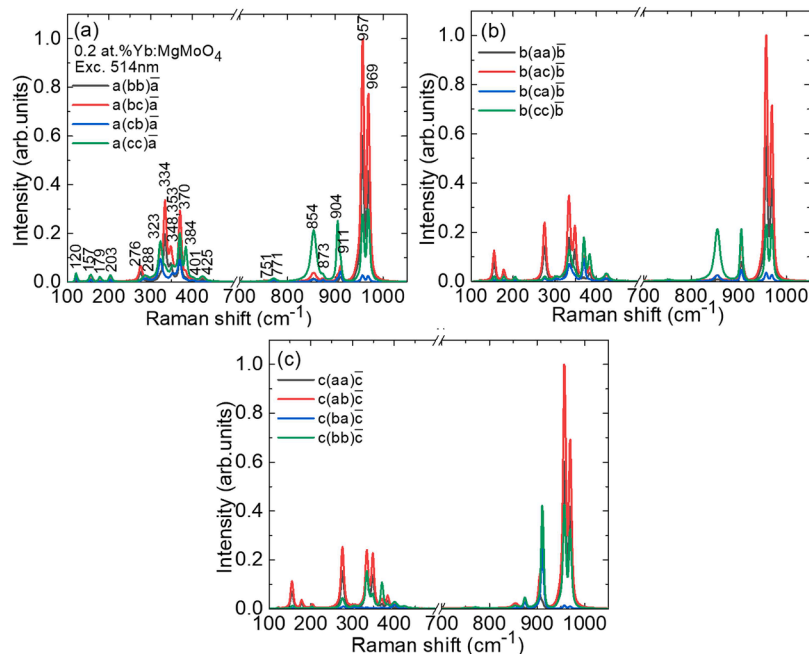


Fig. 6. Polarized Raman spectra of (a) *a*-cut, (b) *b*-cut and (c) *c*-cut 0.2 at. % Yb:MgMoO₄ crystals, $\lambda_{\text{exc}} = 514$ nm, $T = 300$ K. Numbers in (a) - peak frequencies in cm⁻¹.

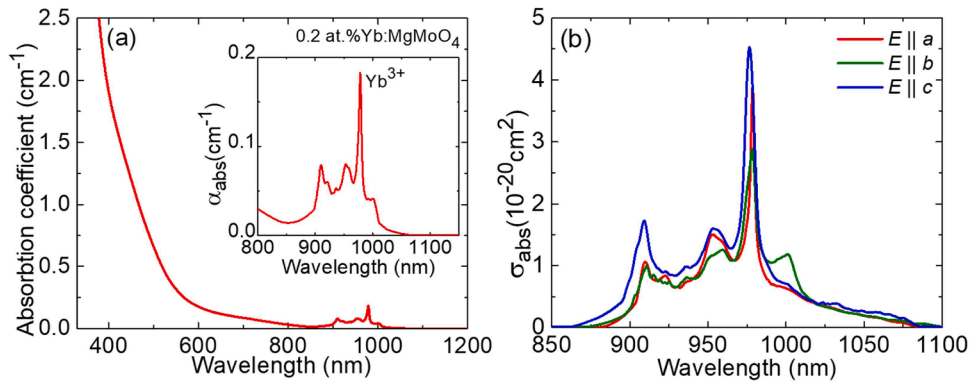


Fig. 7. Optical absorption of the 0.2 at. % Yb: MgMoO₄ crystal at room temperature: (a) an overview unpolarized absorption spectrum, *inset* – absorption of Yb³⁺ ions around 1 μm; (b) absorption cross-sections, σ_{abs} , of Yb³⁺ ions for light polarizations $E \parallel a, b, c$.

vapors, whereas annealing at similar temperatures and durations, but only in air atmosphere without MoO₃ does not change the crystal color.

The absorption near 1 μm is due to the Yb³⁺ ions (the ²F_{7/2} → ²F_{5/2} electronic transition). The RT polarized absorption spectra near 1 μm are shown in Fig. 7(b). They span from ~0.85 to 1.10 μm with an intense distinct peak at ~975 nm, corresponding, evidently, to the transition between the lowest Stark sub-levels of the ground state ²F_{7/2} and the excited state ²F_{5/2} of Yb³⁺ ions (the so-called zero-phonon line, ZPL).

For the ²F_{7/2} → ²F_{5/2} Yb³⁺ transition, at the ZPL, the peak absorption cross-sections σ_{abs} amount to $3.90 \times 10^{-20} \text{ cm}^2$ at 978.6 nm for light polarization $E \parallel a$ (the absorption linewidth, full width at half maximum, FWHM is 4.5 nm), $2.89 \times 10^{-20} \text{ cm}^2$ at 978.4 nm for $E \parallel b$ (FWHM = 11.3 nm) and $4.53 \times 10^{-20} \text{ cm}^2$ at 976.5 nm for $E \parallel c$ (FWHM = 8.7 nm). Note that the absorption ZPL of Yb³⁺ ions in this crystal has a multi-peak structure even at RT and it is relatively broad. These peak σ_{abs} values are comparable with those for the Yb:MgWO₄ crystal ($\sigma_{\text{abs}} = 6.16 \times 10^{-20} \text{ cm}^2$ at 974.0 nm corresponding to an absorption bandwidth of 5.6 nm for light polarization $E \parallel N_g, N_g \hat{a} = 36.4^\circ$) [24]. For Yb,Li:ZnWO₄, lower values were found: $\sigma_{\text{abs}} = 2.40 \times 10^{-20} \text{ cm}^2$ at 972.7 nm corresponding to a broader absorption bandwidth of 8.3 nm also for light polarization $E \parallel N_g, N_g \hat{c} = 12.4^\circ$) [8]. In [37], the maximum absorption cross-section for Yb³⁺ ions in Yb³⁺,K⁺:ZnWO₄ is given without specifying the polarization, $\sigma_{\text{abs}} = 2.6 \times 10^{-20} \text{ cm}^2$ at ~972 nm.

The-stimulated emission (SE) cross-section, σ_{SE} , spectra corresponding to the ²F_{5/2} → ²F_{7/2} transition of Yb³⁺ ions in MgMoO₄ were calculated by means of the Füchtbauer-Ladenburg (F-L) formula based on the measured polarized luminescence spectra:

$$\sigma_{\text{SE}}^i(\lambda) = \frac{\lambda^5}{8\pi \langle n \rangle^2 \tau_{\text{rad}} c} \frac{W_i(\lambda)}{1/3 \sum_{j=a,b,c} \int \lambda W_j(\lambda) d\lambda} \quad (3)$$

Here, λ denotes the wavelength of light, $\langle n \rangle$ represents the average refractive index of the host matrix, c is the speed of light, τ_{rad} stands for the radiative lifetime of the ²F_{5/2} Yb³⁺ manifold, $W_i(\lambda)$ signifies the luminescence spectrum which was calibrated accounting for the spectral response of the optical spectrum analyser, and $i = a, b, c$ indicates the principal light polarization. The factor of 1/3 stands for averaging over three principal polarizations of light.

The luminescence of Yb³⁺ ions in MgMoO₄ spans from 925 nm to 1125 nm. This crystal features a relatively strong polarization anisotropy of emission properties which underlines the possibility to obtain linearly polarized laser emission. In Fig. 8 is the computed σ_{SE} spectrum for the Yb:MgMoO₄ crystal. Yb³⁺ ions feature reabsorption at the laser wavelengths (a quasi-three-level laser scheme) and laser emission usually occurs at the wavelengths well above the ZPL transition. In the spectral range where laser operation is expected, the highest σ_{SE} is $0.76 \times 10^{-20} \text{ cm}^2$ at 1031 nm and $0.46 \times 10^{-20} \text{ cm}^2$ at 1067 nm for light polarization $E \parallel c$.

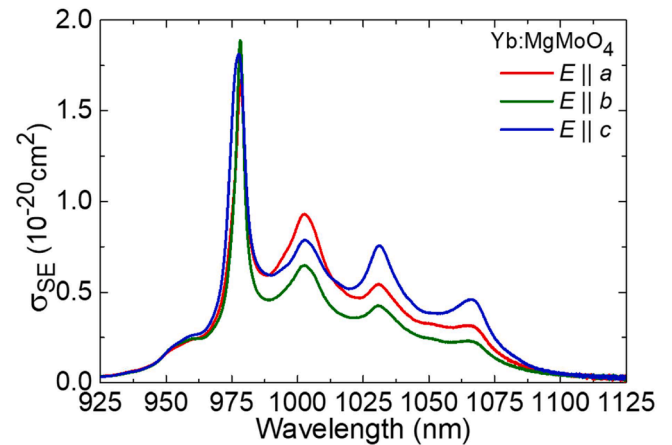


Fig. 8. Stimulated-emission cross-section, σ_{SE} , spectra of Yb³⁺ ions in MgMoO₄ for light polarizations $E \parallel a, b, c$.

Low-temperature (LT) absorption spectra were measured for light polarizations $E \parallel b$ and $E \parallel c$, see Fig. 9. The spectra indicate that, at 12 K, the above-mentioned absorption zero-phonon line splits into at least eight distinct components in the range of 972–980 nm. This may be evidence of the presence of Yb³⁺ ions in several structurally non-equivalent sites in the MgMoO₄ crystal. These sites can be regular magnesium positions Mg1 and Mg2 of the structure (see Section 3.2), as

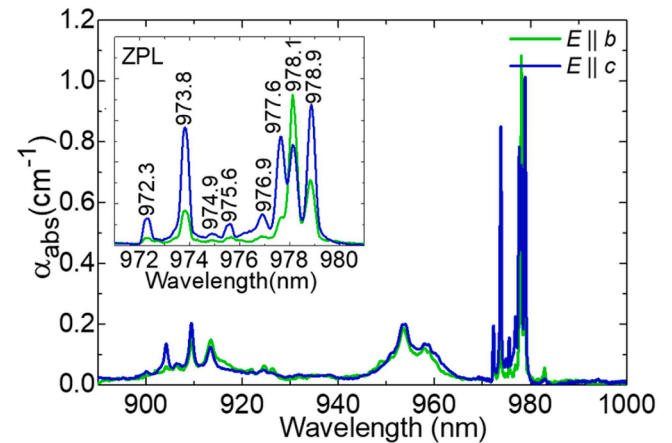


Fig. 9. LT (12 K) absorption spectra of Yb³⁺ ions in the 0.2 at. % Yb:MgMoO₄ crystal for light polarizations $E \parallel b$ and $E \parallel c$. *Inset* presents the structure of the ZPL in absorption, *numbers* – peak wavelengths in nm.

well as disturbed sites, correspond to associates of ytterbium with different kinds of charge compensators (cationic vacancies or accidental impurities, such as alkali ions, iron, silicon, etc.).

The low-temperature absorption spectra of Yb:Li:ZnWO₄ [8] do not indicate any splitting of the zero-phonon line. Simultaneously, this crystal features much higher segregation coefficient for Yb³⁺ ions, namely 0.25–0.5 depending on Li⁺ concentration [18]. Thus, the substitution Mg²⁺ → Yb³⁺ is a less favorable process than the substitution Zn²⁺ → Yb³⁺, and one can expect the formation of various point defect associates, in contrast to the case of the Zn²⁺ → Yb³⁺ substitution, despite the fact that the ionic radii of Mg²⁺ and Zn²⁺ (and the average Mg-O and Zn-O interatomic distances) in the octahedral oxygen coordination are relatively close for the MgMoO₄ and ZnWO₄ crystals, see Table 2.

The LT (12 K) luminescence spectra of Yb:MgMoO₄ were also measured for light polarizations $E \parallel b$ and $E \parallel c$ (under non-selective excitation), see Fig. 10. Similar to the absorption spectra, one can also see eight sharp components of the split zero-phonon line of ²F_{5/2} → ²F_{7/2} transition of Yb³⁺ ions in the range of 972–980 nm. The shapes and the peak positions of the zero-phonon absorption and emission bands are consistent with each other except of the fact that the luminescence of the sharp lines at 978.1 nm and 978.9 nm is strongly suppressed by the reabsorption effect, or the used non-selective excitation wavelength poorly overlaps with the narrow absorption lines of the corresponding Yb³⁺ centers. Besides that, weaker and strongly split emission bands spanning from 1 μm to 1.08 μm corresponding to transitions to other (higher lying) Stark components of the ²F_{7/2} ground state for various Yb³⁺ optical centers are visible.

We have also performed site-selective luminescence studies at 12 K, upon narrow-band excitation at different wavelengths (904.4, 909.6, 917.8, 933, and 976.6 nm). According to the results of these studies shown in Fig. 11, the groups of bands corresponding to Yb³⁺ centers located in at least five kinds of different, structurally non-equivalent crystallographic sites can be distinguished. The first one is efficiently excited at λ_{exc} = 904.4 nm (the transition from the lowest Stark component of the ground state to the highest component of the excited state) revealing a zero-phonon line at the wavelength of 972.5 nm. The second kind of Yb³⁺ centers is almost exclusively excited at λ_{exc} = 909.6 nm, whereas its emission ZPL appears at 973.7 nm. Three different kinds of Yb³⁺ centers were excited under the excitation at 917.8, 933.0 and 976.6 nm (the emission ZPLs at 977.6 nm, 978.1 nm and 978.9 nm). For revealing the structure of the observed Yb³⁺ centers, additional studies are required with use of electron paramagnetic resonance (EPR) and/or other kinds of measurements.

The luminescence decay from the ²F_{5/2} state of Yb³⁺ ions in MgMoO₄ was further studied at RT, as shown in Fig. 12. It can be well described by the single-exponential law yielding a luminescence lifetime τ_{lum} of 0.63 ms. The luminescence lifetimes of Yb³⁺ in Yb:MgWO₄ and Yb:Li:ZnWO₄ are rather close, ~0.36–0.37 ms and ~0.366 ms [4,8]. This relatively long lifetime of Yb³⁺ upper laser level is expected to favor reaching low laser thresholds. The observation of a nearly single-exponential luminescence decay despite the evidence of multiple Yb³⁺ species in MgMoO₄ revealed by LT spectroscopic studies is most probably due to the close transition probabilities for these optical

Table 2

Metal-to-oxygen interatomic distances for the M²⁺O₆ octahedra in the MgMoO₄ and ZnWO₄ crystals.

| MgMoO ₄ (This work) | | | | ZnWO ₄ [38] | |
|--------------------------------|-------------|----------------------|-------------|------------------------|-------------|
| [Mg1O ₆] | Distance, Å | [Mg2O ₆] | Distance, Å | [ZnO ₆] | Distance, Å |
| Mg1-O1 | 2 × 2.0450 | Mg2-O1 | 2 × 2.1310 | Zn-O2 | 2 × 1.981 |
| Mg1-O3 | 2 × 2.1808 | Mg2-O2 | 2 × 2.0700 | Zn-O1 | 2 × 2.032 |
| Mg1-O4 | 2 × 2.0442 | Mg2-O3 | 2.1612 | Zn-O1 | 2 × 2.319 |
| | | Mg2-O5 | 1.9933 | | |
| Average | 2.0900 | Average | 2.09275 | Average | 2.111 |

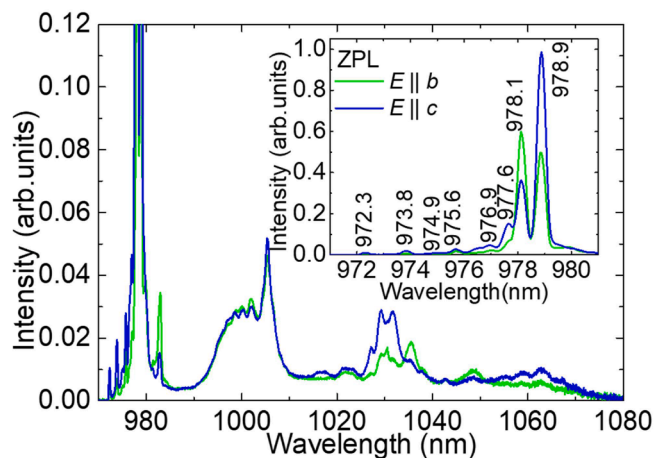


Fig. 10. LT (12 K) luminescence spectra of Yb³⁺ ions in the 0.2 at. % Yb:MgMoO₄ crystal for light polarizations $E \parallel b$ and $E \parallel c$. Inset – a close look at the ZPL structure, numbers – peak wavelengths in nm. Non-selective excitation.

centers.

4. Conclusions

In the present work, being motivated by the recent success of Yb³⁺-doped monoclinic magnesium monotungstate crystal, MgWO₄, in ultrashort-pulse lasers at 1 μm, we focused our attention on its monomolybdate counterpart, Yb³⁺:MgMoO₄, owing to its congruent melting allowing for the growth of large volume crystals by the Czochralski method. 0.2 at. % Yb:MgMoO₄ crystal with a high optical quality and only a weak amber-like coloration was grown and its structure was refined by the Rietveld method. This monomolybdate compound belongs to the monoclinic class (sp. gr. C2/m) and features a structure disorder due to two non-equivalent substitutional Mg²⁺ sites. The segregation coefficient of Yb³⁺ in MgMoO₄ is however relatively low, ~0.03, which is assigned to the difference in the ionic radii, charge and chemical nature of Mg²⁺ and Yb³⁺. We believe that it could be further improved by codoping the crystal with proper charge compensators (e.g., Li⁺), which would be essential for the laser development.

The absorption and emission properties of Yb³⁺ ions in MgMoO₄ were studied for polarized light. This crystal exhibits a notable spectral line broadening at room temperature leading to broadband and almost structureless spectral gain profiles above 1 μm. In this regard, it even outperforms the Yb³⁺:MgWO₄ crystal. This underlines the potential of the studied monomolybdate for 1-μm femtosecond lasers. To reveal the nature of this spectral line broadening, we employed site-selective spectroscopy at cryogenic temperatures (12 K). The multi-site behavior of rare-earth ions in MgMoO₄ was revealed for the first time. A variety of Yb³⁺ species comprising at least five non-equivalent optical centers was revealed. These centers can be regular magnesium positions Mg1 and Mg2 of the structure, as well as disturbed sites, corresponding to associates of ytterbium with different kinds of charge compensators (oxygen vacancies or accidental impurities, such as alkali ions). The intentional codoping, e.g., with Li⁺ ions, could reduce the number of optical centers or change their relative concentrations. This work extends the understanding of rare-earth doping mechanisms in monoclinic M²⁺XO₄ crystals (where M = Mg, Zn and X = W, Mo) and the obtained information would allow engineering the gain bandwidth of Yb³⁺ ions to support the generation of sub-100 fs pulses. The latter could involve the growth of “mixed” Yb:Mg(W,Mo)O₄ crystals.

CRedit authorship contribution statement

Ghassen Zin Elabedine: Writing – original draft, Investigation.

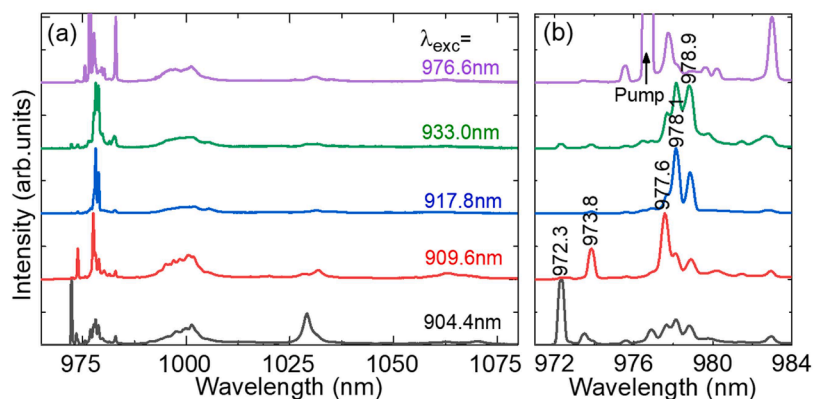


Fig. 11. LT (12 K) luminescence spectra of Yb^{3+} ions in the 0.2 at. % Yb:MgMoO_4 crystal measured under different excitation wavelengths λ_{exc} : (a) full spectra; (b) a close look at the ZPL transition.

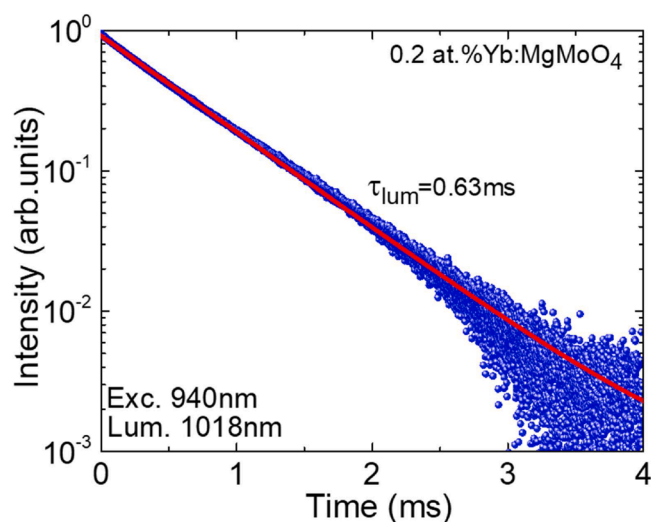


Fig. 12. Luminescence decay curve from the ${}^2F_{5/2}$ Yb^{3+} manifold in the 0.2 at. % Yb:MgMoO_4 crystal (finely powdered sample).

Yana S. Didenko: Writing – original draft, Investigation. **Pavel Loiko:** Writing – original draft, Methodology, Conceptualization. **Kirill A. Subbotin:** Writing – original draft, Investigation. **Anatolii I. Titov:** Investigation. **Liudmila D. Iskhakova:** Investigation. **Denis A. Lis:** Investigation. **Sergei K. Pavlov:** Investigation. **Yulia I. Zimina:** Investigation. **Kristina V. Kulshova:** Investigation. **Rosa Maria Solé:** Methodology. **Magdalena Aguiló:** Supervision, Funding acquisition. **Francesc Díaz:** Supervision, Funding acquisition. **Patrice Camy:** Supervision. **Xavier Mateos:** Writing – review & editing, Supervision, Funding acquisition.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Data availability

Data will be made available on request.

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